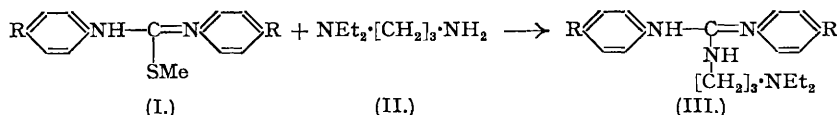


146. Some Derivatives of Diphenylguanidine.

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Certain diaryl-(γ -diethylaminopropyl)guanidines have been synthesised with a view to pharmacological tests.

SEVERAL lines of investigation suggested that diphenylguanidine derivatives containing a second basic centre in the molecule might have interesting pharmacological properties; three such compounds have therefore been synthesised by condensing an appropriate diphenylmethylisothiourea derivative (I) with γ -diethylaminopropylamine (II) to form a guanidine derivative (III) by loss of methylthiol. The necessary isothioureas were prepared by the method of Deck and Dains (*J. Amer. Chem. Soc.*, 1933, 55, 4986) from the corresponding thiourea and methyl sulphate. 4:4'-Dichloro- (I; R = Cl) and 4:4'-dimethoxy-diphenylisothioureas (I; R = OMe) have not previously been described. Condensation with (II) proceeded smoothly to give excellent yields of the guanidine (III; R = H, Cl or OMe), but only their *oxalates* could be obtained crystalline, except that diphenyl-(γ -diethylaminopropyl)guanidine (III; R = H) gave also a crystalline *picrolonate*. For pharmacological test the *hydrochlorides* were prepared; they were intractable gums which, however, gave satisfactory analyses.



EXPERIMENTAL.

(All compounds were dried in a high vacuum at 60° for analysis.)

N,N'-Diphenyl-N-(γ -diethylaminopropyl)guanidine.—Diphenylmethylisothiourea (9.68 g.) (Deck and Dains, *loc. cit.*) and γ -diethylaminopropylamine (5.20 g.) were dissolved in xylene (30 c.c.) and boiled for 15 hours under reflux. The guanidine *oxalate*, precipitated in theoretical yield by addition of excess oxalic acid in ethyl acetate, crystallised from alcohol in colourless felted needles, m. p. 99° (Found: C, 55.1; H, 6.5; N, 10.9. $\text{C}_{20}\text{H}_{28}\text{N}_4 \cdot 2\text{C}_2\text{H}_2\text{O}_4 \cdot \text{H}_2\text{O}$ requires C, 55.2; H, 6.5; N, 10.7%). The *picrolonate* crystallised from acetone after a month and was recrystallised with seeding from alcohol, forming orange rhombs, m. p. 111° with gas-evolution (Found: C, 55.2; H, 5.4; N, 19.1. $\text{C}_{20}\text{H}_{28}\text{N}_4 \cdot 2\text{C}_{10}\text{H}_8\text{O}_5 \cdot \text{N}_4 \cdot \text{H}_2\text{O}$ requires C, 55.2; H, 5.3; N, 19.3%). Other derivatives could not be obtained crystalline.

Dianisylmethylisothiourea.—Prepared in 88% yield by the method of Deck and Dains (*loc. cit.*), the *isothiourea* crystallised from benzene—light petroleum in large polyhedra, m. p. 84—85° (Found: C, 63.1; H, 5.8; S, 10.1. $\text{C}_{18}\text{H}_{18}\text{O}_2\text{N}_2\text{S}$ requires C, 63.5; H, 6.0; S, 10.6%).

Bis-(*p*-chlorophenyl)methylisothiourea.—Prepared similarly in 89% yield, the *isothiourea* crystallised from ether—light petroleum in colourless needles, m. p. 133° (Found: C, 53.8; H, 3.9. $\text{C}_{14}\text{H}_{12}\text{N}_2\text{Cl}_2\text{S}$ requires C, 54.0; H, 3.9%).

N,N'-Dianisyl-N-(γ -diethylaminopropyl)guanidine.—Prepared in the same way as the diphenylguanidine and in almost theoretical yield, this guanidine gave an *oxalate* which crystallised only with extreme difficulty and showed dimorphism; rounded rhombs, m. p. 100—105°, separated first from alcohol, and these gradually changed into pale yellow prisms, m. p. 150° with gas-evolution (Found: C, 50.6; H, 7.0; N, 9.3. $\text{C}_{22}\text{H}_{32}\text{O}_2\text{N}_4 \cdot 2\text{C}_2\text{H}_2\text{O}_4 \cdot 3\text{H}_2\text{O}$ requires C, 50.5; H, 6.8; N, 9.1%). No other derivative could be obtained crystalline; the *hydrochloride* was an extremely hygroscopic gum (Found: C, 56.9; H, 8.0; N, 12.2. $\text{C}_{22}\text{H}_{32}\text{O}_2\text{N}_4 \cdot 2\text{HCl} \cdot \frac{1}{2}\text{H}_2\text{O}$ requires C, 56.7; H, 7.5; N, 12.0%).

N,N'-Bis-(*p*-chlorophenyl)-N-(γ -diethylaminopropyl)guanidine.—Prepared similarly, the guanidine *oxalate* crystallised with great reluctance from alcohol in feathery, cream-coloured needles, m. p. 166—168° with gas-evolution (Found: C, 49.4; H, 5.7; N, 9.4. $\text{C}_{20}\text{H}_{28}\text{N}_4\text{Cl}_2 \cdot 2\text{C}_2\text{H}_2\text{O}_4 \cdot \frac{1}{2}\text{H}_2\text{O}$ requires C, 49.5; H, 5.3; N, 9.6%). No other derivative could be obtained crystalline; the *hydrochloride* was an exceptionally hygroscopic gum (Found: C, 50.9; H, 6.6; N, 11.7. $\text{C}_{20}\text{H}_{28}\text{N}_4\text{Cl}_2 \cdot 2\text{HCl} \cdot \frac{1}{2}\text{H}_2\text{O}$ requires C, 51.0; H, 6.1; N, 11.9%).

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