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Resolution of Neutral DL-Amino-acids via their (-)-Menthyl Ester Derivatives

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In connection with another study, we have recently prepared the (—)-methyl esters of a number of neutral DL-amino-acids, by azeotropic distillation using toluene-p-sulphonic acid (p-Ts) as a catalyst.¹

A routine gas-liquid-chromatographic (g.l.c.) analysis of the crystalline reaction product derived from DL-norvaline however showed it to be a D-norvaline derivative. We have since shown that many

¹ K. Harada and T. Hayakawa, Bull. Chem. Soc. Japan, 1964, 37, 191.

neutral DL-amino-acids can be readily resolved via their (—)-menthyl esters and that sterically pure p-Ts salts of the D-amino-acid (—)-menthyl ester are obtained in good yields (60—95% of the available D-isomer).

product was then diluted with anhydrous ether (20 ml.), filtered, washed with ether, and dried. The p-Ts salts (Table) as isolated and generally without further recrystallization are both chemically and sterically pure (g.l.c. analysis).

TABLE: Physical constants of D-amino-acid (-)-menthyl esters toluene-p-sulphonates

Amino-acid			m.p. (°c)	rotation [α]	G.l.c. Analy	G.l.c. Analyses of N -trifluoracetyl derivative* retention times of diastereoisomers		
				c=1%	steric purity of	(min.)		
				(methanol)	sample (%)	L, ()	D, (-)	
alanine			155-156	-46.5	100	$7 \cdot 2$	$8\cdot 2$	
α-aminobutyric	acid		160 - 162	-47.2	99	8.1	9.0	
leucine			172 - 174	$-44 \cdot 2$	99**	10.0	11.6	
norleucine			153154	$-43 \cdot 1$	99	12.0	13.8	
valine			209-210	$-54 \cdot 4$	100	5.8	$6 \cdot 7$	
norvaline .			159160	-49.9	100	9.0	10.4	
methionine .			149 - 152	-30.7	100	10.9	11.8	

^{*} In a typical assays the sample (50 mg.) was dissolved in ethyl acetate (1 ml.) and triethylamine (0·15 ml.) and methyl trifluoracetate (0·3 ml.) added. After 2 hours at 25° c, the solution was washed (H₂O), dried (Na₂SO₄) and injected into the gas chromatograph. G.l.c. analyses were carried out on a Wilkens 600C aerograph using a 5-foot \times $\frac{1}{4}$ -inch column (5% FFAP on Chromosorb W). During the analyses the separation temperature was 161° c (methionine 200° c) and the nitrogen flow 28 ml./min.

In a typical preparation the DL-amino-acid (0·025 moles), menthol (6 g.), p-Ts (6 g.), benzene (35 ml.) and toluene (15 ml.) were refluxed (Dean and Stark) till a clear solution resulted (30—60 hr.). The reaction mixture was then cooled and filtered to remove traces of unreacted amino-acid. After evaporation of the solvents, the residual oil was set aside for several hours. The partially crystalline

Finally, the p-Ts salts of the p-amino-acid (—)-menthyl esters (1 mM) were hydrolysed with methanolic caustic soda (3 mM) at 25°. The p-amino-acids were isolated by ion exchange chromatography (Dowex 50 W) and shown to be sterically pure by g.l.c. analysis.²

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^{**} Recrystallized from benzene-ether.

² B. Halpern and J. W. Westley, Chem. Comm., 1965, 247.