The Stereochemistry of Tricarbonylcycloheptatrienechromium Derivatives

By P. E. BAIKIE, O. S. MILLS, P. L. PAUSON, G. H. SMITH, and J. VALENTINE

(Department of Pure and Applied Chemistry, University of Strathclyde, Glasgow, C.1, and Department of Chemistry, The University, Manchester)

THE formation of a series of substituted tricarbonylcycloheptatrienechromiums by adding anions to tricarbonylcycloheptatrienechromium salts has been described.¹ This reaction invariably gives a single isomer, although two structures (I and II) are theoretically possible. These are designated exo- (I) and endo- (II) respectively and the X-ray data described below identify the above-mentioned products as the exo-isomers (I). A second and equally stereoselective method for their preparation is the nucleophilic displacement of the methoxy-group from tricarbonyl-exo-methoxycycloheptatrienechromium (I; X = OMe) by other substituents, X. By contrast, the endo-isomers (II) are the preferred products when substituted cycloheptatrienes (III) react with hexacarbonylchromium or with tricarbonyltripyridinechromium in the presence of boron trifluoride. Only in two cases $(X = CN \text{ and } X = C_6H_5)$ has the last mentioned method given both isomers (I and II). In other cases (X = Me, CH_2CO_2Et , $CH(CO_2Et)_2$, $CH_2C \equiv CH$) the endo-isomer (II) is formed with a high degree of stereoselectivity. We believe this must be attributed to the preferred "equatorial" conformation of the parent cycloheptatriene bearing a bulky substituent (III).

The preferred attack of nucleophiles on the tricarbonyltropyliumchromium cation from the side remote from the metal to give the *exo*-isomers (II) may likewise be the result of largely steric factors. It parallels the corresponding reactions of

the biscyclopentadienylcobalt(III) ion. As in the latter case,² the *exo*-configuration of the product has been established by X-ray-crystallographic examination of phenyl derivative (I; $X = C_6H_5$). The crystals are monoclinic, space group $P2_{1/2}$,



with unit cell dimensions a = 10.21 Å ± 0.02 Å, b = 14.01 Å ± 0.03 Å, c = 9.26 Å ± 0.02 Å and $\beta = 105.33^{\circ} \pm 0.15^{\circ}$. The unit cell, U = 1378 Å³, which contains four molecules, yields $d_c = 1.465$ g./c.c. ($d_m = 1.47$ g./c.c.). The structure was analysed from 1030 unique reflexions estimated from precession photographs taken with Mo-Ka radiation. The agreement index, R, is 8.6%(excluding unobserved reflexions) and all hydrogen atoms have been located. The structure closely resembles that of tricarbonylcycloheptatrienemolybdenum.³

(Received, August 18th, 1965; Com. 520.)

¹ J. D. Munro and P. L. Pauson, J. Chem. Soc., 1961, 3475; 3479.

² M. R. Churchill and R. Mason, Proc. Chem. Soc., 1963, 112.

³ J. D. Dunitz and P. Pauling, Helv. Chim. Acta, 1960, 43, 2188.