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2,6-Diamino-2,6-dideoxy-D-mannose Dihydrochloride

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WE have been concerned with the synthesis of diamino-sugars in connection with structural studies on synthetic aminated polysaccharides,1 and the synthesis of 3,6-diamino-3,6-dideoxy-Daltrose has been briefly reported.2 The appearance of a paper³ describing the synthesis of 2,6-diamino-2,6-dideoxy-D-mannose by a reaction sequence involving configurational inversion from a D-altroprecursor, prompts us to report our synthesis of this diamino-sugar by a route which sets out from a D-manno-precursor and involves no inversions at the secondary alcohol positions. A phenyl glycoside of 2-acetamido-2-deoxy-D-mannose was prepared by the fusion method,4 and amination at C-6 was achieved by selective toluene-p-sulphonation, azide replacement, and reduction.

2-Acetamido-1,3,4,6-tetra-O-acetyl-2-deoxy- β -Dmannopyranose⁵ (I) was fused with phenol, containing toluene-p-sulphonic acid or a zinc chlorideacetic acid-acetic anhydride mixture, for 1.3 hr. at 120-125°, to give phenyl 2-acetamido-3,4,6-tri-Oacetyl-2-deoxy-\alpha-D-mannopyranoside (II), which crystallized (68%) from propan-2-ol, and had m.p. 192—193°, $[\alpha]_{\rm D}^{30} + 73.6^{\circ}$ (c 1, chloroform). Treatment of a methanolic solution of (II) with a catalytic amount of sodium methoxide gave phenyl 2-acetamido-2-deoxy-α-D-mannopyranoside⁶ (III), which crystallized (85%) from ethanol-ether and

¹ M. L. Wolfrom, M. I. Taha, and D. Horton, J. Org. Chem., 1963, 28, 3553.

² M. L. Wolfrom, D. Horton, and Yen-Lung Hung, Abstracts Papers Amer. Chem. Soc., 1964, 148, 3D.

³ W. Meyer zu Reckendorf, Ber., 1965, 98, 93.

⁴ E. M. Montgomery, N. K. Richtmyer, and C. S. Hudson, J. Amer. Chem. Soc., 1942, 64, 690; S. Fujise and K. Yokoyama, J. Chem. Soc. Japan, 1951, 72, 728 (Chem. Abs., 1952, 46, 11116).

⁵ A. N. O'Neill, Canad. J. Chem., 1959, 37, 1747.

⁶ All new crystalline compounds gave satisfactory elemental analyses, were homogeneous by thin-layer chromatography on Silica Gel G (E. Merck, Darmstadt, Germany), and gave infrared spectra in agreement with the assigned structures.

had m.p. $94-98^{\circ}$, $[\alpha]_{D}^{18}+50\cdot0^{\circ}$ (c 1, ethanol). A solution of (III) in pyridine was treated with toluene-p-sulphonyl chloride (1·2 equiv.), to give phenyl 2-acetamido-2-deoxy-6-O-toluene-p-sulphonyl- α -D-mannopyranoside (IV; R=H) as a chromatographically homogeneous syrup (89%), characterized by acetylation with acetic anhydride-pyridine to the 3.4-diacetate (IV; R=Ac)

dideoxy- α -D-mannopyranoside (V; R=H). Acetylation of (V; R=H) with acetic anhydride-pyridine gave phenyl 2-acetamido-3,4-di-O-acetyl-6-azido-2,6-dideoxy- α -D-mannopyranoside (V; R=Ac) (57%) (from ether-petroleum), m.p. 144—145°, [α] $_{\rm D}^{\rm B}$ + 22·1° (c 1, ethanol). Reduction of either 6-azido-derivative (V; R=H or Ac) with hydrogen and palladium-charcoal gave products which no

$$(I) \qquad (II) \qquad (III) \qquad (III)$$

$$CH_2 \cdot OAc \qquad CH_2 \cdot OH \qquad OOH \qquad OOH$$

[77%; based on (III)] (from propan-2-ol), m.p. 154°, $[\alpha]_D^{18} + 102^\circ$ (c 1, chloroform). The syrupy 6-toluene-p-sulphonyl derivative (IV; R=H) was heated with excess of sodium azide in aqueous acetone for 28 hr. at 100°, to give a syrupy product (yield quantitative), having azide absorption but no sulphonate absorption in the infrared spectrum, formulated as phenyl 2-acetamido-6-azido-2,6-

longer showed azide absorption at $4\cdot7-4\cdot8~\mu$. Each product was hydrolyzed with 6n-hydrochloric acid for 1 hr. at 100° , to give crystalline 2,6-diamino-2,6-dideoxy- β -D-mannose dihydrochloride (VI), m.p. 157° (decomp.), $[\alpha]_{\rm D}^{21}$ $-10\cdot5 \rightarrow -1\cdot0 \pm 0\cdot5$ (c 1, water).

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