

Folioxanthin

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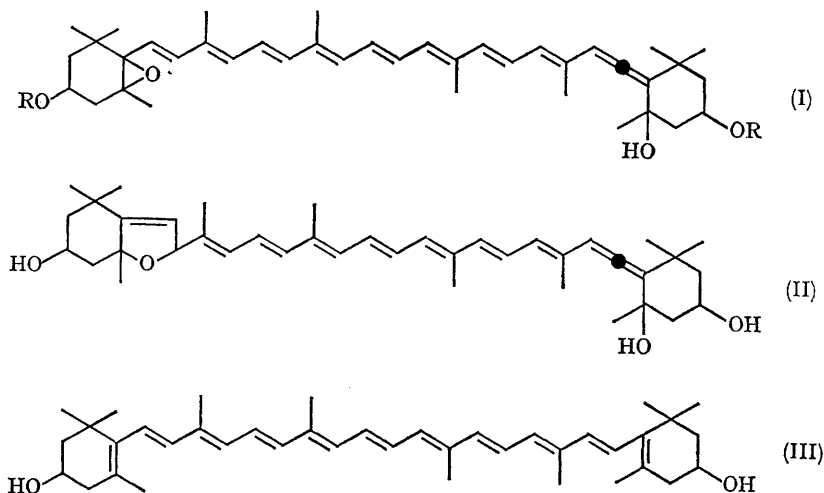
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FOLIAXANTHIN was first isolated from paprika,¹ and has since been found in lucerne, maple, and other leaves.² It has now been shown to have the structure (I); an allenic carotenoid has not previously been reported in higher plants.

Folioxanthin is an unstable amorphous solid, m.p. 128°, which exhibits the typical properties of a carotenoid epoxide.¹ Its light-absorption properties* reveal the presence of a conjugated nonaene chromophore [λ_{\max} 467, 439, and 416 m μ , (10^{-3} ϵ , 120·6, 134·6, and 94 respectively); λ_{\max} (C₆H₆) 478,

448, and 423 m μ], hydroxyl groups (ν_{\max} 3597 cm.⁻¹), and an allene group (ν_{\max} 1923 cm.⁻¹). Acetylation (CH₃COCl/C₅H₅N/C₆H₆) gives a crystalline derivative, m.p. 112° [λ_{\max} (C₆H₆) 478, 448, and 423 m μ ; ν_{\max} 3590, 1923, and 1720 cm.⁻¹] in almost quantitative yield. Mass-spectral studies show folioxanthin to be a trihydroxy-compound, C₄₀H₅₆O₄, and characterise its derivative as a (tertiary) hydroxy-diacetate, C₄₄H₆₀O₆. Dehydration (POCl₃/C₅H₅N) of this derivative gives an epiphasic pigment devoid of hydroxyl groups

* Unless indicated to the contrary, visible and infrared absorption spectra were determined on solutions in 96% ethanol and in chloroform respectively.



(partition tests between petrol and 95% MeOH), but with the same visible light absorption [$\lambda_{\max}(\text{C}_6\text{H}_6)$ 478, 448, and 423 $m\mu$].

On treatment with *ca.* 0.01% ethereal or chloroformic hydrogen chloride foliaxanthin undergoes an epoxide-furanoid oxide rearrangement giving (crystalline) foliachrome, m.p. 148° [λ_{\max} 451, 424, and 401 $m\mu$, (10^{-3} ϵ , 131.5, 136.2, and 87 respectively); $\lambda_{\max}(\text{C}_6\text{H}_6)$ 459, 430, and 406 $m\mu$; ν_{\max} 3595 and 1923 cm^{-1}]; mass spectral studies establish the formula $\text{C}_{40}\text{H}_{56}\text{O}_4$, and indicate the presence of at least 2 hydroxyl groups. Infrared irradiation of an ethereal solution of foliachrome and

lithium aluminium hydride gives (crystalline) zeaxanthin (III) (similar reductions of the epoxides and furanoid oxides of β -carotene and zeaxanthin regenerate the parent carotenoids³).

The results summarised above lead to the structure (I, R = H) for foliaxanthin, (I, R = OAc) for its diacetate, and (II) for foliachrome. These conclusions are consistent with the n.m.r. spectrum of foliaxanthin, and its diacetate.

It has been suggested⁴⁻⁶ that foliaxanthin is identical with neoxanthin, a common pigment of leaves.^{5,7,8}

(Received, May 13th, 1966; Com. 323.)

¹ L. Cholnoky, K. Györgyfy, E. Nagy, and M. Páncél, *Acta Chim. Acad. Sci. Hung.*, 1955, **6**, 143; 1958, **16**, 227; *Nature*, 1956, **178**, 410.

² L. Cholnoky and K. Györgyfy, unpublished results.

³ L. Cholnoky, J. Szabolcs, and Gy. Tóth, unpublished results.

⁴ A. L. Curl and G. R. Bailey, *Food Res.*, 1957, **22**, 323.

⁵ T. H. Goldsmith and N. I. Krinsky, *Nature*, 1960, **188**, 491.

⁶ S. Liaaen-Jensen and A. Jensen, *Progr. Chem. Fats and other Lipids*, 1965, **8**, 129.

⁷ H. H. Strain, "Leaf Xanthophylls", Carnegie Inst., Washington, 1938.

⁸ A. L. Curl, *J. Food Sci.*, 1965, **30**, 426.