

The Stereochemistry of Chanoclavine-I and Isochanoclavine-I

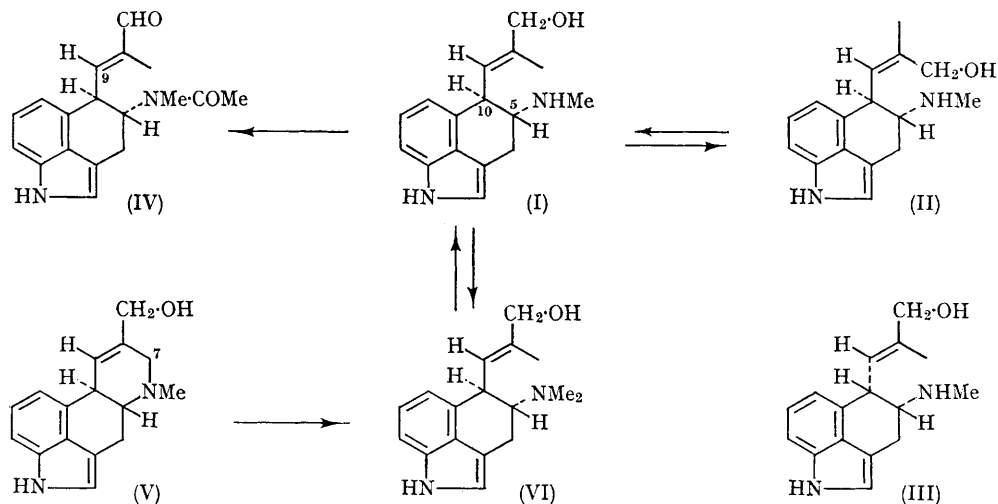
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THE recent discovery¹ of two further isomers of chanoclavine,² together with considerations of their possible role in the biosynthesis of ergot alkaloids, has focussed attention on the stereochemistry of these compounds. The absolute configuration of chanoclavine-I (I), at C-5 and C-10, has already been defined by a correlation³ with festuclavine.^{3,4}

Assignment of structures (II) and (III) for isochanoclavine-I and chanoclavine-II, respectively, as well as of the substitution pattern about the isolated double bond in (I) rests exclusively on comparison of n.m.r. data.¹ We now report experiments which establish the correctness of structures (I) and (II).

First, additional spectroscopic evidence was secured by oxidation of the *N*-acetyl derivative of (I)² with MnO_2 in acetone to form the aldehyde (IV), m.p. 209°, $[\alpha]_D -108^\circ$ ($CHCl_3$), having λ_{max} 227 $m\mu$ ($\log \epsilon$ 4.20 in ethanol after subtraction of the indole chromophore), in which the olefinic proton at C-9, τ 3.4 ($CHCl_3$), is clearly *cis* to the carbonyl group.⁵ Final proof for the substitution pattern about the double bond of (I) was obtained



as follows. Cleavage of the methiodide from elymoclavine (V) with sodium in liquid ammonia⁶ with or without added methanol gave, after chromatographic separation, a 70% yield of *N*-methyl-6,7-seco-elymoclavine (VI), m.p. 162°, $[\alpha]_D -127^\circ$ ($CHCl_3$), identical in all respects with the monomethylation product of (I). The *O*-acetyl-derivative of (VI) afforded on demethylation with diethylazodicarboxylate in ether⁷ the *O*-acetyl derivative of (I), m.p. 111°, $[\alpha]_D -160^\circ$ ($CHCl_3$),

converted by alkaline hydrolysis into a compound identical in all respects with natural chanoclavine-I. Since the geometry of the double bond is known to be preserved in the course of similar reductive cleavages,⁸ a *cis*-relationship of the CH_2OH group and the olefinic proton of (I) is established.

Irradiation of (I) in *t*-butyl alcohol with a low-pressure mercury lamp gave a mixture containing roughly equal amounts of (I) and (II), from which

pure isochanoclavine-I (II),¹ m.p. 190°, $[\alpha]_D -208^\circ$ (pyridine), could be isolated in 30% yield. A similar mixture was obtained on irradiation of (II). Thus, chanoclavine-I and isochanoclavine-I have the same configuration at C-5 and C-10 and differ only in the relative positions of CH_2OH and olefinic proton. The reversible isomerisation (I) \rightleftharpoons (II) is likely to occur by an intramolecular energy transfer mechanism (*cf.* ref. 9).

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