

A Crystalline Active Ester of Benzyloxycarbonylserine

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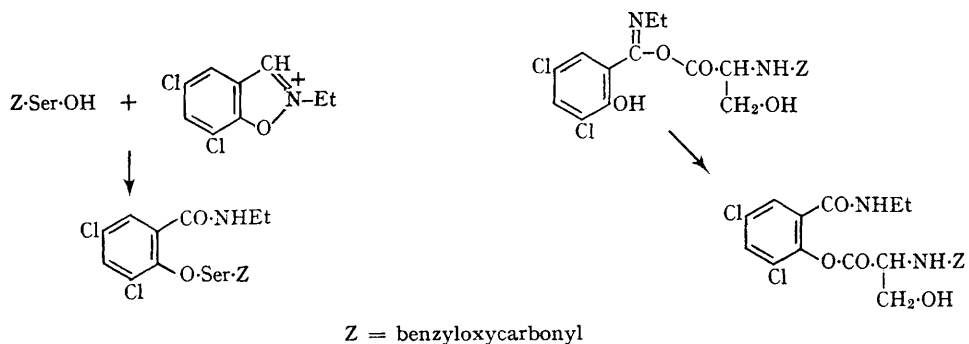
IN the course of an investigation of the usefulness of the benzisoxazolium cation described by Woodward and Kemp,¹ we felt that introduction of electron-withdrawing substituents in the benzene ring might prove beneficial by increasing the "activity" of the intermediate active esters. Accordingly, commercially available 3,5-dichlorosalicylaldehyde was converted into 5,7-dichloro-*N*-ethyl benzisoxazolium fluoroborate by the procedure described by Kemp and Woodward.

We now find that the reagent reacts with a variety of *N*-protected amino-acids to give crystalline active esters in good yield. Even benzyloxycarbonylserine gives a crystalline active

ester in 70—80% yield. Addition of the reagent as a solid to a cooled (ice-salt), stirred solution of benzyloxycarbonylserine and triethylamine in methylene chloride gave the active ester in a few minutes as a crystalline solid: m.p. 118—121° or 156—160° (dimorphous forms; identical i.r. spectra in solution) i.r. (Nujol): 3400, 3310, 3260 sh, 1780, 1690, and 1655 cm.⁻¹.

The product has been reacted with glycine *p*-nitrobenzyl ester to provide the protected dipeptide² in 78.5% yield.

The great difficulty in making active esters of benzyloxycarbonylserine has been commented upon recently by Bodanszky and Ondetti.³ The



success in the present instance evidently depends on the fact that the crucial reaction involves an intramolecular nucleophilic displacement by a phenol rather than an intermolecular one as in

those cases involving use of dicyclohexylcarbodiimide (DCCI).

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¹ D. S. Kemp and R. B. Woodward, *Tetrahedron*, 1965, **21**, 3019.

² B. Iselin and R. Schwyzer, *Helv. Chim. Acta.*, 1962, **45**, 1499.

³ M. Bodanszky and M. A. Ondetti, *Chem. and Ind.*, 1966, 26.