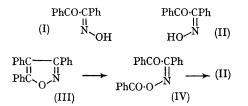
The Structure of the Benzil Monoximes

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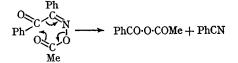
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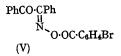
THE accepted structures for benzil α -monoxime (I), and benzil β -monoxime, (II), are based mainly on the fact that ozonization of triphenylisoxazole, (III), affords the benzoate of benzil β -monoxime, (IV).¹



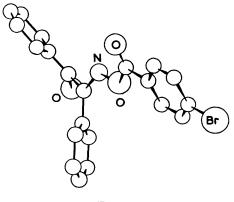
Other facts, however, could be interpreted by reversing the structures assigned to the two forms of benzil monoxime. The acetate of benzil β monoxime is thermally stable at 190°. The acetate of benzil a-monoxime is thermally unstable at 140°, and is converted into substances which on treatment with water yield acetic acid, benzoic acid, and benzonitrile.² Undoubtedly benzonitrile and acetic-benzoic anhydride are the initially-formed products. Since many pyrolytic decompositions occur by intramolecular cyclic mechanisms, one might use the above facts to argue that the acetate of benzil α -monoxime has the acetoxy- and benzoyl groups in the *cis*-configuration and hence the structures for (I) and (II) should be reversed.*



With the above argument in mind, a solution to this problem was sought by a method which would not require the speculation inherent in the interpretation of chemical evidence. Accordingly we have determined the structure of the *p*-bromobenzoate of benzil α -monoxime³ by X-ray crystallographic analysis. The results show clearly that the structure is (V), with the *p*-bromobenzoate group *trans* relative to the carbonyl group. Hence Meisenheimer's assignment of structure to the oximes (I) and (II) is correct.



Crystals of benzil α -monoxime *p*-bromobenzoate (V), C₂₁H₁₄NO₃Br, m.p. 105°, are triclinic, space group $\overline{P1}$ (C₃), with 2 molecules in a unit cell with



FIGURE

Benzil α -monoxime p-bromobenzoate viewed along the c crystal axis.

* This suggestion was originally made in an oral examination for the Ph.D. degree at Ohio State University in 1959 by the candidate, Tadamichi Fukunaga.

a = 8.78, b = 17.01, c = 6.27 Å, $\alpha = 95.1^{\circ}, \beta = 91.1^{\circ}, \gamma = 102.3^{\circ}.$ $D_{\rm m} = 1.47$ g.cm.⁻³, $D_{\rm c} = 1.49$ $g.cm.^{-3}$. The crystals were twinned, but intensity data were estimated from Weissenberg photographs obtained from a specimen with one individual of the twin sufficiently larger than the other to permit accurate indexing.

About 1300 independent structure factors were evaluated, and the analysis proceeded by threedimensional Patterson and minimum functions.

- ¹ J. Meisenheimer, Ber., 1921, 54, 3206.
 ² A. H. Blatt and R. P. Barnes, J. Amer. Chem. Soc., 1934, 56, 1148.
 ³ R. P. Barnes, J. Amer. Chem. Soc., 1938, 60, 1082.

The electron-density distribution revealed the positions of the 25 light atoms in the molecule. Refinement proceeded by Fourier and least squares methods, the final R-value being 16.0% over all the observed structure factors.

The configuration of the molecule obtained from the X-ray analysis is best illustrated in the view obtained in projection along the short c axis, which is shown in the Figure.

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