

# The Chemical Conversion of Trichokaurin into (–)-Kaurene, Atisine, Garryine, and Veatchine

By EIICHI FUJITA,\* TETSURO FUJITA, and MASAYUKI SHIBUYA

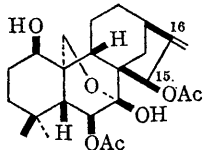
(Institute for Chemical Research, Kyoto University, Takatsuki, Osaka-fu, Japan)

THIS Communication deals with the chemical conversion of trichokaurin (I)<sup>1</sup> into the keto-carboxylic acid (VIII)<sup>2,3</sup>

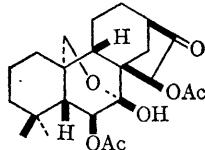
The keto-diacetate (II) [ $C_{23}H_{32}O_7$ , m.p. 186–187° (decomp.),  $\nu_{max}$  (KBr) 3450, 1750, and 1735  $cm^{-1}$ ,  $\delta$  5.45 (1H, doublet,  $J = 1.0$  c./sec., C-15–H) and 5.20 p.p.m. (CDCl<sub>3</sub>) (1H, doublet,  $J = 7.5$  c./sec., C-6–H)] which was derived from trichokaurin (I) through the route described in the preceding Communication,<sup>1</sup> was allowed to react with calcium in liquid ammonia<sup>4</sup> to afford the alcohol (III) [ $\nu_{max}$  (CHCl<sub>3</sub>) 3400 and 1735  $cm^{-1}$ ,  $\delta$  3.77 p.p.m. (CDCl<sub>3</sub>) (1H, doublet,  $J = 5.0$  c./sec., C-6–H)] and the triol (V) [ $\nu_{max}$  (KBr) 3350  $cm^{-1}$ ]. Their structures were confirmed by acetylation to (IV) [ $C_{21}H_{30}O_5$ , m.p. 210–215° (decomp.),  $\nu_{max}$  (KBr) 3350, 1740, and 1720  $cm^{-1}$ ,  $\delta$  2.06 (3H, singlet) and 5.06 p.p.m. (CDCl<sub>3</sub>) (1H,

doublet,  $J = 5.0$  c./sec., C-6–H)] and (VI) [ $C_{23}H_{34}O_6$ , m.p. 194–195° (decomp.),  $\nu_{max}$  (KBr) 3450, 1735, and 1710  $cm^{-1}$ ,  $\delta$  2.07, 2.02 (each 3H, singlet), 4.80 (1H, quartet,  $J = 4.0$  and 7.0 c./sec., C-16–H), and 5.03 p.p.m. (CDCl<sub>3</sub>) (1H, doublet,  $J = 5.0$  c./sec., C-6–H)] respectively.

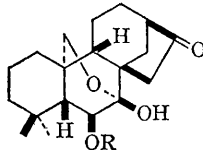
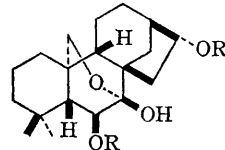
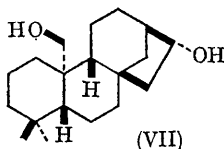
Subsequently, Nagata's modification<sup>5</sup> of the Wolff–Kishner reduction was applied to the triol (V). The product, without purification, was hydrogenated in the presence of Adams' catalyst to yield diol (VII) [ $\delta$  1.88 (6H, singlet) and 4.09 p.p.m. (CDCl<sub>3</sub>) (2H, singlet, C–CH<sub>2</sub>OH)]. The latter, on oxidation with Jones' reagent at room temperature, gave a keto-carboxylic acid, m.p. 238–240°, which proved to be identical with the authentic sample of the compound (VIII),<sup>3</sup> m.p. 240–242°, † kindly supplied by Dr. S. Masamune (University of Alberta), by mixture melting-point



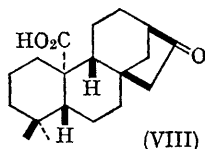
(I)



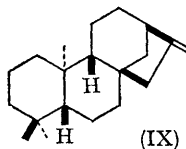
(II)

(III) R=H  
(IV) R=Ac(V) R=H  
(VI) R=Ac

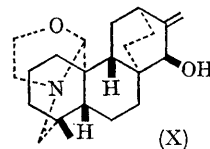
(VII)



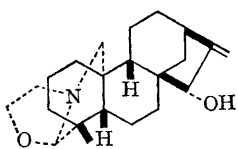
(VIII)



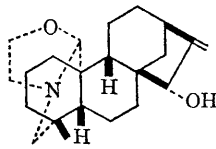
(IX)



(X)



(XI)



(XII)

determination and i.r. and mass-spectral comparison.

Since the compound (VIII) has been converted into (–)-kaurene (IX),<sup>3</sup> atisine (X),<sup>6</sup> garryine (XI),<sup>7</sup> and veatchine (XII),<sup>8</sup> the present transformation means the chemical conversion of trichokaurin into (–)-kaurene and those diterpene alkaloids.

(Received, March 20th, 1967; Com. 272.)

† cf. m.p. 251–252.5° in ref. 3.

<sup>1</sup> E. Fujita, T. Fujita, and M. Shibuya, preceding Communication.

<sup>2</sup> S. Masamune, *J. Amer. Chem. Soc.*, 1964, **86**, 288.

<sup>3</sup> S. Masamune, *J. Amer. Chem. Soc.*, 1964, **86**, 289.

<sup>4</sup> H. Vorbrueggen and C. Djerassi, *J. Amer. Chem. Soc.*, 1962, **84**, 2990.

<sup>5</sup> W. Nagata and H. Itazaki, *Chem. and Ind.*, 1964, 1194.

<sup>6</sup> S. Masamune, *J. Amer. Chem. Soc.*, 1964, **86**, 291.

<sup>7</sup> S. Masamune, *J. Amer. Chem. Soc.*, 1964, **86**, 290.

<sup>8</sup> S. W. Pelletier and K. Kawazu, *Chem. and Ind.*, 1963, 1879.