

Stereochemical Studies concerning the Biosynthesis of Narcotine, Protopine, and Chelidonine

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OUR earlier work with multiply-labelled precursors proved¹ that the opium alkaloid narcotine (IV) is biosynthesised from reticuline (as I) and, importantly, that the carbonyl residue is derived without loss of carbon from the *N*-methyl group of (I). This supported the view² that phthalideisoquinolines are formed from tetrahydroprotoberberines and scoulerine (as II) was suggested¹ as the precursor. The synthesis³ of (–)-[6-¹⁴C,14-³H]-scoulerine (II) and its enantiomer allowed critical tests in *Papaver somniferum* plants. Experiment 1 (Table) shows that the (–)-isomer, which corresponds³ to narcotine⁴ in absolute configuration, is incorporated well with some tritium loss whereas the (+)-form is virtually ineffective (Expt. 2). Degradation of the narcotine⁵ proved 97% of its ¹⁴C activity to be located at C-3 (see IV) so demonstrating the protoberberine → phthalideisoquinoline conversion. Scoulerine has recently been isolated from opium.⁶ The oxidative process whereby C-13 of (II)

becomes the lactonic C-9 in narcotine (IV) was studied with (–)-, and (+)-[3-¹⁴C,9-³H₂]-reticuline. As expected, the (+)-form (I) was the better precursor (Expts. 3 and 4) and loss of ca. 50% of the tritium from C-9 of (I) during its biological conversion into narcotine is in accord with a stereospecific oxidation at C-13 of (–)-scoulerine (II) or a close relative.

The biosynthesis of protopine (V) in *Dicentra* plants has been shown⁷ to involve (+)-reticuline (I). That (–)-scoulerine (II) is a further intermediate in *Chelidonium majus* is established by Expts. 1, 2, 7, and 8. The ³H values recorded should be compared with those for stylophine and are in agreement with the conversion (II) → (III) → (V). The exact mechanism of the second stage requires further study. Again only (–)-scoulerine acted as an effective precursor of stylophine (III) and protopine (V). Conversion of the protopine *via* (VI) into (VII) followed by Kuhn–Roth oxidation gave acetic acid which was

TABLE

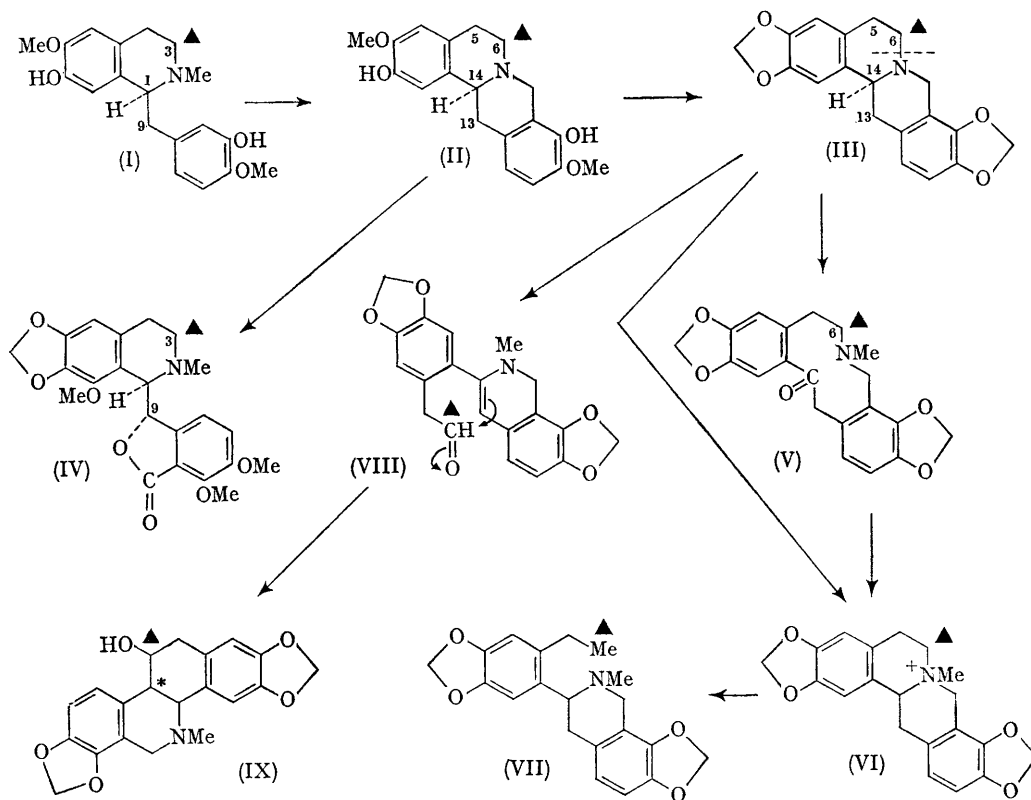
Expt. No.	Precursor	(\pm)-Stylophine ^d (as III)		Narcotine (IV)		Protopine (V)		Chelidonium (IX)	
		Incorp. ^a	% ^b Loss or gain ³ H	Incorp. ^a	% ^b Loss or gain ³ H	Incorp. ^a	% ^b Loss or gain ³ H	Incorp. ^a	% ^b Loss or gain ³ H
1	(-)-[6- ¹⁴ C, 14- ³ H]-(II)	0.28	+10	2.3	-13	0.92	-100	0.61	-100
2	(+)-[6- ¹⁴ C, 14- ³ H]-(II)	0.013	c	0.02	c	0.04	c	0.014	c
3	(+)-[3- ¹⁴ C, 9- ³ H ₂]-(I)	0.53	-4	0.09	-46	—	—	0.58	-18
4	(-)-[3- ¹⁴ C, 9- ³ H ₂]-(I)	0.02	-2	0.03	-55	—	—	0.03	-16
5	(\pm)-[6- ¹⁴ C, 5- ³ H ₂]-(II)	0.3	+10	—	—	—	—	1.2	+23
6	(\pm)-[6- ¹⁴ C, 6- ³ H ₂]-(II)	0.2	c	—	—	—	—	1.2	-38
7	(-)-[6- ¹⁴ C, 6- ³ H ₂]-(II)	0.18	+9	—	—	0.72	+9	0.81	-37
8	(+)-[6- ¹⁴ C, 6- ³ H ₂]-(II)	0.009	c	—	—	0.01	c	0.01	c

^a Incorporations are based upon ¹⁴C; comparable feeding conditions were used for each enantiomeric pair.

^b Calculated relative to the ¹⁴C-label; the figures show the % change in ³H: ¹⁴C ratio from that in the precursor.

^c Not examined.

^d Stylophine is present as a partial racemate containing an excess of the (-)-form. The specific activities of the (-)- and (\pm)-forms proved that little labelling of the (+)-form occurred over the feeding period.



▲ indicates ¹⁴C label

degraded by Schmidt's method to methylamine which contained all the original activity. Protopine (V) was thus proved to be labelled specifically at C-6. Stylophine (III) was similarly degraded with the same result.

Attention is drawn to the small rise in ³H: ¹⁴C ratio (*ca.* 10%) for those biological conversions not involving the ³H-labelled site *e.g.*, stylophine in Expts. 1 and 7 and protopine in Expt. 7. This effect has been observed in other cases and will be

discussed in our full Paper; it is of importance in the sequel for chelidonine.

The late stages leading to chelidonine (IX) in *C. majus* were shown in this laboratory⁸ to be (I) → (II) → (III) → (IX) and the intervention of stylopine (III) has been confirmed.⁹ These results are in agreement with earlier suggestions that chelidonine is derived in some way from the tetrahydroprotoberberine skeleton.¹⁰ Support for a mechanism *via* (VIII) has now been obtained by examining the fate of the hydrogen atoms attached to C-5, -6, -13, and -14 of (-)-stylopine (III) during its biological conversion into chelidonine (IX). (-)-Scoulerine, and in two experiments reticuline (I), labelled with ³H at the positions corresponding to C-5, -6, -13, and -14 of stylopine (III) were used for these studies. For each enantiomeric pair, the substance corresponding in absolute configuration to (-)-stylopine (III) was by far the more effective precursor of chelidonine. Expt. 1 shows that the hydrogen atom at C-14 is lost completely. Expt. 5 indicates that C-5 is not involved in the stylopine → chelidonine transformation but C-6 is clearly affected (Expts. 6 and 7). The losses of ³H recorded in the Table are calculated from the ³H/¹⁴C ratio in the original precursor and when account is taken of the general

rise in this ratio along the biosynthetic pathway, ³H-loss from C-6 becomes close to that expected (50%) for a stereospecific oxidation which does not involve an isotope effect. In contrast, Expt. 3 shows that the ³H:¹⁴C ratio in chelidonine is only 18% below that of the precursor proving a small ³H-loss from C-13. Even allowing for the general rise in ³H level, this loss is well below 50% and suggests for C-13 a nonstereospecific removal of hydrogen involving an isotope effect. Complete loss of ³H when chelidonine from Expt. 3 was converted into the methine by Hofmann degradation established specific ³H-labelling at the starred position in the alkaloid (IX) and the illustrated location of the ¹⁴C label was proved as earlier.⁸ Stereospecific generation of a C-14/N double-bond in (III) followed by isomerisation to the C-13/C-14 enamine would explain these results which contrast with those indicating a sterically controlled attack at C-13 (see II) in narcotine biosynthesis.

Scoulerine labelled at C-6 and C-13 with ³H in known absolute configuration is being used in current work to discover the stereochemistry of the oxidative processes.

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