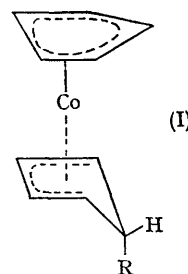


**The Stereochemistry and Conformation of  
Hexamethylcyclohexadienylrhenium Tricarbonyl:  
a Complex exhibiting an Anomalously Low Carbon-Hydrogen  
Stretching Frequency**

By PETER H. BIRD and MELVYN R. CHURCHILL\*

(Department of Chemistry, Harvard University, Cambridge, Massachusetts, 02138)

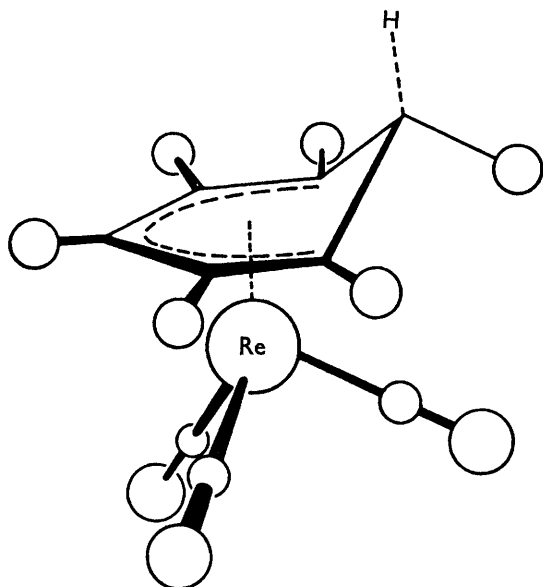
HYDRIDE ion attack on the  $(\pi\text{-C}_5\text{H}_5)_2\text{Co}^+$  ion results in the formation of the cyclopentadiene complex,  $\pi\text{-C}_5\text{H}_5\text{CoC}_5\text{H}_6$ <sup>1</sup> which exhibits a broad, intense C-H stretch at 2742  $\text{cm}^{-1}$ . Deuteration studies have shown that this anomalously low C-H stretch involves the newly-formed hydrogen atom.<sup>1</sup> Substituted complexes,  $\pi\text{-C}_5\text{H}_5\text{CoC}_5\text{H}_5\text{R}$ , show no unusual C-H stretches, and crystallographic studies on two such species (R = Ph,<sup>2</sup> R = PhCO<sup>3</sup>) have demonstrated conclusively that the cyclopentadiene ligand is nonplanar, that the substituent occupies the *exo*-position (see I), and that the *endo*-hydrogen atom is some 3.05 Å from the metal.<sup>2</sup> However, no studies of complexes which possess the anomalous C-H stretch have yet



been reported and there are *still* conflicting views as to whether it involves the *endo*- or *exo*-hydrogen.<sup>4</sup>

Attempts to study the crystal structure of  $\pi\text{-C}_6\text{H}_5\text{CoC}_6\text{H}_5$  have failed, due to the  $X$ -ray sensitivity of this complex.<sup>5</sup> However, the cyclohexadienyl complex,  $[\text{C}_6\text{Me}_6\text{H}]\text{Re}(\text{CO})_3$ , formed by hydride-ion attack on the hexamethylbenzenerehenium tricarbonyl cation,<sup>6</sup> shows a similar broad, intense C-H stretch at low frequency ( $2790\text{ cm}^{-1}$ ). In order to obtain positive evidence on the origin of this stretch, a single-crystal  $X$ -ray analysis of hexamethylcyclohexadienylrhenium tricarbonyl was undertaken.

The complex crystallizes in space-group  $Pbca$  (No. 61) with  $a = 13.512$ ,  $b = 14.449$ ,  $c = 15.552\text{ \AA}$ ,  $Z = 8$ . The structure has been solved



FIGURE

by conventional Patterson, Fourier, and least-squares refinement techniques. Complete three-dimensional data were collected on a Buerger Automated Diffractometer. The discrepancy index is  $R_F = 0.20$  for all 2083 reflections, and  $R_F = 0.12$  for the 1103 reflections with  $|F^2| > 3\sigma(F^2)$ . [The large number of extremely weak reflections is due to the proximity of the rhenium atom to the special position,  $y = 1/4$ .] The overall geometry of the molecule is shown in the Figure. The delocalized system formed by atoms C-2, C-3, C-4, C-5, C-6 is planar (r.m.s. deviation =  $0.025\text{ \AA}$ ), the mean Re-C distance being  $2.35\text{ \AA}$ . The rhenium atom is  $1.86\text{ \AA}$  below this plane whereas C-1, the saturated carbon atom of the cyclohexadienyl system, is  $0.61\text{ \AA}$  above the plane and the *endo*-methyl attached to C-1 is  $0.48\text{ \AA}$  above the plane. The dihedral angle between the planes formed by C-2, C-3, C-4, C-5, C-6 and C-2, C-1, C-6 is  $50^\circ$ .

This crystallographic analysis thus unambiguously demonstrates that the anomalous C-H stretch involves the *exo*-hydrogen atom. This result is consistent with previous crystallographic studies on cyclopentadiene complexes,<sup>2,3</sup> and is expected to be general for metal-cyclopentadiene and metal-cyclohexadienyl systems.

Although the saturated C-1 is  $2.82\text{ \AA}$  from the rhenium atom, it seems likely that the strange properties of the partially-hydridic<sup>1</sup> *exo*-hydrogen atom may result from some long-range interaction between the rhenium atom and carbon-1.

We thank Dr. G. Winkhaus for providing a sample of the complex. This research was supported by a grant from the Advanced Research Projects Agency.

(Received, June 26th, 1967; Com. 646.)

<sup>1</sup> M. L. H. Green, L. Pratt, and G. Wilkinson, *J. Chem. Soc.*, 1959, 3753.

<sup>2</sup> M. R. Churchill and R. Mason, *Proc. Roy. Soc.*, 1964, A, 279, 191.

<sup>3</sup> M. R. Churchill, *J. Organometallic Chem.*, 1965, 4, 253.

<sup>4</sup> F. A. Cotton and G. Wilkinson, "Advanced Inorganic Chemistry", 2nd Edition; Interscience, New York, 1966, p. 778.

<sup>5</sup> M. R. Churchill, Ph.D. Thesis (London), 1964, pp. 71-2.

<sup>6</sup> G. Winkhaus and H. Singer, *Z. Naturforsch.*, 1963, 18b, 418.