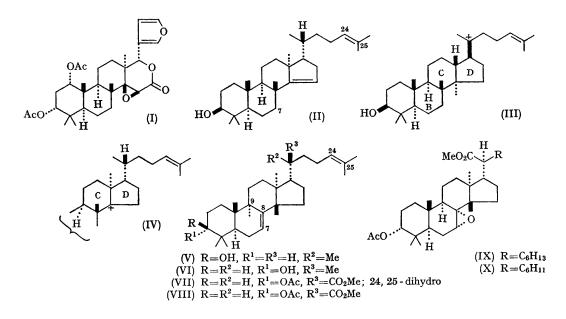
## A Chemical Model for a Possible Oxidative Rearrangement in the Biosynthesis of Tetranortriterpenes: the Preparation of Methyl 3α-Acetoxy-7-oxoapotirucalla-14,24-dien-21-oate

By G. P. COTTERRELL, T. G. HALSALL,\* and M. J. WRIGLESWORTH (The Dyson Perrins Laboratory, Oxford University, Oxford)

A CURRENT problem of terpenoid biosynthesis is that of the tetranortriterpenes such as khivorin (I),<sup>1</sup> gedunin,<sup>2</sup> and limonin<sup>3</sup> found in the Meliaceae and Rutaceae families. It was originally suggested<sup>3</sup> that limonin might arise from apoeuphol (II). This could be biosynthesised either directly from squalene via the ions (III) and (IV) or via butyrospermol (V) or its  $\Delta^{8}$ -isomer, euphol, and subsequent migration back of the C-14 methyl group to C-8. Recently it has been suggested? that a tirucalla-7,14-dien-3-ol [possibly the  $3\alpha$ -ol (VI)] might be the precursor of the tetranortriterpenes. All the known ones have oxygen at C-7 and when it is present as a hydroxyl group the configuration is  $\alpha$ . This suggests that a key step in the biosynthesis is the rearrangement of the  $7\alpha$ ,  $8\alpha$ -epoxide of a tirucallol (possibly degraded) derivative [cf. (IX)] to a 7 $\alpha$ -hydroxyapotirucallol derivative [cf. (XI)]. Evidence that an oxidative rearrangement of the

to a ketone formulated as a 7-oxo-apoeuphol derivative, although the proposed structure was not proven. A rearrangement of the C-14 methyl group has now been accomplished starting from a  $7\alpha,8\alpha$ -epoxide in the tirucallol series.

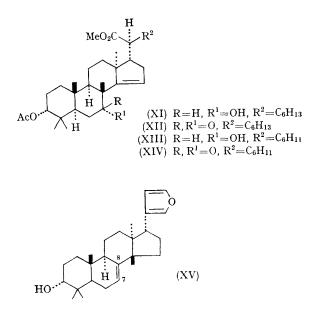
Methyl acetyl-dihydro- $\alpha$ -elemolate (VII) (methyl  $3\alpha$ -acetoxytirucall-7-en-21-oate) on ozonolysis followed by reductive workup with zinc dust readily gave the  $7\alpha$ ,  $8\alpha$ -epoxide m.p. 130–132°,  $[\alpha]_{\rm D}$ –78°, which was smoothly rearranged with BF<sub>3</sub>-Et<sub>2</sub>O in benzene to the  $7\alpha$ -hydroxyapo-derivative (XI). This has not yet been obtained crystalline but was homogeneous (t.l.c.) and had the expected n.m.r. spectrum. It was characterised by oxidation to the corresponding ketone (XII), m.p. 160°,  $[\alpha]_{\rm D}$ –140°, when the chemical shift of the C-15 proton changed significantly from  $\tau$  4·52 to 4·03 showing the close spatial relationship of the keto-group to the double bond.



C-14 methyl group can be brought about was first provided by Spring *et al.*,<sup>5, of, 6</sup> who oxidised dihydrobutyrospermyl acetate with chromic acid

Recently we have found that "elemolic acid" from elemi resin is a mixture of *both* the  $\Delta^{7}$ -isomer (VIII) and the  $\Delta^{8}$ -isomer.<sup>cf.10</sup> Separation of the

isomers is tedious but ozonolysis of the methyl esters from the mixture of acetylated 24,25-dihydroacids gave an easily separable mixture of the  $7\alpha$ ,  $8\alpha$ epoxide (IX) and ketonic material arising from the  $\Delta^{8}$ -isomer. Treatment of the acetylated methyl elemolate mixture ( $\Delta^{7}$ - and  $\Delta^{8}$ -isomers) gave the corresponding admixed 24,25-dibromides. Ozonolysis of these followed by reductive workup with zinc dust, which brought about concomitant debromination, afforded after chromatography, the 7a,8aepoxide (X), m.p. 127.5–129°,  $[\alpha]_{\rm D}$  –80°, which was rearranged with BF<sub>3</sub>-Et<sub>2</sub>O/benzene to the  $7\alpha$ -hydroxyapo-derivative (XIII) which was characterised as the corresponding 7-ketone (XIV), m.p. 146—148°,  $\lceil \alpha \rceil_{\rm D}$  –156°. A route is therefore now available to 7\alpha-hydroxyapotirucallol derivatives which should be of potential use both for the preparation of possible biogenetic precurors of the tetranortriterpenes and for the partial synthesis The present pattern of known of meliacins. structures suggests that the furan derivative (XV) and its  $7\alpha$ ,  $8\alpha$ -epoxide may be key compounds in the biosynthesis. The preparation of these compounds is being attempted.



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<sup>1</sup>C. W. L. Bevan, T. G. Halsall, M. N. Nwaji, and D. A. H. Taylor, J. Chem. Soc., 1962, 768; J. D. Connolly, K. L. Handa, R. McCrindle, and K. H. Overton, Chem. Comm., 1966, 837; E. K. Adesogan, J. W. Powell, and D. A. H.

Taylor, J. Chem. Soc. (C), 1967, 554. <sup>2</sup> A. Akisanya, C. W. L. Bevan, T. G. Halsall, J. W. Powell, and D. A. H. Taylor, J. Chem. Soc., 1961, 3705; G. A. Sutherland, G. A. Sim, and J. M. Robertson, Proc. Chem. Soc., 1962, 768.

<sup>8</sup>D. Arigoni, D. H. R. Barton, E. J. Corey, O. Jeger, L. Caglioti, Sukh Dev, P. G. Ferrini, E. R. Glazier, A. Melera,
<sup>8</sup>D. Arigoni, D. H. R. Barton, E. J. Corey, O. Jeger, L. Caglioti, Sukh Dev, P. G. Ferrini, E. R. Glazier, A. Melera,
<sup>8</sup>K. Pradhan, K. Schaffner, S. Sternhell, J. F. Templeton, and S. Tobinga, *Experientia*, 1960, 16, 41.
<sup>4</sup>C. W. L. Bevan, D. E. U. Ekong, T. G. Halsall, and P. Toft. J. Chem. Soc., (C), 1967, 820.
<sup>5</sup>W. Lawrie, W. Hamilton, F. S. Spring, and H. S. Watson, J. Chem. Soc., 1956, 3272.
<sup>6</sup>D. H. R. Barton, S. K. Pradhan, S. Sternhell, and J. F. Templeton, J. Chem. Soc., 1961, 255.
<sup>7</sup>A. J. Birch, D. J. Collins, Sultan Muhammad, and J. P. Turnbull, J. Chem. Soc., 1963, 2762.