## A New Heteroaromatic Ring System derived from 3,4-Dihydro-4-oxo-1,2,3-benzotriazine

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A RECENT paper¹ on the synthesis of benzotriazolo-[2,1-a]naphtho[1,8-de]triazine prompts us to report our observations on the reaction of 3,4-dihydro-2-(o-nitrophenyl)-4-oxo-1,2,3-benzotriazine (I)² under reflux with triethyl phosphite. The product of the reaction, a yellow crystalline solid,  $C_{13}H_8N_4O$ , m.p. 261— $262^{\circ}$  (55% yield), displayed absorption at 1715 cm. $^{-1}$  (C=O) in the infrared but showed

no bands characteristic of nitro-group absorption while it exhibited three main regions of absorption in the ultraviolet region,  $^3\lambda_{\rm max}$  228 ( $\epsilon$  17,600), 240 ( $\epsilon$  18,900), 267 ( $\epsilon$  10,900), 295 ( $\epsilon$  3,290), 304 ( $\epsilon$  3,080), 358 ( $\epsilon$  17,200), 369 ( $\epsilon$  17,600), and 404 m $\mu$  ( $\epsilon$  17,000). On alkaline hydrolysis it afforded 2-(o-carboxyphenylamino)benzotriazole (IV) which was identified by elemental analysis, infrared,

ultraviolet, and mass spectra. On these grounds C<sub>13</sub>H<sub>8</sub>N<sub>4</sub>O is assigned the 13-oxobenzotriazolo[2,1b]benzo[1,2-e]triazine structure (III). This structure is supported by the mass spectrum which shows the molecular ion peak at m/e 236 and a fragmentation pattern compatible with structure (III).

Treatment of (III) with bromine in glacial acetic acid and nitration at 5° with 75% nitric acid gives dibromo- and dinitro-derivatives, respectively.

The formation of (III) may be explained in terms of a nitrene intermediate (II) since the generation of nitrene-like intermediates in the phosphite deoxygenation of nitro-compounds is well known.<sup>4</sup> Thus interaction of the  $\pi$ -electrons of the triazine ring with the developing nitrene would yield 13-oxobenzotriazolo [2,1-b] benzo [1,2-e]triazine (III).

The only other product obtained, in very small yield, from the reaction, was o-nitrophenylphosphoric acid diethyl ester (V). The only reasonable interpretation for the formation of this ester is to postulate a competing Michaelis-Arbusov reaction<sup>5,6</sup> in which nucleophilic attack by phosphorus at the 1'-position of the nitro-substituted aryl group displaces the triazine residue.

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