Measurement of P-P Coupling Constants in cis- and trans-Square Planar Complexes

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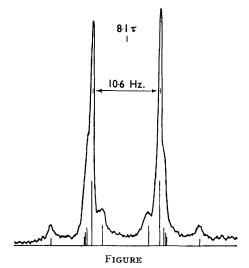
When two phosphine ligands such as MePh₂P are co-ordinated to one acceptor, the ¹H n.m.r. spectrum should be of the $X_nAA'X'_n$ type¹ ($X = {}^{1}H$, $A = {}^{31}P$) from which $J_{AA'}$ can be found in favourable circumstances. Unfortunately, hitherto² all cis-isomers have shown simple doublets ($J_{PP} \sim 0$, $J_{PH'} \sim 0$) whilst trans-isomers usually give triplets ($J_{PP} >> J_{PH}$) allowing only an estimate of J_{PP} to be obtained from the shape of the central line. Here the measurement of a cis P-P coupling from the ¹H n.m.r. spectrum of PdCl₂(Me₃P)₂ and of a trans P-P coupling from the ³¹P resonance spectrum of PdI₂(Me₃P)(Et₃P) is reported.

The ¹H n.m.r. spectrum of PdCl₂(MeP₃)₂ in both dichloromethane and nitrobenzene shows, in

addition to the triplet due to the trans-isomer, the set of lines (see Figure) assignable to the cisisomer. Whilst the main feature of the latter is a sharp doublet, other lines expected of an $X_{\S}AA'X'_{\S}$ system are observable. The spectrum was analysed by the method of Harris¹ to obtain $|J_{AA'}|=8\cdot 0$ and $|J_{AX}-J_{AX'}|=12\cdot 9$ Hz. As the second is greater than $|J_{AX}+J_{AX'}|$ —the separation of the strong doublet—these two couplings must be of opposite sign. The values $|J_{P-Pd-P}|=8\cdot 0$, $J_{P-C-H}=\pm 11\cdot 7_{\S}$ and $J_{P-Pd-P-C-H}=\pm 1\cdot 1_{\S}$ Hz. gave the calculated lines shown in the Figure. Qualitatively similar results were obtained for other Pd^{II} and Pt^{II} complexes.

The methyl resonances of the trans-isomers investigated were triplets, the width of the central

line suggesting that J_{PP} was several hundreds. To evaluate a *trans* P-P coupling, the ³¹P spectrum of an equimolar mixture of $PdI_2(Me_3P)_2$ and



100 MHz. ¹H n.m.r. spectrum of cis-PdCl₂(Me₃P)₂

PdI₂(Et₃P)₂ was measured. After a day, the four lines typical of an AB system had appeared in addition to those of the starting materials. Analysis of the AB system gave |J| = 565 \pm 10 Hz. and shifts the same as those due to $PdI_2(Et_3P)_2$ and $PdI_2(Me_3P)_2$ respectively. In the ¹H n.m.r. spectrum, extra Me₃P resonances appeared as a doublet of doublets, $|J_{PH}| = 10.1$ and $|J_{PH}| = 2.7 \text{ Hz}$. Clearly the new spectrum is due to trans-PdI₂(Me₃P)(Et₃P). The very small change in shifts and closeness of J_{PH} + $J_{P'H}$ (7.4 Hz.) to that observed for trans- $\mathrm{PdI_2(Me_3P)_2}$ (7.2 Hz.) suggests that there is little difference in bonding between the two complexes and, therefore, that J_{PP} must be about 565 Hz. for $trans-PdI_2(Me_3P)_2$ [and $trans-PdI_2(Et_3P)_2$].

The assignment of the larger P-H coupling to the nearer phosphorus atom was confirmed by the methyl resonances of trans-PdI₂(Me₃P)(Et₃P) [present in a mixture of PdI₂(Me₃P)₂ and PdI₂-(Me₃As)₂] which has the wider doublet at the shift expected of the phosphine and the narrower doublet at the shift expected of the arsine.

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¹ R. K. Harris, Canad. J. Chem., 1964, **42**, 2275.

² J. M. Jenkins and B. L. Shaw, J. Chem. Soc. (A), 1966, 770; P. R. Brookes and B. L. Shaw, J. Chem. Soc. (A), 1967, 1079; D. G. Hendricker, R. E. McCarley, R. W. King, and J. G. Verkade, Inorg. Chem., 1966, 5, 639.