

## The Biological Conversion of Loganin into Indole Alkaloids

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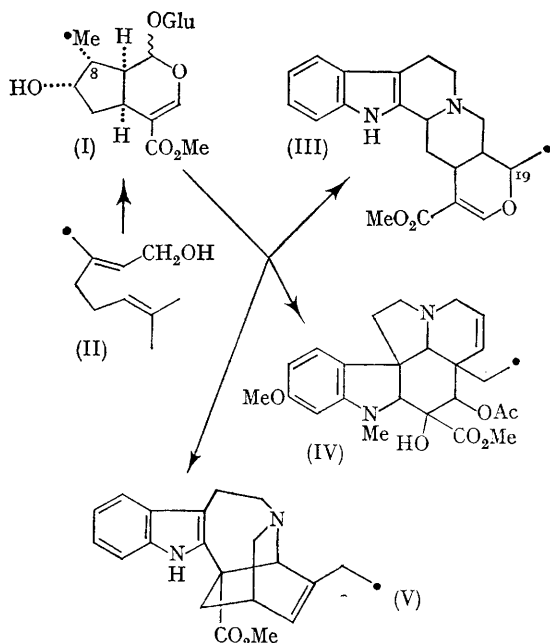
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FOLLOWING the demonstration that geraniol is a good precursor of the non-tryptophan derived portion of different indole alkaloids,<sup>1-4</sup> attention has been centred on the monoterpene glucoside loganin (I)<sup>5,6</sup> as a possible intermediate in the biosynthetic sequence.<sup>7</sup> Preliminary evidence

obtained with *O*-methyl-labelled material was positive but not entirely devoid of ambiguity.<sup>8</sup> The availability of ring-labelled loganin<sup>5</sup> has now made possible a crucial test of this hypothesis.

[<sup>14</sup>C]Loganin carrying the bulk of the label in the C-8 methyl group, as in (I), was isolated from

rhizomes of *Menyanthes trifoliata* after exposure to the methyl-labelled geraniol (II) and purified to constant activity through the penta-acetyl derivative.<sup>5</sup> An aqueous solution of (I) was



supplied to young shoots of *Vinca rosea*. After 6 days, isolation and purification to constant activity gave the following alkaloids: ajmalicine (III; 0.1% incorp.), vindoline (IV; 0.2% incorp.), and catharanthine (V; 0.3% incorp.). Kuhn-Roth degradation of ajmalicine afforded acetic acid containing 93% of the radioactivity. Since C-19 in (III) is known to be derived specifically from C-3 of mevalonate,<sup>2</sup> the label must be restricted to the methyl group. Similar degradation of (IV) and (V) established that all of the label (108 and 107% respectively) was located at the indicated positions. These results document the ability of *Vinca rosea* to convert loganin into the three main types of indole alkaloids. In experiments with labelled sodium mevalonate we have confirmed a previous report<sup>8</sup> that the same plant is able to biosynthesise loganin. Therefore this compound meets two of the important requirements which are characteristic for true intermediates.

Incorporation of ring-labelled loganin into a variety of indole alkaloids has been independently observed by Battersby and his co-workers and is outlined in the accompanying communication.

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<sup>1</sup> A. R. Battersby, R. T. Brown, J. A. Knight, J. A. Martin, and A. O. Plunkett, *Chem. Comm.*, 1966, 346.

<sup>2</sup> P. Loew, H. Goeggel, and D. Arigoni, *Chem. Comm.*, 1966, 347.

<sup>3</sup> E. S. Hall, F. McCapra, T. Money, K. Fukumoto, J. R. Hanson, B. S. Mootoo, G. T. Phillips, and A. I. Scott, *Chem. Comm.*, 1966, 348.

<sup>4</sup> E. Leete and S. Ueda, *Tetrahedron Letters*, 1966, 4915.

<sup>5</sup> S. Brechbühler-Bader, C. J. Coscia, P. Loew, Ch. v. Szczepanski, and D. Arigoni, *Chem. Comm.*, 1968, preceding Communication and refs. cited there.

<sup>6</sup> A. R. Battersby, R. S. Kapil, and R. Southgate, *Chem. Comm.*, 1968, 131.

<sup>7</sup> For a review, cf. A. R. Battersby, *Pure and Appl. Chemistry*, 1967, 14, 117.

<sup>8</sup> A. R. Battersby, R. T. Brown, R. S. Kapil, J. A. Martin, and A. O. Plunkett, *Chem. Comm.*, 1966, 890.