## The Transfer of the Double Bond from 8(9) to 7(8) in Cholesterol Biosynthesis

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RECENT evidence from several laboratories suggests that the transfer of the double bond from 8(9) to 5(6) in cholesterol biosynthesis occurs through the combination of three reactions.<sup>1-4</sup> An initial migration of the double bond from 8(9) to 7(8)which is then followed by an oxidative dehydrogenation<sup>2,3</sup> to give a 5,7-diene system; and finally, a NADPH-linked reduction of the 7(8) bond to furnish the 5(6) double bond of cholesterol.<sup>4</sup> Here we report observations pertinent to the migration of the double bond from 8(9) (IIa) to 7(8) (IIIa). Our main objective was to differentiate between two main mechanisms:

(i) an intramolecular hydrogen transfer mechanism in which one of the C-7 hydrogens is transferred to C-9,

(ii) an "addition-elimination" mechanism in which one of the C-7 protons is lost and one from water is incorporated at C-9 during the biosynthesis. Evidence proves that the migration of the double bond from 8(9) to 7(8) occurs through an "addition-elimination" mechanism.

For our mechanistic studies we needed a <sup>14</sup>Clabelled lanosterol and lanosterol labelled specifically with tritium at C-7. [26,27-<sup>14</sup>C]Lanosterol was synthesised by the method described previously<sup>5</sup> and [7-<sup>3</sup>H<sub>2</sub>]lanosterol was prepared as follows: dry chloroform (10 ml.) was saturated with tritiated HCl gas for 10—15 min. Lanosteryl acetate dibromide (250 mg.) was then added and allowed to equilibrate at room temperature for 4 hr. The resulting mixture of tritiated  $\Delta^{8}$ - and  $\Delta^{7}$ -dibromide acetates was debrominated (Zn/benzene-methanol) and separated on a silica gel-silver nitrate plate. The [7-<sup>3</sup>H<sub>2</sub>]lanosteryl acetate thus prepared incorporates radioactivity at both the C-7 hydrogens. When a mixture of  $[26,27^{-14}C]$ - and  $[7^{-3}H_2]$ lanosterol (IVa and IVb) (T/C = 5.6) was incubated with a 10,000 g. supernatent of rat liver



homogenates under aerobic conditions,<sup>4</sup> the biosynthetic cholesterol (Ia; T/C = 1.9) which was isolated as the dibromide had lost 66% of the tritium activity. That the remaining activity was still located at C-7 of cholesterol was shown by the conversion of the biosynthetic cholesteryl acetate (Ib) into  $3\beta$ -acetoxycholest-5-en-7-one which resulted in the removal of 82% of the remaining activity (T/C = 0.34). This experiment therefore proves that in the migration of the double bond from 8(9) to 7(8) during biosynthesis, one of the hydrogens at C-7 is lost and the other remains undisturbed. (The loss of 66% of the activity instead of 50% shows the uneven incorporation of label at C-7 $\alpha$  and C-7 $\beta$  during the chemical synthesis.)<sup>†</sup>

In order to locate the origin of the C-9 hydrogen during the biosynthesis, non-radioactive cholest-8(9)-enol (IIa) was incubated for 2 hr. with rat liver microsomes<sup>6</sup> under anaerobic conditions in the presence of tritiated water (counts/min./mg. atom of hydrogen =  $3.4 \times 10^8$ ). This resulted

in the incorporation of 10<sup>5</sup> counts/min. in the biosynthetic cholest-7-enol.

After the addition of carrier, biosynthetic  $3\beta$ -acetoxycholest-7-ene (IIIb; counts/min./ mmole =  $6 \times 10^4$ ) was oxidised with chromium trioxide-acetic acid<sup>7</sup> to give a mixture of  $3\beta$ acetoxy-8.9-epoxycholestan-7-one (VI: counts/ min./mmole = 1150) and  $3\beta$ -acetoxy-8,14-epoxycholestan-7-one (V; counts/min./mmole =  $5.8 \times$ 10<sup>4</sup>). The complete retention of activity in (V) and its complete removal in (VI) unambiguously proves that the C-9 hydrogen of the biosynthetic cholest-7-enol (IIIa) contained all the radioactivity.

We therefore conclude that the biological transfer of the 8(9)-double bond to 7(8) occurs through an "addition-elimination" mechanism.

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† The distribution of radioactivity between the two hydrogen atoms at C-7 of lanosterol may vary from experiment to experiment, because of the difficulty involved in maintaining a known concentration of hydrogen chloride in the reaction mixture. Also one sample of labelled lanosterol prepared by the equilibration method contained 15–20% of the activity at positions other than C-7.

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