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## Base-catalysed Isomerisation of Epoxy-esters: The Partial Synthesis of Methyl Coriolate from Methyl Vernolate and of Racemic Methyl Helenynolate from Methyl Crepenynate

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WE have previously speculated on the biological relationship between linoleic acid, the epoxy-acids, vernolic and coronaric, the hydroxy-dienoic acids, coriolic and dimorphecolic, and the several conjugated octadecatrienoic acids.<sup>1</sup> We now report the successful base-catalysed conversion of two epoxy-esters into isomeric unsaturated hydroxyesters.

Methyl vernolate (I) was treated with lithium diethylamide in anhydrous ether for 1 hr. at  $0^{\circ.2}$ . The major product, recovered in about 60% yield by preparative t.l.c., was a hydroxy-ester accompanied by other products including a diethylamide ( $\sim 20\%$ ) with the same u.v. spectrum as the major ester.

$$R^{1}CH-CH\cdot CH_{2}\cdot CH = CHR^{2}$$
 (I)

$$R^{1}CH(OH) \cdot CH = CH \cdot CH = CHR^{2}$$
(II)

$$R^{4}C:C\cdot CH_{2}\cdot CH = CHR^{2}$$
 (III)

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$$R^{1}C:C\cdot CH_{2}\cdot CH-CHR^{2}$$
 (IV)

$$R^{1}C:C\cdot CH = CH \cdot CH(OH) \cdot R^{2}$$
 (V)

$$\mathbf{R}^1 = \mathbf{CH}_3 \cdot [\mathbf{CH}_2]_4 \qquad \mathbf{R}^2 = [\mathbf{CH}_2]_7 \cdot \mathbf{CO}_2 \mathbf{M} \mathbf{e}$$

The hydroxy-ester had a u.v. absorption band at 233 m $\mu$  ( $E_{1\,\rm cm}^{18}$  850 MeOH) and i.r. bands at 3595, 945, and 980 cm.<sup>-1</sup>, indicative of a hydroxyester with a conjugated (*cis, trans*) diene system. The n.m.r. spectrum contained a complex series of peaks in the region  $\tau$  3·3—4·9 (four olefinic protons) and a peak at  $\tau$  5·9 [C(OH)·H]. In all these respects the spectrum resembles that reported for methyl coriolate.<sup>3</sup>

Identity of structure was confirmed by the following: (a) The hydrogenated product gave dodecanedioic and tridecanedioic acids when oxidised with chromic  $acid.^4$  (b) von Rudloff oxidation<sup>5</sup> of the unsaturated hydroxy-acid gave azelaic acid as the only dibasic acid. (c) The ester was readily dehydrated (1,2 and 1,6) by heating with 0.1N-methanolic hydrogen chloride;6 the product was a mixture of conjugated trienes (8,10,12- and 9,11,13-) from its u.v. spectrum and its oxidation to suberic and azelaic acids. (d) Partial reduction with potassium azodicarboxylate? gave two hydroxy-monoene acids; the cis-isomer gave azelaic acid as the only dibasic acid when oxidised and the trans-isomer gave undecanedioic acid along with a small amount of the Ca-acid (probably from a little unchanged dienoic ester remaining in this fraction). This evidence is consistent with the product being methyl 13hydroxyoctadeca-cis-9,trans-11-dienoate (methyl coriolate). Since vernolic acid<sup>8</sup> and coriolic acid<sup>3</sup> have the same absolute configuration at C(13) we

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expect our product to be identical with the natural product.

Methyl crepenynate (III), isolated from Afzelia cuanzensis seed oil by preparative t.l.c. on silica impregnated with silver nitrate,9 was converted into its mono-epoxide (IV) by reaction with a 2M solution of monoperphthalic acid in ether (12 hr.). This was isolated on thin layers of silica and isomerised with lithium diethylamide. The major product, isolated in 50% yield, was a hydroxyester with a *trans*-enyne chromophore ( $\lambda_{max}$  228 m $\mu$  $[E_{1 \text{ cm.}}^{1\%} 600]$  and 238 m $\mu$   $[E_{1 \text{ cm.}}^{1\%} 510]$ , i.r. absorption

bands at 3590 and 950 cm.<sup>-1</sup>). A second product with similar spectroscopic properties is probably the diethylamide of the same ester which we consider to be racemic methyl helenynolate (V).<sup>10</sup>

Other natural products such as the prostaglandins and several short chain poly-ynes contain the group trans-CH(OH)·CH=CH- which might arise, biologically, from the cis-olefin -CH=CH.  $CH_2$ - by expoxidation and rearrangement.

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