## Photochemically Induced Reaction between a t-Butyl Side-chain and an ortho-Nitro-group

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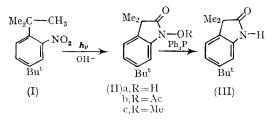
## Germany)

The photochemistry of *o*-nitrobenzyl compounds involving attack at the  $\alpha$ -carbon atom has been investigated recently.<sup>1</sup> Here a photochemical attack at the  $\beta$ -carbon atom of the side-chain is reported.

Irradiation of a 2% solution of 1-nitro-2,5-di-tbutylbenzene (I) in methanol containing sodium hydroxide (0.2M) with a high-pressure mercury arc through a Pyrex filter (3 hr., 12% conversion) followed by alkali extraction of the mixture gives 1-N-hydroxy-3,3-dimethyl-6-t-butylindol-2(3H)one (IIa) (34%), m.p. 176–177°. If the irradiation

is carried out without addition of alkali, the yield drops to 13%. When (I) is refluxed in methanolic sodium hydroxide solution for 24 hr. in the dark, <0.02% of (IIa) is detected by u.v. analysis.

Structure (IIa) is confirmed<sup>†</sup> by a broad concentration-independent OH-absorption from 3500 to 2500 cm.<sup>-1</sup>,  $\nu$ (C=O) at 1692 cm.<sup>-1</sup>; n.m.r. signals at  $\tau$  -0·3 Br(1H), 3·20 (1H), 3·34 (2H), two singlets at 8·72 and 8·75 (together 15H); *m/e* 233 (*M*<sup>+</sup>), 217 (*M* - O, *m*\* 205), 190 (*M* - HCNO, *m*\*



155; 217 — HCN,  $m^*$  166). Addition of ferric chloride to a solution of (IIa) in ethanol produces an intense blue colour<sup>2</sup> which changes into purple upon addition of water and fades completely upon addition of either acid or alkali.

The acetate (IIb) [from (IIa) with acetic anhydride--pyridine], m.p. 122°, has  $\nu(C=O)$  at 1809 and 1728 cm.<sup>-1</sup>, n.m.r. signals at  $\tau$  3.31 (2H), 3.64 (1H), 7.77 (3H), 8.65 (6H, gem-Me), and 8.77 (9H, Bu<sup>t</sup>); m/e 275 (M<sup>+</sup>), 233 (M - CH<sub>2</sub>CO, m\* 197.5), 217, and 190 (233 – HNCO,  $m^*$  155). Methylation of (IIa) with diazomethane proceeds slowly to give (IIc), m.p. 86°, showing m/e 247  $(M^+)$ . Deoxygenation of (IIa) with triphenylphosphine gives triphenylphosphine oxide and a compound of m.p. 192°, v(N-H) 3450 cm.<sup>-1</sup>,  $\nu$ (C=O) 1715 cm.<sup>-1</sup>;  $\tau$  0.18 (1H), 2.95 (m, 3H), 8.60 (s, 6H), 8.68 (s, 9H); m/e 217 (M), 202  $(M - 15, m^* 188), 174 (202 - CO, m^* 150),$  to which the  $\gamma$ -lactam structure (III) is assigned. Furthermore, the u.v. spectrum of (III) is in accord with that<sup>3</sup> of a series of methylated oxindoles.

The neutral fraction of the photolysis mixture consists mainly of starting material and at least seven minor components, which so far have not been characterized.

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 $\dagger$  All compounds gave satisfactory elemental analyses. I.r. spectra were taken in CCl<sub>4</sub> or CHCl<sub>3</sub>; n.m.r. spectra in CDCl<sub>3</sub>. A cold inlet system was used with an Atlas CH 4 mass spectrometer.

<sup>1</sup> J. A. Barltrop and N. J. Bunce, *J. Chem. Soc.* (C), 1968, 1467; H. Morrison and B. H. Migdalof, *J. Org. Chem.*, 1965, **30**, 3996; G. Wettermark, *J. Amer. Chem. Soc.*, 1962, **84**, 3658.

<sup>2</sup> Note the similar reaction of an α-arylamino-nitrone with ferric chloride: H. G. Aurich, *Chem. Ber.*, 1965, 98, 3917. <sup>3</sup> Mme. Ramart-Lucas and Mlle. Biquard, *Bull. Soc. chim. France*, 1935, 2, 1383.