# The Stereochemistry of 6-Bromo-7-oxoditerpenoids 

By R. C. Cambie,* G. R. Clark, D. R. Crump, and T. N. Waters
(Department of Chemistry, University of Auckland, New Zealand)

On the basis of n.m.r. measurements Wheeler and his co-workers ${ }^{1}$ have suggested that the bromide of methyl 6-bromo-12-methoxy-7-oxopodocarpa-8,11,13-trien-16-oate (Ib) has a $\beta$-configuration rather than the $\alpha$-configuration assigned by Wenkert, ${ }^{2}$ Cambie, ${ }^{3}$ and their co-workers. Since Wheeler's arguments appeared to apply equally
well for an $\alpha$-configuration we have determined the structure of (Ib) by $X$-ray methods in order to obtain an unequivocal solution.

The crystals were orthorhombic with $a=10 \cdot 85$, $b=11 \cdot 16, c=15.05 \AA$ and space group $P 2_{1} 2_{1} 2_{1}$. Three-dimensional intensity data were collected with $\mathrm{Cu}-K_{\alpha}$ radiation and the structure was solved
by Patterson and heavy-atom syntheses, the $R$ factor being reduced to $8.8 \%$ by least-squares refinements. The analysis confirms that the bromine is $\alpha$ and shows that the conformation of ring-B is nearly a classical boat (see Figure). $\dagger$ The distances of bromine to the carbonyl $O, C(15)$, $C(5)$, and $C(7)$ atoms are $3.25,3 \cdot 43,2 \cdot 91$, and $2.70 \AA$ respectively, while the carbonyl O is $0.61 \AA$ above, and the $\operatorname{Br} 1.68 \AA$ below the plane of the
shielding by the $C(7)$-carbonyl group alone. For example, conversion of totaryl methyl ether (III) to (IIIa) causes the C(17)-methyl to be shielded by 0.11 p.p.m. while subsequent introduction of a $\mathrm{C}(6)$-bromine atom causes a further deshielding of 0.13 p.p.m. rather than shielding. If the bromine here possessed a $\beta$-configuration [i.e. (IIIc)] a marked deshielding of both the $C(17)$ - and $C(16)$ methyl groups would be expected as is shown below.

Table
N.m.r. data ( $\delta$ ) for methyl groups of diterpenoids

|  | $\mathrm{R}^{1}=\mathrm{H}_{2}, \mathrm{R}^{\mathbf{2}}=\mathrm{R}^{8}=\mathrm{H}$ |  |  |  | $\mathrm{R}^{\mathbf{1}}=\mathrm{O}, \mathrm{R}^{\mathbf{2}}=\mathrm{R}^{\mathbf{d}}=\mathrm{H}$ |  |  |  | $\mathbf{R}^{\mathbf{1}}=\mathrm{O}, \mathrm{R}^{\mathbf{2}}=\mathrm{Br}, \mathrm{R}^{\mathbf{3}}=\mathrm{H}$ |  |  |  | $\mathrm{R}^{\mathbf{1}}=\mathrm{O}, \mathrm{R}^{\mathbf{2}}=\mathrm{H}, \mathrm{R}^{8}=\mathrm{Br}$ |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{C}(15)$ | C(16) | $\mathrm{C}(17)$ |  | (C15) | $\mathrm{C}(16)$ | (C17) |  | $\mathrm{C}(15)$ | $\mathrm{C}(16)$ | $\mathrm{C}(17)$ |  | $\mathrm{C}(15)$ | $\mathrm{C}(16)$ | $\mathrm{C}(17)$ |
| (I) | 1.27 | - | 1.03 | (Ia) | 1.25 | - | $1 \cdot 10$ | (Ib) | 1.53 | - | 0.85 | (Ic) | - | - | - |
| (II) | 0.94 | 0.94 | $1 \cdot 19$ | (IIa) | 0.93 | 1.00 | 1.25 | (IIb) | $1 \cdot 15$ | 1.03 | 1.27 | (IIc) | $1 \cdot 10$ | $1 \cdot 43$ | 1.77 |
| (III) | 0.95 | 0.95 | 1.21 | (IIIa) | 0.91 | 1.01 | $1 \cdot 10$ | (IIIb) | $1 \cdot 15$ | 1.08 | 1.23 | (IIIc) | - | - | - |
| (IV) | - | 1.25 | 1.20 | (IVa) | - | 1.33 | 1.26 | (IVb) |  | 1.50 | 1.26 | (IVc) | - | - | - |

aromatic ring. The dihedral angle $\mathrm{H}_{\alpha}(\mathrm{C} 5)-\mathrm{C}(6) \mathrm{H}_{\beta}$ is $154 \pm 8^{\circ}$ in agreement with the observed coupling constant of $J=7 \mathrm{c} . / \mathrm{sec}$. from the n.m.r. spectrum. Comparison of (Ib) with (Ia) (see Table) shows that the C(15)-methyl of (Ib) is deshielded by 0.28 p.p.m., an amount comparable to that induced by a 1,3-diaxial interaction in a steroid. ${ }^{4}$ Oxidation of (I) to (Ia) causes a deshielding of the $\mathrm{C}(17)$-methyl group but introduction of a $\mathrm{C}(6)$-bromine (Ib) then causes a shielding with respect to (Ia). This can be interpreted as a change in conformation of ring-B from a half-boat (Ia) to a boat, the C(17)-methyl of (Ib) thus entering the cone of shielding of the carbonyl group. However, such first-order approximations do not correlate with the chemical shifts observed for related diterpenoids [e.g. (II), (III), and (IV)], and thus the shift of the $C(17)$-methyl group cannot be satisfactorily interpreted in terms of the expected


Figure

(I)




Re-investigation of the bromination of sugiyl methyl ether (II) which earlier gave a $6 \beta$-bromoderivative, ${ }^{3}$ has now afforded two epimers, m.p.'s $199^{\circ}$ and $156^{\circ}$, which have been assigned the $6 \beta$ -bromo- and $6 \alpha$-bromo-structures, (IIb) and (IIc), respectively. Whereas the $\mathrm{C}(16)$ - and $\mathrm{C}(17)$-methyl groups of the $\beta$-epimer are deshielded by 0.43 and 0.03 p.p.m. those of the $6 \alpha$-isomer are only deshielded by 0.10 and 0.03 p.p.m. Moreover the $\mathrm{C}(15)$-methyl group of (IIc) shows a marked deshielding ( 0.22 p.p.m.) as would be expected for a $6 \alpha$-bromo-configuration. Dreiding models of the $6 \beta$-isomer (IIb) give an $\mathrm{H}_{\alpha} \mathrm{C}(5)-\mathrm{C}(6) \mathrm{H}_{\alpha}$ dihedral angle of $c a .60^{\circ}$ for a ring-в half-boat conformation in agreement with a coupling constant of $3 \mathrm{c} . / \mathrm{sec}$. A model of the $6 \alpha$-isomer (IIc) with a ring-B boat conformation gives a didedral angle of $c a .130^{\circ}$ in agreement with the observed coupling constant of $8.5 \mathrm{c} . / \mathrm{sec}$.
$\dagger$ Full details of this analysis will be published elsewhere.

Thus configuration of $\mathrm{C}(6)$-bromo-derivatives of 7-oxoditerpenoids is best assigned by a consideration of the deshielding effects of the bromine atom. Bromination of methyl 7-oxoabieta-8,11,13-trien15 -oate (IV) produces no shift in the C(17)-methyl resonance but the $C(16)$-methyl is deshielded by 0.17 p.p.m., and hence a $6 \alpha$-bromo-configuration can be assigned here. A coupling constant of
$12.5 \mathrm{c} . / \mathrm{sec}$. indicates a dihedral angle $\mathrm{H}_{\alpha} \mathrm{C}(5)^{-}$ $\mathrm{C}(6) \mathrm{H}_{\beta}$ of about $150^{\circ}$, a result consistent with a ring- $B$ boat conformation. This conformation would be similar to that of (IIc) but with the C(6)bromine twisted slightly upwards to increase the dihedral angle as a result of interaction with the $\mathrm{C}(15)$-carboxymethyl group.
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