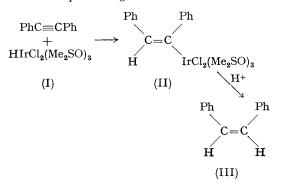
Stepwise and Catalytic cis-Hydrogenation of an Alkyne under Single Phase Conditions, including the Isolation of an Intermediate, an Alkyne-Transition Metal Hydride Adduct

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cis-Alkenes are the main products first formed in the reduction of alkynes by molecular hydrogen in the presence of metals¹ or certain complexes of ruthenium² or rhodium.³ Such reactions are generally thought to proceed via intermediates in which carbon is bonded temporarily to a metal atom. A stable intermediate has now been obtained by keeping a solution of diphenylacetyland hydrodichlorotris(dimethyl ene (72 mg.)sulphoxide)iridium (I) (100 mg.) in propan-2-ol (5 ml., containing 2% water) at 73° for 90 minutes. The adduct separates during the reaction; after cooling the yield is 80 mg. Crystallisation from chloroform-pentane gives the adduct as cream



needles, m.p. 196-198°, correct analysis for C₂₀H₂₉Cl₂O₃S₃ (cf. II), H-Ir group absent (i.r. and n.m.r. evidence). Treatment of this compound with boiling methanol containing concentrated hydrochloric acid (10% v/v) for 2 min. gives cisstilbene (> 90%), identified by gas chromatography and by preparation of the methoxychloromercuri-compound, PhCH(OMe)CH(HgCl)Ph, m.p. and mixed m.p. 145-146° (lit.,4 m.p. 143°). If retention of configuration occurs in the replacement of the iridium-containing group by hydrogen (as is general for the protonolysis of vinyl-metal bonds), the adduct can be formulated as (II); its formation then involves *cis*-addition of the metal hydride to the triple bond.

cis-Hydrogenation can be effected in one operation by having acid present in the reaction solution together with the hydride, in catalytic proportion. A reaction starting with a molar ratio of 20:1:2 of diphenylacetylene:hydride: hydrogen chloride in propan-2-ol (2% water) at 65° gives cis-stilbene (ca. 35% after 3 hr. and ca. 60% after 8 hr., the corresponding yields of transstilbene being ca. 1 and 2%). The adduct (II) can be used in place of hydride to give closely similar results.

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