Correlation between Phosphorus-Tungsten Coupling Constants and Carbonyl Stretching Frequencies in Phosphorus-ligand Derivatives of Tungsten Hexacarbonyl

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We reported earlier that there is a correlation between the most intense (E mode) carbonyl stretching frequencies in tertiary phosphine-tungsten pentacarbonyl compounds and the tungsten-183-phosphorus-31 nuclear spin-spin coupling constants in the compounds.¹ All the ligands were tertiary phosphines: hence, J_{W-P} varied by only 80 Hz while the CO stretching frequencies (E mode) had a range of only 8 cm.⁻¹.

We have extended the study over a significantly larger range of J_{W-P} (211 Hz) and v_{CO} (25 cm.⁻¹) by including tertiary phosphites, mixed phenylalkoxyphosphines, and tris(dialkylamino)phosphines as ligands. The results are given in the Table and the data are plotted in the Figure. The least-squares line for all the points is shown by the solid line: $v_{\rm E}$ (cm.⁻¹) = 1918 + 0.087 J (Hz), with a standard deviation (σ) of 3.7 cm.⁻¹. The correlation of all the data collectively is not as good (correlation coefficient, r = 0.86) as that of the data for the individual classes of ligands. For example, the tertiary phosphines give r =0.97 and $\sigma = 0.6$ cm.⁻¹, whereas the phosphites have r = 0.98 and $\sigma = 1.2$ cm.⁻¹. The qualitative relationship is, however, clear. Thus, the less basic ligands of phosphorus (the poorer σ -donors and better π -acceptors) have the larger phosphorus-tungsten coupling constants. Since the use of carbonyl stretching frequencies for determining the π -acceptor ability of ligands has been under attack recently,² the fact that there is a correlation between v_{CO} and J_{W-P} does not prove that J_{W-P} is a measure of π -acceptor ability. One would expect the best σ -donors to have the largest coupling constants, because it is generally supposed that the coupling mechanism is transmitted through σ -bonds with s-electronic character.³ The experimental results indicate the opposite. Until theoretical developments show otherwise, we conclude that the σ -bond is strengthened by a synergic π -interaction and/or

The carbonyl stretching frequency (E mode) and phosphorustungsten coupling constant in LW(CO),

	Ligand		E (cm1)	$J_{\mathrm{W-P}}$ (Hz)	
1	(PhO) ₃ P	• •	1959	411	
2	(MeO) ₃ P		1953	398	
3	(EtO) ₃ P	••	1949	391	
4	(BunO) ₃ P	• •	1947	390	
5	(PriO) ₃ P		1944	378	
6	(MeO) ₂ PhP		1950	323	
7	$(Me_2N)_3P$	• •	1950	297	
8	$(Et_2N)_3P$		1948	296	
9	MeŌ·Ph ₂ P		1945	280	
10	Ph ₃ P	• •	1942	280	
11	MePh ₂ P	••	1939	245	
12	$EtPh_2P$	• •	1938	240	
13	Ph_2Pr^iP		1937	240	
14	Ph ₂ Bu ⁿ P		1938	250	
l5	Ph ₂ Bu ^t P		1937	240	
16	PhBu ₂ ⁿ P	• •	1937	235	
17	Bu₃n₽		1934	200	

there is coupling transmitted via π -bonds or another mechanism.



FIGURE. Correlation of v_{CO} (E mode) and J_{W-P} for LW(CO)₅ compounds (-– —, phosphites;– --, all compounds; ----phosphines). The numbers correspond to the compounds in the Table

It seems (Figure) that J_{W-P} is a better measure of π -acceptor ability (or inversely as σ -donor ability) than v_{CO} . In particular, J_{W-P} increases as follows: R_3P <RPh₂P < Ph₃P \sim (RO)Ph₂P < (R₂N)₃P < (RO)₂PhP < $(RO)_{3}P < (PhO)_{3}P$, which appears to be a quite reasonable order based on inductive effects of the substituents on phosphorus. However, the order of increasing v_{CO} is not the same. Indeed, the amino-phosphines and dimethoxyphenylphosphine are included in the span of phosphites, which is not expected.

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