

An Alternative Synthesis of (\pm)-Dehydrogriseofulvin by Enzymic Phenolic Oxidation by Homogenised Potato Peelings

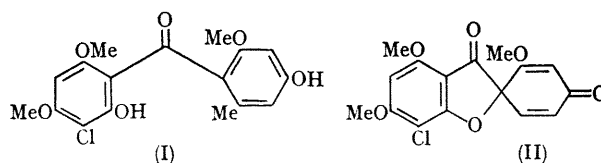
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We have reported¹ that head-to-tail coupling of 1,2,3,4-tetrahydro-7-hydroxy-1-(4-hydroxyphenethyl)-6-methoxy-2-methylisoquinoline by enzymic phenolic oxidation gives the expected promelanthoidine. Similarly, *N*-methylclaurine undergoes enzymic phenolic oxidation with homogenised potato peelings-hydrogen peroxide.² Since the formation of dehydrogriseofulvin (II), a precursor of griseofulvin, has been assumed to be by phenol oxidation of griseophenone A (I),³ many attempts to obtain (II) *in vitro* by phenol oxidation of (I) with potassium ferricyanide⁴ and horseradish-hydrogen peroxide⁵ have been carried out. Here, we report an alternative conversion of (I) into (II) with homogenised potato peelings-hydrogen peroxide.

A phosphate-buffered solution (pH 7.6–7.7) of griseophenone A (I), derived from natural griseofulvin, was mixed with homogenised potato peelings in the presence of hydrogen peroxide and ethanol and set aside for 2 days at 4–8°. A further and similar quantity of homogenised potato peelings was then added. The mixture was worked

up after 4 days and gave starting material (I) and (\pm)-dehydrogriseofulvin (II), m.p. (benzene) 288–209° [lit.,⁷ m.p. 288–290°], $[\alpha]_D = 0^\circ$, (12% yield) whose i.r. (in CHCl_3) and n.m.r. (in CDCl_3) spectra were superimposable on those of an authentic sample. The compound was optically inactive.



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