

The Crystal Structures of Eight-co-ordinate (Tetrakis-*NN*-diethyldithiocarbamate) Complexes

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Summary The crystal structures of thorium tetrakis-*(NN*-diethyldithiocarbamate) and of tetraethylammonium neptunium tetrakis-*(NN*-diethyldithiocarbamate) have been determined.

THE preparation and unit-cell dimensions of quadrivalent actinide tetrakis-*(NN*-diethyldithiocarbamate) complexes of the type $M\text{dte}_4$ ($M = \text{Th, U, Np, and Pu}$) and of trivalent actinide and lanthanide complexes of the type $\text{NEt}_4M\text{dte}_4$ ($M = \text{La-Lu}$ except Pm, Np, and Pu) have recently been reported.¹⁻³ I.r. spectral studies have indicated that the *NN*-diethyldithiocarbamate groups are probably bidentate, implying an eight-co-ordinate environment for the central metal atom. Since the co-ordination of eight sulphur atoms to a metal has not previously been recorded it seemed worthwhile investigating the structures of Thdte_4 and $\text{NEt}_4\text{Npdte}_4$.

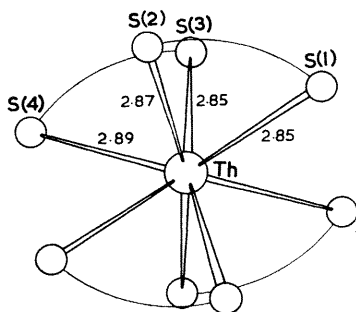


FIGURE 1. The co-ordination polyhedron in Thdte_4 viewed down the crystallographic two-fold axis.

were deduced from Patterson syntheses and the co-ordinates of the light atoms from subsequent Fourier syntheses. Refinement by full-matrix least-squares has reduced the residuals to 0.112 for Thdte_4 and 0.117 for $\text{NEt}_4\text{Npdte}_4$. Refinement is being continued.

The configuration of the sulphur atoms around the thorium atom is shown in Figure 1 and the configuration about the neptunium atom in Figure 2. In the thorium complex the sulphur atoms are arranged as a distorted dodecahedron. However, the co-ordination polyhedron in the neptunium complex is different from any previously observed eight-co-ordinate polyhedron. It is best regarded as an almost plane pentagonal arrangement of five sulphur atoms, $\text{S}(1)$ — $\text{S}(5)$, with one sulphur atom, $\text{S}(6)$, located below, and the remaining two, $\text{S}(7)$ and $\text{S}(8)$, above, this plane. The neptunium atom is slightly above the plane of the five sulphur atoms.

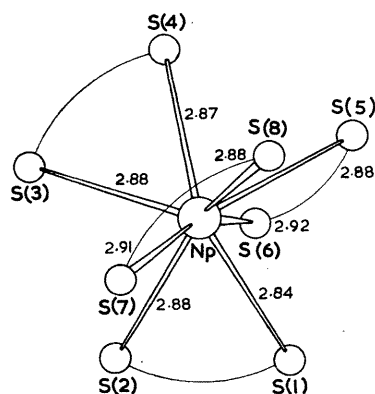


FIGURE 2. The co-ordination polyhedron in $\text{NEt}_4\text{Npdte}_4$.

The complexes were prepared as previously described^{2,4} and were recrystallised from anhydrous benzene [Thdte_4] or methylene dichloride [$\text{NEt}_4\text{Npdte}_4$]. Thdte_4 is monoclinic, space group $C2/c$, $a = 19.16$, $b = 11.74$, $c = 16.11$ Å, $\beta = 116.42^\circ$, $Z = 4$ and $D_c = 1.687$ g.cm.⁻³; $\text{NEt}_4\text{Npdte}_4$ is monoclinic, space group $P2_1/c$, $a = 11.68$, $b = 20.49$, $c = 16.11$ Å, $\beta = 116.5^\circ$, $Z = 4$, $D_c = 1.532$ g.cm.⁻³.

Three dimensional X-ray data were collected by the equi-inclination Weissenberg technique using $\text{Cu-K}\alpha$ radiation ($\lambda = 1.5418$ Å). The intensities were estimated visually and after application of the usual geometrical corrections 1209 and 1263 nonzero intensities were obtained for Thdte_4 and $\text{NEt}_4\text{Npdte}_4$, respectively. The metal co-ordinates

The mean Th-S and Np-S distances are 2.86 and 2.89 Å, respectively; the remaining interatomic distances do not differ significantly from those reported for other *NN*-diethyldithiocarbamate complexes.⁵ The mean M-S distances are very similar to the mean U-S distance (2.80 Å) recently reported⁶ for the complex $\text{NMe}_4\text{UO}_2\text{dte}_3$.

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