## Synthesis of 8-Cyanoheptafulvene

By Masaji Oda and Yoshio Kitahara\*

(Department of Chemistry, Faculty of Science, Tohoku University, Sendai, Japan)

THE only simple 8-monosubstituted heptafulvene previously reported, 8-vinylheptafulvene, is unstable. 8-Cyanoheptafulvene was expected to be more stable, and suitable for theoretical and chemical investigation of heptafulvenes.

Attempts to abstract hydride from 7-cyanomethylcycloheptatriene (I)<sup>2</sup> with triphenylmethyl fluoroborate were unsuccessful, possibly because the cyanomethyl group hinders attack by triphenylcarbonium ion at C-7. Therefore compound (I) was converted into a mixture of 1-cyanomethylcycloheptatriene (II) and 3-cyanomethylcycloheptatriene (III) (ca 1:4 from n.m.r. spectrum) by heating at 140-150°. The mixture smoothly afforded, as expected,

$$\begin{array}{c} CH_2 \cdot CN \\ CH_2 \cdot CN \\$$

cyanomethyltropylium fluoroborate (IV) (85%) with triphenylmethyl fluoroborate (CHCl<sub>3</sub>, room temperature).

The fluoroborate (IV) was moderately stable [nonhygroscopic colourless crystals, m.p. 108-109° (decomp.),  $\nu_{\text{max}}$  (KBr), 3025, 2200, 1548, 1490, 1449, 1060, 728, 710, and 694 cm<sup>-1</sup>; n.m.r. (CF<sub>3</sub>CO<sub>2</sub>H)  $\tau$  0.78 (s, 6H) and 5.28 (s, 2H)]. The n.m.r. values are in good agreement with those for the methylene and the seven-membered-ring protons of both the azulenium<sup>3</sup> and the heptalenium ion.<sup>4</sup>

8-Cyanoheptafulvene (V) (80%) was easily obtained from the fluoroborate (IV) (Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>) as a red liquid [u.v. (iso-octane):  $\lambda_{\text{max}}$  562sh (log  $\epsilon$  1.67), 518sh (2.30), 478sh (2·43), 444sh (2·67), 412sh (2·71), 353sh (4·01), 337 (4·31), 326 (4·30), 264sh (3·67), 253sh (3·96) and 237nm (4·00); i.r. (film): ymax 3050w, 3030m, 2205s, 1640m, 1555s, 1520m, 1470m, 1273m, 1250m, 863m, 828m, 763s, and 749m cm.-1; n.m.r. (CCl<sub>4</sub>):  $\tau$  3.42 br (d, J 12.5 Hz H-1), 3.97 (m, H-2-H-6), and 5.52 (s, H-8)].

The data are consistent with the structure shown. The ring protons ( $\tau$  3.97) were shifted ca.  $\tau$  0.4 to lower field compared with those of (I),1 indicating a greater contribution of the dipolar resonance structure than in (I). The n.m.r. spectrum of (V) in trifluoroacetic acid exhibited signals at  $\tau$  0.76br (s, 6H) and 5.37br (s, 2H), indicating the protonated structure (VI).

With dimethyl acetylenedicarboxylate (refluxing xylene, higher temperature than the case of parent hydrocarbon heptafulvene)<sup>5</sup> (V) gave dimethyl 1-cyanoazulene-2,3-dicarboxylate (VII) [red-purple needles, m.p. 175-176°,  $v_{\rm max}$  (KBr), 3020, 2220, 1740, 1708, 1455, 786, and 745 cm. $^{-1}$  $\lambda_{\text{max}}$  (EtOH), 233 (log  $\epsilon$  4·51), 266 (4·29), 292 (4·59), 302 (4.71), 337 (3.81), and 528nm. (2.90)].

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