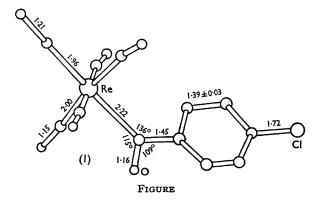
Crystal Structure of p-Chlorobenzoylpentacarbonylrhenium

By I. S. Astakhova, * A. A. Johannsson, V. A. Semion, Yu. T. Struchkov, K. N. Anisimov, and N. E. Kolobova (Institute of Organo-Element Compounds, Academy of Sciences of the U.S.S.R., Vavilova 28, Moscow, U.S.S.R.)

Summary The σ-bond Re-C(benzoyl) length 2.22 Å coincides with the covalent radii sum 2.25 Å.

RECENTLY we reported^{1,2} on the synthesis and properties of halogen-containing benzoyl derivatives of manganese and rhenium carbonyls, and now we present the results of an X-ray study of p-chlorobenzoylpentacarbonyl rhenium, p-Cl·C₆H₄·CORe(CO)₅ (I). The crystals are yellow needles, triclinic, with $a = 10.15 \pm 0.02$, $b = 11.16 \pm 0.02$, c = $6\cdot10\pm0\cdot03$ Å, $\alpha=97\pm1$, $\beta=95\pm1$, $\gamma=81\pm1^{\circ}$,



 $U = 677 \text{ Å}^3$, $D_{\rm m} = 2.18$, $D_{\rm c} = 2.28$, Z = 2, M = 465.84. Intensities of ca. 800 non-zero independent reflexions were estimated visually without regard to the absorption correction (an equi-inclination Weissenberg goniometer, unfiltered copper radiation). The structure was solved by the usual heavy-atom technique and refined by the full-matrix leastsquares method with isotropic temperature factors, to R 0.13 with the overall temperature-factor $B = 5.2 \text{ Å}^2$. Standard deviations in bond lengths are 0.01 for Re-C, 0.02 for C-O, 0.03 Å for C-C, and in bond angles $2-3^{\circ}$ depending on the atomic numbers.

The Re atom co-ordination is that of a slightly distorted octahedron (see Figure). The mean Re-CO (equatorial) bond-length is 2.00 Å with C-O 1.15 Å, R-CO (axial) is 1.96 Å with C-O 1.15 Å. These differences in bond lengths are analogous with, although less than, those found in other pentacarbonyl metal derivatives, XM(CO)₅^{3-,5} which suggest greater back-donation $d_{\pi}(M) \to \pi^{*}(CO)$ to the apical CO-group. The mean value of the C(equatorial) -Re-C(axial) angles is $90.7 \pm 2.4^{\circ}$; and the Re-C-O angles are somewhat distorted (mean value 172°). The Re-C σ -bond length is $2 \cdot 22$ Å.

The authors thank B. P. Bir'yukov for helpful discussions.

(Received, February 7th, 1969; Com. 166.)

- ¹ A. N. Nesmeyanov, K. N. Anisimov, N. E. Kolobova, and A. A. Johannsson, *Doklady Akad. Nauk S.S.S.R.*, 1967, 175, 1293.

 ² B. V. Lokshin, K. N. Anisimov, N. E. Kolobova, A. A. Johannsson, and A. N. Nesmeyanov, *Izvest. Akad. Nauk S.S.S.R.*, Ser.
- khim., 1968, 2247.
- M. R. Churchill and R. Bau, Inorg. Chem., 1967, 6, 2086.
 B. P. Bir'yukov, K. N. Anisimov, Yu. T. Struchkov, N. E. Kolobova, O. P. Osipova, and M. Ya. Zakharova, Zhur. strukt. Khim.,
- 1967, 8, 554.

 ⁵ R. E. Bryan, J. Chem. Soc. (A), 1968, 696.
- ⁶ N. W. Alcock, Chem. Comm., 1965, 177.