

## The Brevianamides: a New Class of Fungal Alkaloid

By A. J. BIRCH\* and J. J. WRIGHT

(Research School of Chemistry, Australian National University, P.O. Box 4, Canberra, A.C.T.)

**Summary** Formulae are suggested for brevianamide-A and brevianamide-E.

*PENICILLIUM BREVI-COMPACTUM* gives a low yield of a mixture of related neutral compounds, mostly pigments, brevianamides A—E. Brevianamide-A is a yellow pigment,  $C_{21}H_{23}N_3O_3$ , with a  $\psi$ -indoxyl chromophore, the presence of which is confirmed by borohydride reduction and acid dehydration to deoxybrevianamide-A.<sup>1</sup> Chemical and spectral evidence was obtained for the presence of a 2,3-disubstituted indole ring in the latter. No double bonds other than those in the  $\psi$ -indoxyl system could be detected in brevianamide-A, and ozonolysis and methylation gave  $C_{17}H_{24}N_2O_6$ , as expected for fission of the pyrrole ring, loss of the benzene ring and indolic N.

Amide bands at 1670 and 1690  $cm^{-1}$ , lack of the amide-II band, and reduction by diborane of deoxybrevianamide-A to the basic  $C_{21}H_{27}N_3$ , suggested the presence of a diketopiperazine ring. Hydrolysis gave no simple amino-acids,

but base gave an amino-acid which rapidly regenerated brevianamide-A on acidification. No evidence for the NCHCO unit of a diketopiperazine could be obtained ( $\tau$  5.5—6.1 absent); there was no exchange with MeOD and base, and no racemisation. With six rings to be expected, these positions were assumed to be involved in ring-closures.

Chemical and spectral evidence of possible biogenetic precursors was sought. The presence of  $CMe_2$  in brevianamide-A ( $\tau$  8.70, 9.10), with differences in environment, obliterated in deoxybrevianamide-A ( $\tau$  8.48), indicates proximity of the  $Me_2$  to the CO and NH of the  $\psi$ -indoxyl system. Both compounds show spectral evidence for an isolated  $CH_2$  group, which in the latter compound is adjacent to the aromatic ring. Lack of other CMe resonances and the presence of a resonance at  $\tau$  6.7 (2H) similar to that due to  $NCH_2$  in proline dimer suggested a possible proline unit. The mass spectrum showed a major loss of  $C_5H_9$  and suggested a terpene unit.

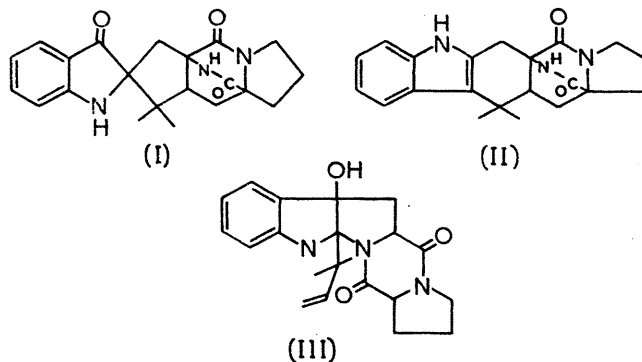
The evidence above, biogenetic considerations, and

relations to known fungal metabolites suggested a number of possible formulae including (I). Feeding experiments were conducted with [ $3\text{-}^{14}\text{C}$ ]tryptophan, [ $2\text{-}^{14}\text{C}$ ]mevalonic lactone, [ $2\text{-}^{14}\text{C}$ ]acetate, [ $Me\text{-}^{14}\text{C}$ ]methionine and L- [ $U\text{-}^{14}\text{C}$ ]proline (incorporations 0.6, 0.003, 0.025, 0.000, and 0.09% respectively). In view of the low yields (2–10 mg./l.) the positive incorporations were considered to be significant.

Examinations in detail of mass spectra and  $^1\text{H}$  n.m.r. spectra, checked by irradiation, strongly support (I) for brevianamide-A and (II) for deoxybrevianamide-A. Notably dibromination of the aromatic ring, and methylation of the CONH and exchange with MeOD, permitted assignments of many mass spectral peaks.

The colourless brevianamide-E,  $\text{C}_{21}\text{H}_{25}\text{N}_3\text{O}_3$ , has an indoline chromophore, unaffected by acid or base. It contains OH ( $3600\text{ cm.}^{-1}$ ), NH ( $3370\text{ cm.}^{-1}$ ), and amide ( $1680$ ,  $1690\text{ cm.}^{-1}$ ) but lacks the amide-II band at  $1550\text{ cm.}^{-1}$ , indicating the likelihood of the presence of a diketopiperazine ring. In this case the NCHCO hydrogens are detectable in the  $^1\text{H}$  n.m.r. spectrum, one coupled to the non-equivalent protons of an otherwise isolated  $\text{CH}_2$  (ABX system  $J_{\text{AB}}$  13,  $J_{\text{AX}}$  11,  $J_{\text{BX}}$  8Hz.). Hydrolysis gave one molar equivalent of proline. The terpene unit is present as  $\text{CMe}_2\text{CH:CH}_2$ ;  $\tau$  8.75 s (2Me) and three vinylic H which

could be assigned to  $\text{CH:CH}_2$ , confirmed by hydrogenation to Et. The mass spectrum shows a major loss of 69 units,



increased to 71 in the dihydro-derivative. Reduction with zinc and acetic acid gave deoxybrevianamide-E with removal of one oxygen and production of an indole chromophore. We consider brevianamide-E to be (III) and this is supported by detailed spectra.

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<sup>1</sup> B. Witkop and J. B. Patrick, *J. Amer. Chem. Soc.*, 1951, **73**, 2188.