Cycloaddition Reactions of Tropone with Enamines

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WE report on the reactions of tropone (I) with enamines. Reaction with 1-morpholinocyclohexene (II) without solvent (or in EtOH, C_6H_6 or tetrahydrofuran) at room temperature gave a l: l addition product (III) (77% yield), m.p.103—104°, ν_{max} (KBr) 3030, 3015, 1631, 1617, 1196, 1118, 790, and 702 cm.⁻¹, λ_{max} (cyclohexane) 213 (ϵ 21,900) and 307 (4140) nm. The i.r. and u.v. data are comparable with those of 1-ethoxycycloheptatriene,¹ indicating the presence of a 1-alkoxycycloheptatriene moiety in (III). The n.m.r. spectrum (Table) further confirmed the structure. The large coupling constant (9.2 Hz.) between 7-H and 8-H suggests a *cis*-relationship between these protons. 8-H and the morpholino-group are probably cis, as the trans-form cannot be constructed from Dreiding models. Compound (III) was easily hydrolysed by dilute mineral acids to give a diketone 2-(2'-oxocyclohexyl)cyclohepta-3,5-dien-1-one (IV) (89% yield), m.p. 108—109°, ν_{max} (KBr) 3025, 1705, 1597, and 705 cm.⁻¹; λ_{max} (EtOH) 235 (ϵ 5350) and 285 (550) nm.; n.m.r. (100 MHz., CDCl₃) τ 3·76 (m, 4-H, 5-H), 4·22 (ddd, 11·0, 6·7, 3·8, 6-H), 4·52 (dd, 10·5, 5·2, 3-H), 6·4—7·1 (m, 7a- and 7b-H, 2-H, 1-H) and 7·5—8·9 (m, 8H). Hydrogenation (Pd/C) of (IV) gave a tetrahydro-compound (V); colourless liquid; i.r., carbonyl at 1715 (six-membered) and 1698 cm.⁻¹ (seven-membered). Hydrolysis of (III) with

Compound (III)	Proton assignment 2-H 3—5-H 6-H 7-H 8-H Morpholino-group	Chemical shift (τ) $4 \cdot 40$ $3 \cdot 60 - 4 \cdot 05$ $5 \cdot 14$ $ca. 7 \cdot 2$ $7 \cdot 48$ $7 \cdot 8 - 8 \cdot 7$ $6 \cdot 35$ $7 \cdot 20$	Multiplicity dt m ddd ddd m (8H) m (4H) m (4H)	Coupling constants (Hz.) $J_{2,8} = 6 \cdot 2$ $J_{2,4} = J_{2,7} = 1 \cdot 5$ $J_{5,6} = 10 \cdot 0$ $J_{6,7} = 3 \cdot 5$ $J_{4,6} = 1 \cdot 5$ $J_{7,8} = 9 \cdot 2$
(IX)	1-H 3-H 4-H 5-H 6-H 7-H 8-H 9-H CH ₃ Morpholino-group	$\begin{array}{c} 6\cdot85\\ 4\cdot25\\ 3\cdot10\\ 6\cdot60\\ 3\cdot43\\ 4\cdot01\\ 8\cdot06\\ 7\cdot56\\ 8\cdot83\\ 6\cdot35\\ 7\cdot2-7\cdot8\end{array}$	ddd ddd ddd ddd ddd bq dd d d m(4H) m(4H)	$J_{1,8} = 2 \cdot 3, J_{5,9} = 3 \cdot 0$ $J_{2,4} = 11 \cdot 3, J_{7,8} = 1 \cdot 0$ $J_{4,5} = 8 \cdot 7, J_{8,9} = 7 \cdot 0$ $J_{3,5} = 0 \cdot 5, J_{8,Me} = 6 \cdot 8$ $J_{5,6} = 7 \cdot 0$ $J_{6,7} = 8 \cdot 2$ $J_{1,7} = 7 \cdot 2$ $J_{1,6} = J_{5,7} = 1 \cdot 0$ $J_{1,8} ca. 0 \cdot 5$

TABLE N.m.r. spectral data (100 MHz, CDCl₃)

Abbreviations: d, doublet; t, triplet; m, multiplet; b, broad; q, quintet.



 $D_2O-D_2SO_4$ (5°, 5 min.) resulted in the introduction of one deuterium at C-7 of (IV), the n.m.r. signals of 6-H changing to double-doublet from doublet-doublet-doublet, and the integrated area at τ 6·4—7·1 (4H) decreasing to 3H. Under the same conditions, deuterium exchange in (IV) was negligible. This result suggests that the mechanism of the hydrolysis is as shown in the Scheme.

With 1-morpholinocyclopentene (VI), (I) also gave a 1:1 addition product (VII) in high yield.

However, with 1-morpholinopropene (VIII) under the same conditions, (I) gave another type of 1:1 addition product (IX) (66% yield), m.p. 129-130°, vmax (KBr) 3040, 1660 and 1626 cm.⁻¹, λ_{max} (EtOH) 215, (sh) (ϵ 6400) and 291 (2480) nm.; λ_{max} (EtOH-0.2N-HCl) 224 (6580) and 258 (sh) (1260) nm. The i.r. data indicate the presence of an $\alpha\beta$ -unsaturated carbonyl group in (IX). The change of the u.v. spectrum in acidic solution from that in neutral solution (the former is typical for the bicyclo[3,2,2]nona-3,6-dien-2-one (8,9-dihydrohomobarrelenone) structure),2 suggests spatial interaction of the morpholino-group with the $\alpha\beta$ -unsaturated carbonyl group. The exact structure was elucidated by 100 MHz. n.m.r. and double-resonance spectra (Table). The coupling constants, J_{8-9} 7.0, J_{1-8} ca. 0.5 and J_{5-9} 3.0 Hz. suggest[†] that 8-H (exo) and 9-H (endo) are *trans*. In this configuration, the nitrogen of the morpholino-group and C-4 are near enough (2.50 Å in the Dreiding model) to interact, accounting for the u.v. spectra.

Possible mechanisms for these addition reactions are shown in the Scheme.

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