Emission Spectrum of the AsSb Molecule

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A SYSTEM of red-degraded bands in the region 3660—3860 Å attributed to the new molecule AsSb has been observed in a microwave discharge through a flowing mixture of AsCl₃ and SbCl₅. The spectrum has a simple vibrational structure and is very similar to that of AsP,¹ PSb,² and the known AsN (A¹ Π -X¹ Σ ⁺) system.³

Pressure of approximately $20 \ \mu$ m. each of AsCl₃ and SbCl₅ were used. The microwave discharge was generated by a Burdick model M.W/200 microwave unit operating at 2450 MHz. and rated at 125 w. The emission spectrum was originally recorded on Ilford HP 3 plate using a Hilger medium quartz spectrograph but has since been re-photographed in the second order of a 6.5 m. concave grating spectrograph with a dispersion of 0.5 Å/m. at the Institute of Physics, University of Stockholm, Sweden.

The Table gives the vibrational assignment, the wavelength in air, and the wave-number *in vacuo* of the band heads of the isotopic molecules As¹²¹Sb and As¹²³Sb. The higher relative intensities of the former molecule are consistent with the relative natural abundances of $57\cdot25\%$ and $42\cdot75\%$ for ¹²¹Sb and ¹²³Sb, respectively. From the observed band heads the calculated vibrational constants of the lower electronic state are the following:

	$\omega_{e}(\text{cm.}^{-1})$	$\omega_{e}X_{e}(\text{cm.}^{-1})$		
As ¹²¹ Sb	343.0	0.8		
As ¹²³ Sb	341.7	0.7		

The ω_e values satisfy the relation $\omega_e^i = \rho \omega_e$ where the value of ρ for the two isotope molecules is 0.99688. The

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Band heads of the AsSb spectrum

at all	$As^{121}Sb$	$As^{121}Sb$		As ¹²³ Sb		Tantona shift
0-0	A A (III all)			$\nabla \operatorname{CIII}$ (<i>in vacuo</i>	·) 1 -	isotope sint
00	3662.35	$27297 \cdot 1 R$ 8	3	$27297 \cdot 1 R$	8	0 b
1-1	3680.65	27160·4° R	L			-
0-1 3708·74 3708·94	3708.74	$26955 \cdot 7 R$ 10	0	$26956 \cdot 8 R$	6	-1.1
	3708.94	26954·2 O 10	0			
0-2 3756.08	3756.08	$26615 \cdot 9 \tilde{R}$ 9	9	26617.9 R	6	-2.0
				$26617 \cdot 1 O$	6	
0-3 3804	3804.47	$26277 \cdot 4 R$ (6	$26280.9~\widetilde{R}$	4	-3.5
	3804.60	26276·5 Q 6	6	26279.5 Q	4	-3.0
0-4	3853.33	25944.2° R	1	$25946 \cdot 9$ ° \widetilde{Q}	1	

R. R head.

Q. Q head.

^a Relative intensities. ^b Isotope shift not detected.

^c These bands are extremely weak and therefore uncertain.

small observed isotope shifts are as expected in view of the large value of ρ and therefore further supports the vibrational assignment. In addition to these values for the lower state an estimate of 204.7 cm.⁻¹ for $\Delta G'_{\frac{1}{2}}$ may be made from the weak 1-1 band.

Although the bands of the v' = 0 progression are strong the expected 1-0 band at 3636 Å was not observed. The absence of this band and the sudden drop in intensity of the 1-1 band relative to that of 0-1 suggests that predissociation occurs at v' = 1.

Preliminary rotational analysis suggests that the AsSb bands as well as those of AsP and PSb are due to a $\Pi - \Sigma$ transition, analogous to the bands of AsN. At present, the

- ¹ K. K. Yee and W. E. Jones, *Chem. Comm.*, 1969, in the press.
 ² K. K. Yee and W. E. Jones, to be published.
 ³ J. W. T. Spinks, *Z. Phys.*, 1934, 88, 511.

rotational structure of AsSb is insufficiently resolved for detailed analysis but attempts will be made to re-photograph the spectra at higher resolution.

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