

## Preparation of $\sigma$ -Complexes Analogous to Intermediates in Nucleophilic Aromatic Substitution Reactions

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**Summary** Stable  $\sigma$ -complexes are formed by reaction of polynitro-compounds with haloforms or tetrachloromethane in the presence of base.

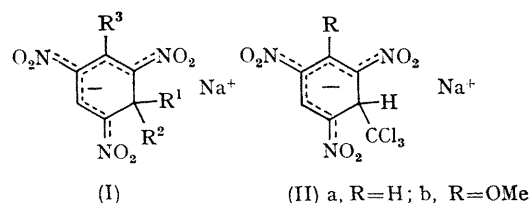
NUCLEOPHILIC aromatic substitution reactions proceeding by way of intermediate Meisenheimer complexes (I) are well known.<sup>1</sup> Stable  $\sigma$ -complexes are now reported (II) which are formed in reactions of polynitro-compounds with haloforms or  $\text{CCl}_4$  in the presence of alkoxide ion or amines.

Thus, a 1.0M-solution of 1,3,5-trinitrobenzene in  $\text{Me}_2\text{SO}$  (2.5 ml.) and 5M-NaOMe in MeOH (0.5 ml) were kept for 10–20 min., and chloroform (3.6 ml.) was added. The mixture was kept at 40° for 5 hr., then diluted with water and ice, cooled, and filtered. The product (IIa) (red crystals) was unstable. The n.m.r. spectrum (in  $\text{Me}_2\text{SO}$ ) showed signals at  $\delta - 8.61$  (doublet,  $J$  1.5 Hz.),  $-6.35$  (triplet,  $J$  1.5 Hz.), of relative intensity 2:1.

When sodium methoxide was added to a solution of 1,3,5-trinitrobenzene in  $\text{Me}_2\text{SO}$ , the n.m.r. spectrum indicated the formation of a complex (I;  $\text{R}^1 = \text{H}$ ,  $\text{R}^2 = \text{OMe}$ ,  $\text{R}^3 = \text{H}$ ). When chloroform was added to the solution,

the signals corresponding to (I;  $\text{R}^1 = \text{H}$ ,  $\text{R}^2 = \text{OMe}$ ,  $\text{R}^3 = \text{H}$ ) were replaced by those of (IIa).

Treatment of (IIa) with water or dilute acid (or merely leaving it exposed to the air) regenerated 1,3,5-trinitrobenzene. Addition of NaOMe to a solution of (IIa) in  $\text{Me}_2\text{SO}$  resulted in the formation of (I;  $\text{R}^1 = \text{H}$ ,  $\text{R}^2 = \text{OMe}$ ,  $\text{R}^3 = \text{H}$ ), *i.e.*, complexes (I;  $\text{R}^1 = \text{H}$ ,  $\text{R}^2 = \text{OMe}$ ,  $\text{R}^3 = \text{H}$ ) and (IIa) are in equilibrium.



2,4,6-Trinitroanisole, or the corresponding complex (I;  $\text{R}^1 = \text{R}^2 = \text{OMe}$ ;  $\text{R}^3 = \text{H}$ ), under similar conditions, gave complex (IIb), which showed n.m.r. signals at  $\delta - 8.64$  and  $-6.64$  (doublets,  $J$  1.7 Hz.), relative intensity 1:1.

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\* R. Foster and C. A. Fyfe, *Rev. Pure and Appl. Chem. (Australia)*, 1966, 16, 61.