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Cleavage of Si-Mn Bonds by Phosphites

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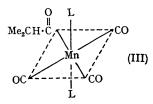
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Summary Trialkyl and triaryl phosphites cleave the siliconmanganese bond in triphenylsilylpentacarbonylmanganese (0).

REACTIONS of Ph₃SnMn(CO)₅^{1,2} and Ph₃GeMn(CO)₅³ with phosphines and phosphites (phos) in the absence of solvent at elevated temperatures have been shown to give Ph₃XMn-(CO)₄(phos) (X = Ge or Sn) products. We have found similar reactions to occur in solution. The reaction with PPh₃ of Ph₃SiMn(CO)₅⁴ in decalin at 130° also yields a product of this type, trans-Ph₃SiMn(CO)₄PPh₃ [v(CO) 2051 (vw), 1993 (w), and 1952 (s) cm.⁻¹, CHCl₃ solvent],[†] characterized by elemental analysis. In contrast, Ph₃SiMn(CO)₅ reacts with phosphites, P(OR)₃ (R = Ph, Me, or Pr¹) according to equation (1).

$$\begin{array}{c} \operatorname{Ph}_{3}\operatorname{SiMn}(\operatorname{CO})_{5} + \operatorname{P}(\operatorname{OR})_{3} \rightarrow \\ \operatorname{RCOMn}(\operatorname{CO})_{3}[\operatorname{P}(\operatorname{OR})_{3}]_{2} + \operatorname{OP}(\operatorname{OSiPh}_{3})_{3} & (1) \\ (\mathrm{I}) & (\mathrm{II}) \end{array}$$

Thus, reaction of 5.44 g. of Ph₃SiMn(CO)₅ with 14 g. of P(OPr¹)₃ (L, a ten-fold excess)in 100 ml. n-heptane at 80° under a CO atmosphere (to inhibit decarbonylation) gave, after 10 hr. a white precipitate of (II), which was collected by filtration, and a straw-coloured filtrate. Compound (II) was found to be identical (mixed m.p., i.r. spectrum) with an authentic sample of OP(OSiPh₃)₃ prepared from Ph₃SiOH and H₃PO₄ in toluene.⁵ The filtrate was chromatographed under CO on silica (n-heptane-diethyl ether eluant), and after evaporation of the solvent the residue was recrystallized from n-heptane to give colourless crystals of (I; $R = Pr^{i}$) which slowly became yellow even under vacuum. This complex was characterized by elemental analysis, i.r. spectrum $\int v(CO) 2031$ (w), 1948 (vs), 1906 (s) cm.⁻¹; acyl v(CO) 1600 cm.⁻¹, n-heptane solvent],[†] mass spectrum (peaks at m/e 598, assigned to $[Pr^{i}Mn(CO)_{3}L_{2}]^{+}$ and 583, assigned to $[COMn(CO)_{3}L_{2}]^{+}$), and ¹H n.m.r. spectrum. The latter (L methyl, τ 8.72, d, J 7 Hz.; isobutyl methyl, τ 9.18, d, J 7 Hz.; relative intensities 6:1, CCl₄ solvent, Me₄Si standard) demonstrates the magnetic equivalence of the two L groups, and on the basis of the data (I) is assigned the structure (III) below.



The relatively mild conditions under which this reaction occurs (ca. 80° compared with >130° for the reaction of Ph₃SiMn(CO)₅ with PPh₃) suggest RMn(CO)₅ to be formed early in the reaction, and that this species, in turn, undergoes carbonyl insertion. Carbonyl insertion reactions of RMn-(CO)₅ to give RCOMn(CO)₃(phos)₂ products have been found to proceed at 50—70°.⁶ Such alkylmanganese species might possibly arise through a Michaelis–Arbuzov rearrangement involving attack by the phosphite on the triphenylsilicon moiety, equation (2)

$$Ph_{3}SiMn(CO)_{5} + P(OR)_{3} \rightarrow \begin{bmatrix} OR \\ | \\ Ph_{3}Si-P^{+}-OR + Mn(CO)_{5}^{-} \\ | \\ OR \end{bmatrix}$$

$$O$$

$$Ph_{3}Si-P-R + RMn(CO)_{5} \qquad (2)$$

$$O$$

$$(IV)$$

The reaction of Et_3SiBr with triethyl phosphite to give a product analogous to (IV) has been reported.⁷ Steps which might result in the conversion of (IV) into (II) are not clear,

† Relative band intensities: s = strong; m = medium; w = weak; v = very.

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but the observed ligand dependency of the reaction rate and the formation of a phosphate, and not a phosphite product are consistent with an Arbuzov mechanism.

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