Enzymic Formation of a Tricyclic Sesterterpene Alcohol from Mevalonic Acid and all-trans-Geranylfarnesyl Pyrophosphate

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Summary Mevalonic acid lactone and the chemically synthesized all-trans-geranylfarnesyl pyrophosphate are converted into a tricyclic sesterterpene alcohol, ophiobolin F, by incubation with $100,000 \times g$ supernatant fraction from cell-free system of Chochliobolus heterostrophus.

A PHYTOTOXIC substance, ophiobolin A, isolated from the plant pathogenic fungi, was the first known sesterterpene.¹ Later, several related compounds were found to occur in fungal metabolites² or in insect secretions.³ Biosynthetic studies of the ophiobolins using the intact cell system were reported and experimental evidence for the genesis of these compounds was presented.⁴,⁵ We have found that a crude enzyme system obtained from the cell-free homogenate of the C. heterostrophus catalysed the formation of the tricyclic sesterterpene alcohol (I) from MVA and all-transgeranylfarnesyl pyrophosphate.

Mycelia of C. heterostrophus were ground in a mortar with sand and 0·1M-phosphate buffer (pH 7·35) containing MgCl₂, and the homogenate was successively centrifuged, first at $1000 \times g$, for 10 min., then at $100,000 \times g$ for 1 hr. The $100,000 \times g$ supernatant was incubated with [2-14C]-MVA in the presence of ATP, for 3 hr. at 37°. After the addition of KOH, the incubation mixtures were extracted with ether, and the extracts were chromatographed on silica-gel plates. 56% of the radioactivity of the added substrate appeared in a band having the same $R_{\rm F}$ value as ophicbolin F (I)6 which had been isolated from the same fungus as a very minute component, and its structure was tentatively assigned as (I). The conclusive identification of the enzymic reaction product and ophiobolin F is based on autoradiography on t.l.c., gas-liquid radiochromatography, and g.l.c. mass spectral analysis.

The structure and stereochemistry of ophiobolin F (I) were confirmed by chemical correlation with ophiobolin C (II),

whose structure and absolute configuration were already established.⁷ Reduction of ophiobolin C (II) with lithium

aluminium hydride afforded a triol, m.p. $102-104^{\circ}$, which was hydrogenated with palladium-charcoal in ethanol to give the mono-alcohol derivative (III), $C_{25}H_{44}O$, M^+ 360, intense peaks at m/e 345, 342, 327, 257, 247, 229, and 207.

The n.m.r. spectrum of (III) exhibited signals at δ 0.79, 0.82, 0.89 (9H, three secondary methyls), 0.97 (s, C-11 Me), 1.13 (s, C-3 Me) and at 1.52 (s, C-7 Me). The identical mono-alcohol (III) was also derived from ophiobolin F by catalytic hydrogenation.

All-trans-geranylfarnesyl pyrophosphate was synthesized by a known procedure, as follows. Condensation of the all-trans-geranylgeranylacetone (IV)8 (prepared from geranyl-linolool with triethyl phosphonoacetate in tetrahydrofuran in the presence of sodium hydride) afforded ethyl geranylfarnesoate (V) as a cis, trans-mixture in a ratio of ca. 1:4. Chromatographic separation of the mixture yielded all-trans-ethyl geranylfarnesoate (Va), $[(M^+ 400)]$, n.m.r. signals of methyl groups appeared at δ 1.58 (12H), 1.67 (3H), 2.12 (3H)] and a terminal cis-isomer (Vb), $\lceil (M^+) \rceil$ 400), n.m.r. signals at δ 1.58 (12H), 1.66 (3H), 1.88 (3H)] both of which were reduced by lithium aluminium hydride to give all-trans- and a terminal cis-geranylfarnesol (VIa, VIb), respectively. All-trans-isomer (VIa) showed methyl proton signals at δ 1.58 and 1.66 which differ from the reported value for natural geranylfarnesol recently isolated from insect wax.9 The natural compound was assumed to be a terminal cis-isomer from a comparison of the its n.m.r. spectrum with that of the synthetic compound which exhibited signals at δ 1.58, 1.66, and 1.72. Tritiumlabelled geranylfarnesol obtained by lithium aluminium tritiide reduction of (Va) was converted into the pyrophosphate ester (VII) by Cramer's procedure as modified by Kandutsch *et al.*¹⁰ The synthetic [1,1-3H₂]-all-transgeranylfarnesyl pyrophosphate (VI) was incubated with $100,000 \times g$ supernatant fraction under the same conditions described above to give ophiobolin F (I) (ca. 20%). Negligible incorporation of radioactivity into (I) was observed when [1,1-3H2]-cis-geranylfarnesyl pyrophosphate and [1,1-3H2]-all-trans-geranylfarnesol were used as substrates. The data presented provide the first experimental evidence for the intermediary role of geranylfarnesyl pyrophosphate in sesterterpene biosynthesis. Investigations on co-factor requirements, substrate specificity, and the separation of cyclizing enzyme activity from the others are in progress.

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