

Photoaddition of Carboxylate Esters to Diphenylacetylene

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Summary Irradiation of some carboxylate esters and tolan (diphenylacetylene) gives the corresponding β -alkoxy- α -benzoylstyrene derivatives.

WHILE the photoaddition of aldehydes or ketones to acetylenes has been considerably explored,¹ no studies on

findings concerning the photoadditions of some carboxylate esters with tolan.

A solution of carboxylate ester (0.03 mole) and tolan (0.01 mole) in benzene (250 ml.) was irradiated through Pyrex with a 250 w high-pressure mercury arc for 25 hr. In the present study, dimethyl terephthalate, methyl

TABLE

Photoaddition of carboxylate esters to tolan

Carboxylate esters	Products (yield %) ^a	M.p.(°)
Dimethyl terephthalate	<i>trans</i> -(32)	124—126
	<i>cis</i> - (32)	91—93
Methyl <i>p</i> -cyanobenzoate	(86)	150—152
Diethyl oxalate	(25)	48—50

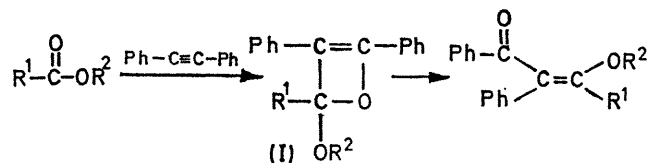
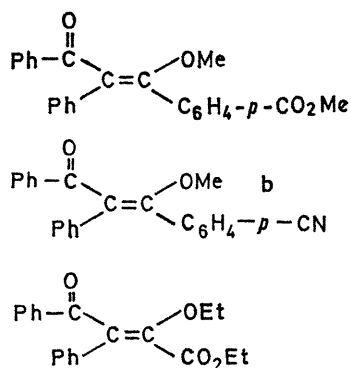
^a Figures in parentheses denote yield based on unrecovered carboxylate ester.

^b The *cis*-isomer rapidly changed to *trans*- even during recrystallization. Therefore, the *cis*-isomer was not isolated in pure form.

the photoaddition of carboxylate esters to acetylenes have previously been reported. In continuation of our study of the photochemistry of carboxylate esters having electron-withdrawing groups,² we report here our preliminary

p-cyanobenzoate and diethyl oxalate were selected as the carboxylate esters. The photoproducts were characterized by elemental analysis, i.r. and n.m.r. spectra, and molecular-weight measurements. The results are presented in the Table.

For these photoadditions, we suggest the following mechanism which is similar to those postulated for the photoaddition of *p*-benzoquinone to tolan;¹ *i.e.*, the reaction probably proceeds through the attack of the carbonyl n,π^* triplet state on the tolan triple bond to produce an unstable oxetene intermediate (I), followed by ring opening of (I) to give the observed photoproduct.



(Received, September 22nd, 1969; Com. 1437.)

¹ H. E. Zimmerman and L. Craft, *Tetrahedron Letters*, 1964, 2131; D. Bryce-Smith, G. I. Fray, and A. Gilbert, *ibid.*, p. 2137.

² Y. Odaira, T. Shimodaira, and S. Tsutsumi, *Chem. Comm.*, 1968, 757; T. Tominaga, Y. Odaira, and S. Tsutsumi, *Bull. Chem. Soc. Japan*, 1967, 40, 2451; Y. Shigemitsu, H. Nakai, and Y. Odaira, *Tetrahedron*, 1969, 25, 3039.