

The Formation of Cinnamylphenols by means of an *ortho-ortho*-Rearrangement

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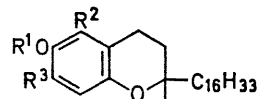
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Summary Two minor products of Claisen rearrangement of 3-phenylprop-2-enyl tosyl ether are apparently formed by an *ortho-ortho*-rearrangement forbidden by the Woodward-Hoffmann rules.

THE current interest¹ in aromatic sigmatropic rearrangements prompts us to report our recent findings on the Claisen rearrangement of 3-phenylprop-2-enyl tosyl ether (I). The ether was heated (240°, 3 min.) in the absence of solvent to give a mixture of three products which consisted of a major product, tocol (II), as a result of cleavage,² and two minor rearrangement products (III, IV), which were obtained as pale yellow oils. The products were obtained in the proportion 20:4:3.

The u.v. spectrum of (III) [λ_{\max} in EtOH 252 (ϵ 16,900) 286 (4900), 295 nm (5900)] was characteristic³ of a *trans*-cinnamylphenol and its structure as 5-(3'-phenylprop-2'-enyl)tocol was assigned by inspection of its n.m.r. spectrum (see Table), which showed resonances characteristic⁴ of the $\cdot\text{CH}:\text{CH}:\text{CH}_2$ group present in cinnamylphenols and a singlet for the two aromatic protons. Acetylation of (III) yielded a monoacetate (V) [λ_{\max} in EtOH 250 (ϵ 22,100), 285 (6100), and 293 nm (4350); ν_{\max} 1760 cm^{-1}] the n m r. spectrum of which indicated a characteristic AB system for the *ortho*-substituted aromatic protons. The structure

of the second rearrangement product as 7-(3'-phenylprop-2'-enyl)tocol (IV) was assigned from its u.v. spectrum [λ_{\max} in EtOH 251 (ϵ 18,300), 287 (5950), and 295 nm. (6700)] and n.m.r. spectrum (see Table).



	R ¹	R ²	R ³
(I)	CH ₂ :CH:CHPh	H	H
(II)	H	H	H
(III)	H	CH ₂ :CH:CHPh	H
(IV)	H	H	CH ₂ :CH:CHPh
(V)	Ac	CH ₂ :CH:CHPh	H
(VI)	CH ₂ :CH:CHMe	H	H
(VII)	H	CHMe:CH:CH ₂	H
(VIII)	H	H	CHMe:CH:CH ₂
(IX)	H	CH ₂ :CH:CHMe	H

The formation of the cinnamylphenols (III, IV) is one of the few examples of an apparent *ortho-ortho*-Claisen rearrangement,^{1,2,5} a type that is forbidden by the Woodward-Hoffmann selection rules.⁶ The Claisen rearrangement of the but-2-enyl ethers of tocol and other tocopherols

60 MHz N.m.r. data for tocol and its derivatives (in CDCl₃)

Compound	Aromatic protons			·CH : CH·CH ₂ · residue	Acetoxy- or hydroxy-group	Phenyl group	Unassigned protons
	5-H	7-H	8-H				
(I)	3.35	3.28	3.28	3.10—3.90(2)m 5.38(2)d(J 5)	—	2.66(5)s	7.10—9.30(40)m
(II)	3.41	3.36	3.36	—	5.00(1)s	—	7.00—9.30(40)m
(III)	—	3.38	3.38	3.50—3.80(2)m 6.47(2)m	5.50(1)s	2.75(5)s	7.00—9.30(40)m
(IV)	3.47	—	3.37	3.30—3.95(2)m 6.52(2)d(J 5)	5.60(1)s	2.72(5)s	7.00—9.30(40)m
(V)	—	3.15	3.31	3.40—4.00(2)m 6.58(2)d(J 5)	7.76(3)s	2.75(5)s	7.00—9.30(40)m
(VI)	3.35	3.29	3.29	3.70—4.55(2)m 5.60(2)d(J 5)	—	—	7.10—9.30(43)m
(IX)	—	3.41	3.41	4.00—4.80(2)m 6.70(2)d(J 5)	5.50(1)s	—	7.00—9.30(43)m
(VII)	—	3.38	3.38	>CH·CH : CH ₂ 3.30—4.90(3)m 6.00—6.40(1)m	4.95(1)s	—	7.10—9.30(43)m
(VIII)	3.47	—	3.39	3.50—5.05(3)m 6.10—6.60(1)m	5.50(1)s	—	7.10—9.30(43)m

Chemical shifts are given on the τ scale and proton integrals in parentheses. Coupling constants, J , are in Hz. s = singlet; d = doublet; m = multiplet.

was previously shown^{2b} to give normal and abnormal products, and kinetic studies³ showed that the product of abnormal rearrangement was formed intramolecularly, probably by an *ortho-ortho*-shift. The rearrangement of but-2-enyl tocol ether (VI) to give two normal products (VII, VIII) and one abnormal product (IX) has been re-examined and the structures of the products confirmed by n.m.r. spectroscopy (Table). The rearrangement of the

3-phenylprop-2-enyl ether of tocol apparently follows a similar course, being unusual in that none of the normal product was isolated.

The anomalous *ortho-ortho*-shift that occurs during the pyrolysis of tocopheryl ethers does not occur to any great extent in phenyl or *para*-alkoxyphenyl ethers^{2b} and appears to be characteristic of the 6-chromanil structure.

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⁵ F. J. Dinan and H. Tieckelmann, *J. Org. Chem.*, 1964, **29**, 892.

⁶ R. B. Woodward and R. Hoffmann, *J. Amer. Chem. Soc.*, 1965, **87**, 2511; R. Hoffmann and R. B. Woodward, *ibid.*, p. 4389.