

## X-Ray Determination of the Structure of Capaurimine

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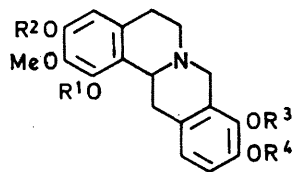
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**Summary** An X-ray analysis of capaurimine mono-*p*-bromobenzoate has confirmed that capaurimine is 5,6,13,13a-tetrahydro-1,10-dihydroxy-2,3,9-trimethoxy-8*H*-dibenzo[*a,g*]quinolizine.

CAPAUIMINE, one of the protoberberine alkaloids isolated from *Corydalis* species,<sup>1</sup> had been assigned structure (I) through chemical degradations by Manske,<sup>2</sup> but compound (I) synthesised by one of the present authors was not identical with natural capaurimine.<sup>3</sup> We have therefore reinvestigated capaurimine and found that it is 5,6,13,13a-tetrahydro-1,10-dihydroxy-2,3,9-trimethoxy-8*H*-dibenzo[*a,g*]quinolizine (II).<sup>4</sup> It was predicted that on account of a non-bonded interaction of the hydroxy-group at the 1-position with the hydrogens at the 13-position, a *cis*-quinolizidine conformation would be preferred.<sup>4</sup> We have now undertaken an X-ray analysis of capaurimine mono-*p*-bromobenzoate (III) in order to determine its three-dimensional molecular structure.

Treatment of capaurimine, isolated from *C. pallida* var. *tenuis* Yatabe by us,<sup>5</sup> with *p*-bromobenzyl chloride in

pyridine gave its mono-*p*-bromobenzoate (III), which was recrystallised from methanol as pale yellow plates. C<sub>27</sub>H<sub>26</sub>O<sub>6</sub>NBr, M.W. = 540.402, m.p. 177—178°, [*D*<sub>c</sub> = 1.483



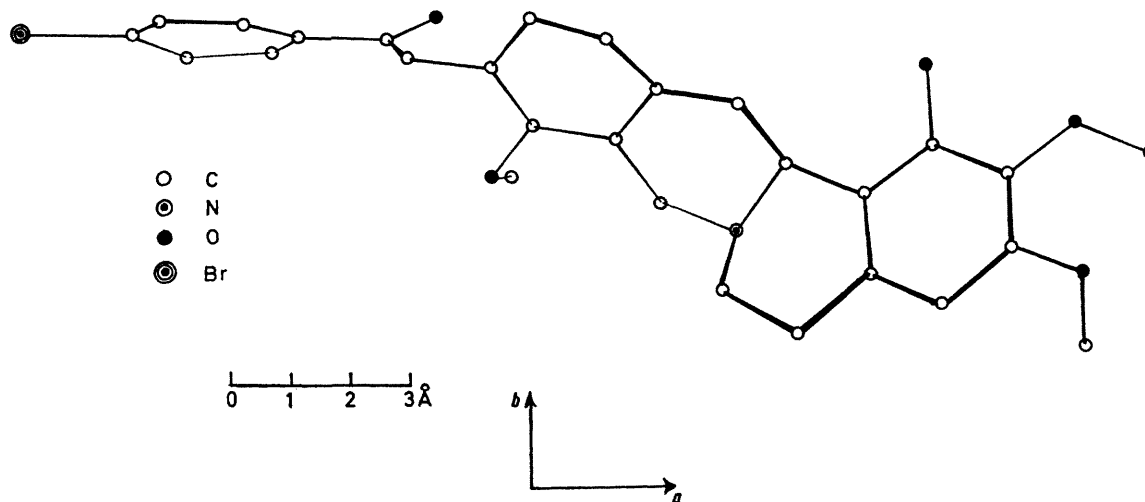
- (I) R<sup>1</sup> = R<sup>3</sup> = H, R<sup>2</sup> = R<sup>4</sup> = Me  
 (II) R<sup>1</sup> = R<sup>4</sup> = H, R<sup>2</sup> = R<sup>3</sup> = Me  
 (III) R<sup>1</sup> = H, R<sup>2</sup> = R<sup>3</sup> = Me, R<sup>4</sup> = CO·C<sub>6</sub>H<sub>4</sub>Br  
 (IV) R<sup>1</sup> = H, R<sup>2</sup> = R<sup>3</sup> = R<sup>4</sup> = Me  
 (V) R<sup>1</sup> = R<sup>4</sup> = Me, R<sup>2</sup> = R<sup>3</sup> = H

g cm<sup>-3</sup>, *D*<sub>m</sub> = 1.499 g cm<sup>-3</sup>, monoclinic, *a* = 13.182, *b* = 7.705, *c* = 13.369 Å, β = 117.0°, space group *P*2<sub>1</sub>, *Z* = 2.

The intensity data were collected by multiple-film equi-inclination Weissenberg photographs using Cu-*K*<sub>α</sub> radiation.

Relative intensities were estimated visually. The structure was solved by the heavy-atom method. In spite of a disturbance by a false mirror plane at  $y = 0$ , it was possible

The molecular structure of capaurimine mono-*p*-bromobenzoate as projected along the *c*-axis is shown in the Figure. As shown in the Figure, three methoxyl groups exist at the



FIGURE

to pick out the atoms in the *p*-bromobenzoyl group. Two successive three dimensional Fourier syntheses showed clearly the whole molecule.

The structure was refined by least-squares method. After three cycles of refinements, the discrepancy factor was reduced to 0.12. Further refinements are in progress.

2, 3, and 9-positions and *p*-bromobenzoyl group exists at the 10-position. Furthermore, the B/C ring junction is *cis*-fused as those of the hydrobromide of capaurine (IV)<sup>6</sup> and the position isomer of capaurimine (V).<sup>7</sup> On the basis of this result, the correctness of the structure of capaurimine (II) suggested by us<sup>4</sup> has been confirmed.

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<sup>1</sup> R. H. F. Manske, *Canad. J. Res.*, 1940, **18**, B, 80; R. H. F. Manske, *ibid.*, 1942, **18**, B, 49.

<sup>2</sup> R. H. F. Manske, *J. Amer. Chem. Soc.*, 1947, **69**, 1800.

<sup>3</sup> T. Kametani, K. Fukumoto, H. Yagi, H. Iida, and T. Kikuchi, *J. Chem. Soc. (C)*, 1968, 1178.

<sup>4</sup> T. Kametani, M. Ihara, and T. Honda, *J. Chem. Soc. (C)*, in the press.

<sup>5</sup> T. Kametani, M. Ihara, and T. Honda, *Chem. Comm.*, 1969, 1301; T. Kametani, M. Ihara, and T. Honda, *J. Chem. Soc. (C)*, 1970, 1060.

<sup>6</sup> T. Kametani, M. Ihara, K. Fukumoto, H. Yagi, H. Shimanouchi, and Y. Sasada, *Tetrahedron Letters*, 1968, 4251; H. Shimanouchi, Y. Sasada, M. Ihara, and T. Kametani, *Acta Cryst.*, 1969, B **25**, 1310.

<sup>7</sup> T. Kametani, K. Wakisaka, T. Kikuchi, M. Ihara, H. Shimanouchi, and Y. Sasada, *Tetrahedron Letters*, 1969, 627; H. Shimanouchi, Y. Sasada, K. Wakisaka, T. Kametani, and M. Ihara, *Acta Cryst.*, 1970, B **26**, 607.