

## The $\text{CS}_2^-$ Group as a Conformational Probe for Cyclic Amines: Nuclear Magnetic Resonance Study

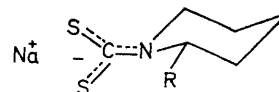
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**Summary** The n.m.r. spectra of the *N*-carbodithioic acid sodium salts of piperidines having equatorial  $\alpha$ -alkyl substituents indicate that conformational inversion of the piperidine ring occurs upon formation of these derivatives.

THE exceptionally strong deshielding of the equatorial 2,6-protons by the  $\text{CS}_2^-$  group in the *N*-carbodithioic acid salts of piperidines provides an excellent means of distinguishing between the equatorial and the axial protons.<sup>1</sup> However, we have found that the observation of this strong deshielding in the salts does not necessarily indicate that the deshielded proton has an equatorial orientation in the free base. A conformational inversion of the piperidine ring was found to occur upon formation of the *N*-carbodithioic acid salts of piperidines which contained  $\alpha$ -alkyl equatorial substituents, thus reversing the axial-equatorial orientation of the ring protons.

These peaks are due to the two equatorial protons, thus showing that the alkyl substituent is in the axial position. In the case of the *cis*-2,6-dimethylpiperidine the two  $\alpha$ -protons are found at  $\tau$  4.2 indicating that the two protons are equatorial and the methyl groups have been forced into a 1,3-diaxial interaction with one another. The assignment of the axial or equatorial orientation to the 2,6-protons of the salts is verified by their splitting patterns. The axial protons appear as triplets (splitting of about 13 Hz; geminal and vicinal diaxial coupling) further split by *ca.* 3 Hz (vicinal axial-equatorial coupling). The equatorial protons show the expected splitting patterns with no vicinal coupling greater than 4 Hz.



FIGURE

Chemical shifts ( $\tau$  values) of  $\alpha$ -protons (60 MHz)

Compound	Amine <sup>a</sup>		Salt <sup>b</sup>	
	Axial	Equatorial	Axial	Equatorial
4-Methylpiperidine .. ..	7.40	6.91	6.82	4.63
2-Methylpiperidine .. ..	7.4 $\pm$ 0.1	6.90	6.85	4.07, 4.52
2-Ethylpiperidine .. ..	7.4 $\pm$ 0.1	6.90	6.83	4.1, 4.47
2-Propylpiperidine .. ..	7.4 $\pm$ 0.1	6.90	6.83	4.1, 4.4
<i>cis</i> -2,6-Dimethylpiperidine ..	7.44	—	—	4.20

<sup>a</sup> In  $\text{CDCl}_3$ ,  $\text{Me}_4\text{Si}$  as internal standard.

<sup>b</sup> In  $\text{D}_2\text{O}$ , sodium 2,2-dimethyl-2-silapentane-5-sulphonate as internal standard.

The chemical shifts of the 2,6-protons in the free base and in the *N*-carbodithioic acid sodium salts of some substituted piperidines are shown in the Table. The signals due to the equatorial protons in the salts are found near  $\tau$  4, whereas those due to the axial protons are found near  $\tau$  7. It may be seen that there are two signals near  $\tau$  4 in the n.m.r. spectra of each piperidine salt which has one  $\alpha$ -substituent.

This large steric interaction between the  $\text{CS}_2^-$  group and the equatorial alkyl substituent must be due to the partial double bond character of the C-N bond which places the  $\text{CS}_2^-$  group in an orientation which causes it nearly to eclipse the equatorial substituent (see Figure).

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<sup>1</sup> H. Booth and A. H. Bostock, *Chem. Comm.*, 1967, 637.