

Chemical Communications

(The Journal of the The Chemical Society, Section D)

NUMBER 23/1970

9 DECEMBER

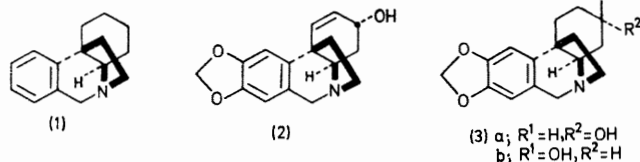
General Methods of Alkaloid Synthesis. A New Approach to the Synthesis of the 5,10b-Ethanophenanthridine *Amaryllidaceae* Alkaloids. The Total Synthesis of (\pm)-3-*epi*-Elwesine

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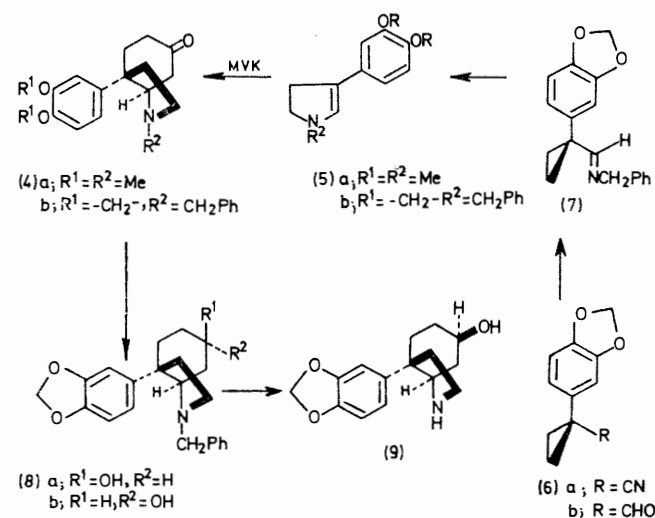
Summary Methyl vinyl ketone annelation of the Δ^2 -pyrroline (**5b**) prepared by acid-catalysed thermal rearrangement of the cyclopropyl imine (**7**) has been exploited in an efficient eight-step stereoselective total synthesis of racemic 3-*epi*-elwesine (**3b**).

MEMBERS of one major group of the *Amaryllidaceae* alkaloids¹ contain the 5,10b-ethanophenanthridine nucleus (**1**) as exemplified by crinine (**2**). We report a synthesis of 3-*epi*-elwesine (**3a**), a minor alkaloid of *Galanthus elwesii* Hook. f.,² which might be employed with only minor modification in the synthesis of a number of the crinine family of alkaloids.



The method of approach we envisaged was a logical extension of two fundamental and increasingly important general principles of alkaloid synthesis which we and others have been developing. The first of these exploits the acid-catalysed, thermally induced rearrangement of cyclopropyl imines as a useful general approach to pyrrolines and has provided simple efficient syntheses of the pyridine alkaloid myosmine and apoferrerosamine³ and was a key step^{4a,b} in a five-stage synthesis of the *Aizoaceae* alkaloid mesembrine

(**4a**).⁴ Also of importance in the latter synthesis was the application of the methyl vinyl ketone annelation to an endocyclic enamine (cf. **5a** to **4a**); a reaction which has also been exploited in the synthesis of the *Erythrina*⁵ and hasubanan⁶ skeletons as well as a useful *Aspidosperma* alkaloid precursor. The somewhat deceptive similar structural features of mesembrine (**4a**) and the crinine-type alkaloids such as elwesine (**3a**) had, from the very beginning of our investigation,⁷ not escaped our attention.



Our synthesis begins with piperonyl cyanide, whose conversion into the cyclopropane carbonitrile (**6a**)† (m.p.

† Each intermediate reported in this communication had been subjected to i.r., and n.m.r. analysis. Supporting data were obtained from low-resolution mass spectral data and/or combustion analysis.

74.5—75.5°) was achieved in 65—75% yield by using $(\text{CH}_2\text{Br})_2$ and LiNH_2 in glyme at room temperature. This result is in contrast to the employment of NaNH_2 /glyme which gave only low yields of the cyclopropane and is in consonance with our previous studies.^{4a} Di-isobutyl-aluminium hydride reduction of (6a) yielded aldehyde (6b) (m.p. 62.5—63.5°, 75—85%) whose transformation into aldimine (7) (m.p. 67—67.5°) was accomplished by simply stirring a benzene solution with anhydrous CaCl_2 for 2—3 days (72—90%). Thermal rearrangement of this cyclopropyl imine proceeded smoothly in 72—80% yield by employing NH_4Cl as the acidic catalyst. The methyl vinyl ketone annelation of this endocyclic enamine (m.p. 62.5—63°) gave only complex unstable mixtures containing little if any of the desired product. However, 55—65% yields of the pure *cis*-octahydroindole (4b) (m.p. 98.5—101°)

could be secured by employing acid catalysis.^{6c} NaBH_4 reduction of (4b) provided a 3:1 mixture of two epimeric alcohols. Debenzylation⁹ of the major product (8a, *vide infra*, m.p. 229—231°) yielded (9) (m.p. 179—180°, 100%) whose HCHO-induced Pictet-Spengler¹⁰ cyclization completed the synthesis of (\pm)-3-*epi*-elwesine (3b, m.p. 187—188.5°, 65%). The structural and stereochemical assignments were confirmed by comparison with an authentic optically active sample[†] and by oxidation to the known racemic ketone m.p. 171.5—174.5 (lit.¹¹ 171—173°).

We thank the Robert A. Welch and National Science Foundations for financial support. The n.m.r. and mass spectrometers were purchased with funds provided by the National Science Foundation. A Sloan Fellowship to R. V. S. is gratefully acknowledged.

(Received, September 7th, 1970; Com. 1517.)

† Kindly provided by Professor W. C. Wildman.

¹ W. C. Wildman in "The Alkaloids," ed. R. H. F. Manske, Academic Press, London and New York, 1968, vol. 11, p. 308.

² H.-G. Boit and W. Döpke, *Naturwiss.*, 1961, **48**, 406.

³ R. V. Stevens, M. C. Ellis, and M. P. Wentland, *J. Amer. Chem. Soc.*, 1968, **90**, 5576.

⁴ (a) R. V. Stevens and M. P. Wentland, *J. Amer. Chem. Soc.*, 1968, **90**, 5580; (b) S. L. Keely, jun., and F. C. Tahk, *ibid.*, p. 5584; (c) T. J. Curphy and H. L. Kim, *Tetrahedron Letters*, 1968, 1441.

⁵ R. V. Stevens and M. P. Wentland, *Chem. Comm.*, 1968, 1104.

⁶ D. A. Evans, *Tetrahedron Letters*, 1969, 1573.

⁷ R. V. Stevens, R. K. Mehra, and R. L. Zimmerman, *Chem. Comm.*, 1969, 877.

⁸ R. V. Stevens and M. C. Ellis, *Chem. Comm.*, 1967, 5185.

⁹ According to the procedure of G. Büchi, D. Coffen, K. Korsis, P. Sonnet, and F. Ziegler, *J. Amer. Chem. Soc.*, 1966, **88**, 3099.

¹⁰ H. Whitlock and G. Smith, *J. Amer. Chem. Soc.*, 1967, **89**, 3600.

¹¹ I. Irie, S. Uyeo, A. Yoshitake and A. Ito, *J. Chem. Soc. (C)*, 1968, 1802.