Structural Revision and Synthesis of a Novel Fungal Diaryl Ether

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Summary The structure 2-hydroxy-6-methoxy-4-methylphenyl 3-hydroxy-5-methylphenyl ether (1) proposed for the fungal metabolite LL-V125 α has been revised to 2-hydroxy-4-methoxy-6-methylphenyl 3-hydroxy-5methylphenyl ether (2); this has been confirmed by synthesis.

THE isolation of a diaryl ether, LL-V125 α , from a fungus of the order Sphaeropsidales has recently been reported.¹ The erroneous structural assignment (1) for this metabolite followed from an incorrect interpretation of the ¹H n.m.r. spectra of this compound and its derivatives. We now report an amended structure (2) for LL-V125 α , and a synthesis of this metabolite.

We have interpreted the data reported for this natural product in terms of structure (2), and hence structures (3), (4), and (5) follow for the di-O-methyl ether, the demethylation product, and the diacetate, respectively. Tri-Omethylalectol (3), a depsidone degradation product, had previously been synthesised by Asahina and Fuzikawa² in 0.2% yield by Ullmann condensation between mono-Omethylorcinol³ (6) and di-O-methyl-2-bromo-orcinol⁴ (7). Modification of this method gave (3), m.p. 94.5-96° (lit.,² 96°) (ca. 20%); mass spectrum, M^+ , m/e 288; τ (CDCl₃; all 60 MHz) 3.67 (m, 5H, ArH), 6.25 (s, 3H, OMe), 6.31 (s, 6H, OMe), and 7.77 and 7.87 (each s, 3H, Me). Demethylation of (3) gave alectol^{\dagger} (4); mass spectrum, M^+ , m/e 246; τ (CDCl₃) 3·18br (s, 2H, D₂O-exchangeable OH), 3·74 (m, 6H, $5 \times \text{Ar}H$ and $1 \times D_2\text{O}$ -exchangeable OH), 7.82 and 8.02 (each s, 3H, Me). The physical properties of both (3) and (4) are similar to those reported for the di-O-methyl ether, and the demethylation product, respectively, of LL-V125a.1

Ullmann condensation between mono-O-benzylorcinol[‡] (8) and (9)§ gave (10) (ca. 25%), which on hydrogenolysis afforded LL-V125a (2) (98%), m.p. 121-122° (lit., 121.5- 122.5° ; mass spectrum, M⁺, m/e 260; τ (CDCl₃) 3.72 (m, 5H, ArH), 4.42 and 4.62 (each br s, 1H, OH), 6.23 (s, 3H, OMe), and 7.78 and 7.94 (each s, 3H, Me) virtually identical with that reported for natural LL-V125 α . The i.r. spectra of the natural and synthetic products, kindly determined by Dr. E. L. Patterson, were identical. The diacetate (5) of synthetic LL-V125x had m.p. 72-73° (lit.,¹ 75-76°).

In contrast to these results, Ullmann condensation between (6) and di-O-methyl-4-bromo-orcinol¶ (11) gave (12), m.p. 141-142°; mass spectrum, M+, m/e 288; 7



(CCl₄) 3.65 (s, 2H, ArH), 3.80 (m, 1H, ArH), 3.93 (m, 2H, ArH), 6.30 (s, 6H, OMe), 6.33 (s, 3H, OMe), and 7.70 and 7.78 (each s, 3H, Me). Demethylation of (12) gave (13), m.p. 148-150°; mass spectrum, M+, m/e 246; 7 (CDCl₃) 3.69br (s, 5H, $2 \times \text{ArH}$ and $3 \times D_2\text{O}$ -exchangeable OH), 3.80 (m, 3H, ArH), and 7.87 and 7.97 (each s, 3H, CH.).

The above evidence indicates that LL-V125 α has structure (2).

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- † All new compounds gave satisfactory elemental analyses or high-resolution mass spectra.

- Obtained by partial hydrogenolysis of di-O-benzylorcinol. § Prepared by bromination of mono-O-methylorcinol followed by benzylation. ¶ Readily obtained by treatment of 4-lithio-di-O-methylorcinol with 1,2-dibromoethane.
- ¹ W. J. McGahren, W. W. Andres, and M. P. Kunstmann, J. Org. Chem., 1970, 35, 2433.
- ² Y. Asahina and F. Fuzikawa, Ber., 1934, 67, 163.

* Prepared in 98% yield by monodemethylation of di-O-methylorcinol by the general method of G. I. Feutrill and R. N. Mirrington, Tetrahedron Letters, 1970, 1327.

⁴ See D. D. Ridley, E. Ritchie, and W. C. Taylor, Austral. J. Chem., 1970, 23, 147; we found the direct bromination of di-O-methylorcinol to be more convenient.