Action of Acid on Photothebainehydroquinone: Stereochemistry of Opening of Cyclopropanes

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Summary The action of acid on photothebainehydroquinone leads to two products, both of which result from opening of the cyclopropane ring following 'end-on' attack of the proton.

THERE are two reported exceptions^{1,2} to the general rule that the main attack of a proton on a cyclopropane ring is by orthogonal approach to the bond being broken and leads to retention of configuration on the carbon atom accepting the proton. We report two further exceptions. Photothebainehydroquinone (1)³ when heated under reflux with concentrated hydrochloric acid for 3 h, gave the products (2a) and (3a).⁴ The structures of these compounds follow from their properties: both compounds have n.m.r. signals corresponding to 4 aromatic protons, 1 OMe, and 1 NMe group; in addition (2a) (C_5H_5N) has a singlet at 4.68 p.p.m. [C(5)-H], the only proton which is exchangeable in basic D,O] and a doublet (5 Hz) at 4.10 p.p.m. [C(7)-H, split by C(8)-H]; (3a) (100 MHz spectrum; C_5D_5N) has a singlet at 4.53 p.p.m. [C(5)-H], a doublet (8 Hz half masked by the NMe peak) at 2.36 p.p.m. [C(8)-H split by C(7)-H_A], a doublet (16 Hz) at 3.73 p.p.m. [C(7)-H_B split by C(7)-H_A], and a doublet of doublets (J = 16 and 8 Hz) at 2.97 p.p.m. $[C(7)-H_{A}]$; the C(5)-H and C(8)-H protons in (3a) are the only ones exchangeable in basic D_2O ; (2a) has v_{max} 1725 cm⁻¹, and (3a) 1745 cm⁻¹; and, in contrast to compounds with free phenolic groups at C(4) (e.g. flavothebaone),⁵ neither compound coupled with diazotised sulphanilic acid. Examination of models reveals that in (2a) the dihedral angle between C(7)-H and C(8)-H_A is 40° and in (3a) the the angle between C(8)-H and C(7)-H_A is 20°.

When (1) was heated with PCl_3 in D_2O compounds (2b) and (3b) were obtained. The n.m.r. spectrum of (2b) was the same as that of (2a) except that the doublet at 4.10 p.p.m. had become a singlet, and integration indicated one proton less in the region 1.65-2.30 p.p.m. The n.m.r. spectrum of (3b) was the same as that of (3a) except that the doublet at 3.73 had become a singlet and the signals at 2.36 and 2.97 p.p.m. had disappeared. Thus the attacks of deuterium at C(7) [to give (3b)] and C(8) [to give (2b)] took place with inversion of configuration and must have resulted from an 'end-on' approach of reagent. The reasons for the abnormal stereochemistries of these attacks are not clear; orthogonal attack at C(8), but not that at C(7), appears to be hindered sterically. Possibly, protonation on the ether bridge prevents, electrostatically, orthogonal approach for both openings.



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