

## Synthesis of Model Aryltetralin Lignans: Adducts of *o*-Methylbenzophenone with Unsymmetrical Dienophiles

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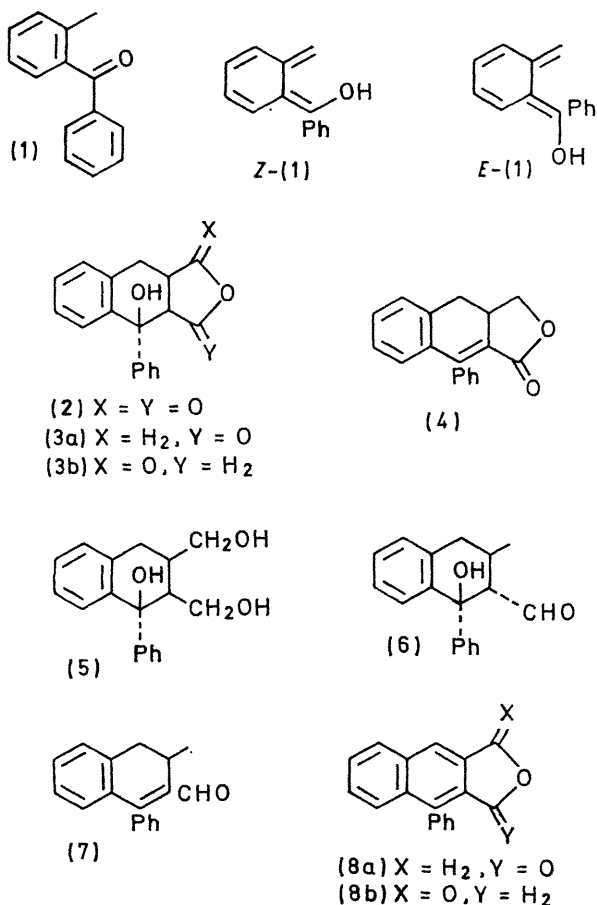
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**Summary** Photoirradiation of *o*-methylbenzophenone in the presence of unsymmetrical dienophiles yields aryltetralin adducts of defined configuration and structurally related to natural lignans.

SINCE the demonstration<sup>1</sup> that some *o*-alkylbenzophenones may undergo photoenolization, there has been considerable interest in the mechanism of formation and nature of transient intermediates.<sup>2-5</sup> Recently, evidence for the formation of both enols *Z*-(**1**) and *E*-(**1**) from (**1**) has been provided, the former being shorter lived in aprotic solvents

owing to internal ketonization.<sup>3b</sup> Chemical evidence for the longer lived enol *E*-(**1**) exists from the structure of the product (**2**) obtained by irradiation of (**1**) in the presence of maleic anhydride.<sup>6</sup>

There has been little exploitation of this reaction in synthesis,<sup>6,7</sup> and only symmetrical dienophiles have been used as Diels-Alder traps of the photoenols. The synthesis of lignans of the aryl-naphthalene (*e.g.* helioxanthin<sup>8</sup>), aryl-dihydronaphthalene (*e.g.* collinusin<sup>9</sup>), and aryltetralin (*e.g.* otobain<sup>10</sup>) classes prompted a study with particular regard to stereospecific lactone formation. We have found



that when *o*-methylbenzophenone (**1**) is irradiated (Hanovia 8A-1 high-pressure lamp, Pyrex filter) in benzene under reflux with 4-hydroxybut-2-enoic acid lactone<sup>11</sup> (an unsymmetrical dienophile previously reported as rather unreactive<sup>12</sup>) the only isolated product (50% yield) was an adduct, m.p. 144–145° to which we attribute the structure (**3a**) (i.r. and n.m.r. spectra). Chemical evidence, excluding structure (**3b**), was obtained by treatment of the adduct with hydrochloric acid in methanol to effect rapid and quantitative dehydration to the known arylidihydronaphthalene lactone (**4**).<sup>13</sup> The relative configurations at the three asymmetric carbon atoms were established by reduction of the adduct with lithium aluminium hydride to yield a triol (**5**), m.p. 132–136°, identical with that obtained by similar reduction of the anhydride (**2**) of established stereochemistry.<sup>8</sup> The same carbonyl orientation in adduct formation was observed in irradiation of (**1**) in the presence of the unsymmetrical and *trans*-dienophile, crotonaldehyde, to give as the sole isolated product (27%) the aldehyde (**6**), m.p. 156–157°, which also was readily dehydrated by heating with florisol in benzene, to yield the conjugated aldehyde (**7**), m.p. 115–116°,  $\nu_{\max}$  (KBr) 1656  $\text{cm}^{-1}$ .

With regard to general aspects of lignan synthesis, 1-hydroxyphenyltetralins are readily aromatized to phenylnaphthalenes [*e.g.* (**3a**)  $\rightarrow$  (**4**)  $\rightarrow$  (**8a**)], and lactones of type (**8a**) can be conveniently converted into the isomeric lactones (**8b**) by reduction with lithium aluminium hydride followed by silver carbonate–Celite oxidation.<sup>8</sup>

Satisfactory elemental analyses were obtained for all compounds.

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<sup>1</sup> (a) N. C. Yang and C. Rivas, *J. Amer. Chem. Soc.*, 1961, **83**, 2213; (b) E. F. Zwicker, L. I. Grossweiner, and N. C. Yang, *ibid.*, 1963, **85**, 2671.

<sup>2</sup> J. N. Pitts, jun., H. W. Johnson, jun., and T. Kuwana, *J. Phys. Chem.*, 1962, **66**, 2456.

<sup>3</sup> (a) G. Porter and M. R. Topp, *Proc. Roy. Soc.*, 1970, *A*, **315**, 163; (b) G. Porter and M. F. Tchir, *Chem. Comm.*, 1970, 1372; (c) A. Beckett and G. Porter, *Trans. Faraday Soc.*, 1963, **59**, 2051.

<sup>4</sup> (a) W. A. Henderson, jun. and E. F. Ullman, *J. Amer. Chem. Soc.*, 1965, **87**, 5424; (b) E. F. Ullman and K. R. Huffman, *Tetrahedron Letters*, 1965, 1863; (c) K. R. Huffman, M. Loy, and E. F. Ullman, *J. Amer. Chem. Soc.*, 1965, **87**, 5417; (d) E. F. Ullman, *Accounts Chem. Res.*, 1968, **33**, 3469; (e) K. R. Huffman, M. Loy, W. A. Henderson, jun., and E. F. Ullman, *J. Org. Chem.*, 1968, **33**, 3469; (f) K. R. Huffman, M. Loy, W. A. Henderson, jun., and E. F. Ullman, *Tetrahedron Letters*, 1967, 931.

<sup>5</sup> (a) M. Pfau, N. D. Heindel, and T. F. Lemke, *Compt. rend.*, 1965, **261**, C, 1017; (b) N. D. Heindel, E. W. Sarver, and M. A. Pfau, *Tetrahedron Letters*, 1968, 3579; (c) M. Pfau, E. W. Sarver, and N. D. Heindel, *Compt. rend.*, 1969, **268**, C, 1167; (d) N. D. Heindel, J. Molnar, and M. Pfau, *Chem. Comm.*, 1970, 1373.

<sup>6</sup> F. Nerdel and W. Brodowski, *Chem. Ber.*, 1968, **101**, 1398.

<sup>7</sup> S. M. Mellows and P. G. Sammes, *Chem. Comm.*, 1971, 21.

<sup>8</sup> T. L. Holmes and R. Stevenson, *Tetrahedron Letters*, 1970, 199; *J. Chem. Soc. (C)*, 1971, in the press.

<sup>9</sup> E. Block and R. Stevenson, *Chem. and Ind.*, 1970, 894.

<sup>10</sup> I. Maclean and R. Stevenson, *J. Chem. Soc. (C)*, 1966, 1717.

<sup>11</sup> R. Palm, H. Ohse, and H. Cherdron, *Angew. Chem. Internat. Edn.*, 1966, **5**, 994.

<sup>12</sup> J. J. Leavitt, Ph.D. Dissertation (Harvard University, 1942), quoted by H. L. Holmes, *Org. Reactions*, 1948, **4**, 72.

<sup>13</sup> L. H. Klemm, D. Hsu Lee, K. W. Gopinath, and C. E. Klopfenstein, *J. Org. Chem.*, 1966, **31**, 2376.