Hydrogen Bromide Induced Anthrasteroid Rearrangements

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Summary 17β -Acetoxy-androstane derivatives containing two potential sites of unsaturation in rings A and B undergo rearrangement to anthrasteroids on treatment with acetyl bromide and hydrogen bromide generated in situ from acetvl bromide and propan-2-ol.

WE reported in the preceding communication¹ that treatment of steroids possessing three potential sites of unsaturation in rings A and B with acetyl bromide and HBr results in mixtures of aromatic products, composed predominantly of 4-methyloestra-1,3,5(10)-triene derivatives. The minor products formed in these reactions were anthrasteroids. Thus cholestenone (1a) and testosterone acetate (1b) gave in addition to the above oestratriene derivatives the anthrasteroids (2a) (25%) and (2b)[†] respectively.

Similar treatment of steroids possessing two sites of unsaturation in rings A and B, and a third one in ring D, gave anthrasteroids, lacking functional groups in ring D. Thus heating of either (3a), (4a), or (5a) with acetyl bromide and propan-2-ol in a sealed ampoule, under the conditions described in the preceding communication, gave in each case the same mixture of anthrasteroids in about 40% yield which was separated into (9a), (9b), and (9c). Analogous cholestane derivatives (3b), (4b), and (5b) did not yield aromatic compounds under these conditions, indicating, that the third site of unsaturation in ring D is essential for this aromatization to occur.

When (3a), (4a), or (5a) were submitted to the above reaction conditions for shorter periods, the dibromide (6a) was isolated as the main product in each case. (6a) was readily dehydrobrominated to the monobromide (7a), which is probably an intermediate in the formation of the anthrasteroids (9a), (9b), and (9c).[‡] This transformation involves $[(7a) \rightarrow (9)]$ epimerization at C-14 followed by elimination of the 17β -acetoxy group. The resulting double bond migrates from ring D by a series of HBr additions and eliminations giving the benzenonium ion (8a) having at C-14 the thermodynamically more stable β -configuration. The formation of anthrasteroids (9b) and (9c) may be explained by an additional rearrangement involving either a 1,2-migration of the C-18 methyl group from C-13 to C-14, or two 1,2-migrations (C-12 \rightarrow C-14 and C-8 \rightarrow C-13) to give benzenonium ions (8b) and (8c). Ring B benzenonium ions were postulated previously to be the intermediates in the acid catalysed "anthrasteroid rearrangement" of $\Delta^{5,7,9(11)}$ trienes observed in the cholestane and ergostane series.^{3,4}

The 17β -acetoxy group undergoes elimination followed by HBr addition only when rings c/D are cis fused and the acetoxy group assumes a quasi-axial conformation. Thus



 \dagger The formation of 17-oxo-anthrasteroid from and rost-4-en-3,17-dione by acetyl bromide and α -bromopropionic acid has been reported previously.2

[‡] It is possible that (7) preceeds (6) in these rearrangements.

treatment of 3β , 17β -diacetoxy- 5α , 14β -androstane (10) with acetyl bromide and propan-2-ol gave the 17β -Br analogue (11).§ 3β , 17β -Diacetoxy- 5α , 14α -androstane was recovered unchanged on similar treatment and in the anthrasteroid

(2b) the 17β -acetoxy substituent (quasi-axial conformation) was replaced by a bromine atom.

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§ This bromo-derivative was also formed by HBr addition to the $\Delta^{16,17}$ olefin obtained by the elimination of 17β -acetoxy group from (10).

¹ J. Libman and Y. Mazur, preceding communication.

² J. Libman and Y. Mazur, preceding communication.
² J. Schmitt, J. J. Panouse, M. Pluchet, A. Hallot, P. J. Cornu, and P. Comoy, Compt. rend., 1964, 259, 1652; Bull. Soc. chim. France, 1965, 1934.
³ K. Tsuda and R. Hayatsu, J. Amer. Chem. Soc., 1955, 77, 3089; K. Tsuda, R. Hayatsu, J. A. Steele, D. Tanaka, and E. Mosettig, *ibid.*, 1963, 85, 1126; J. A. Steele, L. A. Cohen, and E. Mosettig, *ibid.*, 1963, 85, 1134; O. Tanaka and E. Mosettig, *ibid.*, 1963, 85, 1131.
⁴ N. L. Wendler, "Rearrangements in Steroids," Vol. 2, Ed. P. DeMayo, J. Wiley, New York, 1964, pp. 1063—1067.