# Crystal and Molecular Structure of $\left[\mathrm{Mn}(\mathrm{CO})_{4}\left(\mathrm{C}_{2} \mathrm{PPh}_{3}\right) \mathrm{Br}\right]$-a Co-ordination Compound of the Unusual Carbonyl-ylide Product, $\mathbf{P h}_{3} \mathbf{P}^{+}-\mathrm{C} \equiv \mathrm{C}$ :- 

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Summary The crystal and molecular structure of the reaction product of $\left(\mathrm{Ph}_{3} \mathrm{P}\right)_{2} \mathrm{C}$ and $\mathrm{Mn}(\mathrm{CO})_{5} \mathrm{Br},\left[\mathrm{Mn}(\mathrm{CO})_{4}{ }^{-}\right.$ $\left.\left(\mathrm{C}_{2} \mathrm{PPh}_{3}\right) \mathrm{Br}\right]$, has been determined and the complex found to contain the unusual phosphonium acetylide, $\mathrm{Ph}_{3}$. $\mathrm{P}^{+}-\mathrm{C} \equiv \mathrm{C}:-$

The reactions of the bis-ylide, hexaphenylcarbodiphosphorane, ${ }^{1,2}$ with several metal carbonyl compounds have recently been reported by Kaska and his co-workers. ${ }^{3}$ One of these reactions is:

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\left(\mathrm{Ph}_{3} \mathrm{P}\right)_{2} \mathrm{C}+\mathrm{Mn}(\mathrm{CO})_{5} \mathrm{Br} \rightarrow \mathrm{Ph}_{3} \mathrm{PO}+\mathrm{Mn}(\mathrm{CO})_{4}\left(\mathrm{C}_{2} \mathrm{PPh}_{3}\right) \mathrm{Br} .
$$

The mode of bonding and the structure of the ligand $\mathrm{C}_{2} \mathrm{PPh}_{3}$ has been the subject of some conjecture. As a first step in establishing the structural chemistry of this new class of compounds, we have determined the crystal and molecular structure of $\mathrm{Mn}(\mathrm{CO})_{4}\left(\mathrm{C}_{2} \mathrm{PPh}_{3}\right) \mathrm{Br}$ by $X$-ray diffraction methods.

Yellow-orange needle crystals of the material darken upon prolonged exposure to air. Several crystals were mounted in thin-walled glass capillaries under an argon atmosphere. Precession photographs displayed the symmetry and systematic absences consistent with space group $P 2_{1} / c$. Accurate cell constants were determined from least-squares refinement of 12 reflections centred on a Picker FACS-1 diffractometer. The lattice parameters are
(Мо- $K_{\alpha}, \lambda=0.71069 \AA$ ) $a=9.564(3), b=12 \cdot 019(3), c=$ $21.237(8) \AA, \quad \beta=107.02(2)^{\circ} ; \quad D_{\mathrm{c}}=1.52, \quad D_{\mathrm{m}}=1.47(2) \mathrm{g}$ $\mathrm{cm}^{-3}, Z=4$.
Intensities were collected by the $\theta-2 \theta$ scan technique out to a Bragg diffraction angle of $45^{\circ}$ with Mo- $K_{\alpha}$ radiation. The structure was solved by Patterson, least-squares, and Fourier techniques. The co-ordination geometry consists of an octahedrally co-ordinated manganese atom as the cis-isomer surrounded by four carbonyl ligands, the bromide, and the ligand $\mathrm{Ph}_{3} \mathrm{PC}_{2}$ (see Figure). The present model for


Figure. $A$ perspective drawing of $\operatorname{Mn}(\mathrm{CO})_{4}\left(\mathrm{C}_{2} \mathrm{PPh}_{3}\right) \mathrm{Br}$.
least-squares refinement constrains the phenyl carbon atoms to be rigid groups with individual atomic isotropic temperature factors. For 1214 reflections with $F^{2}>3 \sigma\left(F^{2}\right)$ the present weighted $R$ factor is $7 \cdot 7 \%$. There is partial disorder (of the order of $20 \%$ ) between the bromide atom and one carbonyl cis to it [atoms $\mathrm{C}(1)$ and $\mathrm{O}(1)$ in the Figure]. The evact characterization of this disorder and final refinement of the structure is in progress. The present model has the partially disordered carbonyl fixed at a position centrosymmetrically related through the manganese atom to the trans carbonyl [atoms $\mathrm{C}(3)$ and $\mathrm{O}(3)$ ]. This disorder does not aftect the primary chemical feature of this structure, which is the geometry of the $\mathrm{Ph}_{3} \mathrm{PC}_{2}$ ligand.
The $\mathrm{Mn}-\mathrm{C}-\mathrm{C}$ angle of the ligand is $176^{\circ}$ and the $\mathrm{C}-\mathrm{C}-\mathrm{P}$ angle is $165^{\circ}$. The $\mathrm{C}-\mathrm{C}$ bond length within the $\mathrm{Ph}_{3} \mathrm{PC}_{2}$ group is $1 \cdot 20 \AA$. This distance shows that the carboncarbon bond is essentially acetylenic in character. The $\mathrm{Mn}-\mathrm{C}(5)$ distance is $1.99 \AA$ and the other metal carbon
distances are in the range $1.80-1.87 \AA$. The remaining bond lengths and angles for the carbonyl and phenylphosphine groups agree with the values expected based on previous structures.

The $\mathrm{Ph}_{3} \mathrm{PC}_{2}$ ligand is best regarded as a simple phosphonium molecule in which one of the organic groups is the acetylide anion. The structural formula should then be written $\mathrm{Ph}_{3} \mathrm{P}^{+}-\mathrm{C} \equiv \mathrm{C}$ :- . Although other resonance forms can be written, this one is dominant and should be of use in explaining and predicting the subsequent chemistry of $\mathrm{Ph}_{3} \mathrm{PC}_{2}$ and similar compounds.

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