

The Configurations of Dimethylmyleran (2,5-Bismethanesulphonyloxyhexane)

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Summary Reaction of the two isomeric forms of dimethylmyleran with sulphide or hydroxide ions gives cyclic products from which the *meso*- and (\pm)-configurations can be deduced.

DIMETHYLMYLERAN (2,5-bismethanesulphonyloxyhexane) (I) was introduced clinically in 1957 for the treatment of chronic granulocytic leukemia.¹ Whereas its synthesis

produces a mixture of *meso*- and (\pm)-forms, only one, as yet unidentified, isomer² has been isolated.³

Fractional recrystallisation[†] from benzene and methanol of the product from mesylation of a mixture of *meso*- and (\pm)-hexane-2,5-diols separated the isomers:[‡] m.p. 104—105° and m.p. 38—39°. Refluxing ethanolic sodium sulphide converted the high-m.p. isomer into *cis*-2,5-dimethyltetrahydrothiophen (II) and the low-m.p. isomer

[†] A similar separation has been achieved by Professor W. C. J. Ross, Chester Beatty Research Institute, London.

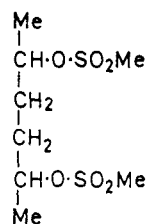
[‡] Satisfactory analytical and spectral data were obtained for all compounds.

into *trans*-2,5-dimethyltetrahydrothiophen (III), isolated as the respective bis-HgCl₂ (m.p. 177—179°; 67%) and mono-HgCl₂ (m.p. 114°; 60%) complexes. Steam distillation of these complexes from 4*N*-HCl liberated the sulphides (b.p. 138—139°)⁴ which, with dilute hydrogen peroxide in acetone gave, respectively, *cis*- and *trans*-2,5-dimethyltetrahydrothiophen *S*-oxide (VI). The 100 MHz n.m.r. spectra (D₂O) of the sulphoxides agreed with the assigned configurations; 2- and 5-H give two superimposed quartets (δ 3.18 p.p.m.) in the *cis*-isomer whereas in the *trans*-isomer they are non-equivalent (δ 3.03 and 3.08 p.p.m.). This sequence indicates that as the displacement reaction proceeds with full inversion of configuration,⁵ the high-m.p. isomer of dimethylmyleran is *meso* and the low-m.p. isomer the (\pm) form.

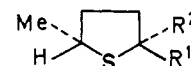
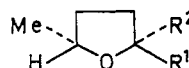
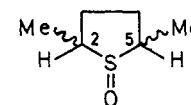
Base hydrolysis (2*N* aqueous NaOH) of the *meso*- and (\pm)-isomers gives, respectively, *cis*- and *trans*-2,5-dimethyltetrahydrofuran (IV) and (V) in quantitative yield, their n.m.r. spectra (CDCl₃) being in agreement with those previously established.⁶ Identical products are obtained by hydrolysis in refluxing 50% aqueous acetone⁷ and in water at 38°.

Each cyclisation produces one isomeric product, determined by g.l.c. (silicone XL-60 on acid-washed DMCS Chromosorb W), showing that the reactions of dimethylmyleran follow an S_N2 mechanism and not S_N1 as has been generally accepted.^{7,8}

Initial anti-fertility examination in male mice of both isomers at equivalent dose levels indicates the *meso*- as



(I)

(II) R¹ = H R² = Me(III) R¹ = Me R² = H(IV) R¹ = H R² = Me(V) R¹ = Me R² = H

(VI)

being more active than the (\pm)-isomer in inhibiting spermatogonial development.⁹

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¹ H. R. Bierman, A. G. Knudson, K. H. Kelly, and G. M. Timmis, *Proc. Amer. Ass. Cancer Res.*, 1957, **2**, 188.

² (a) G. M. Timmis, R. F. Hudson, R. D. Marshall, and H. R. Bierman, *U.S.P.* 3,041,241/1962 (*Chem. Abs.*, 1962, **57**, 13898c) give m.p. 97.5°; (b) R. D. Marshall, Ph.D. Thesis, London, 1955, 51 gives m.p. 97.5°; (c) G. Ferrari and E. Marcon, *Boll. chim. farm.*, 1957, **96**, 429 give m.p. 98—100°.

³ The low-m.p. isomer is mentioned (S. S. Brown, J. L. Everett, and G. M. Timmis, B.P. 861,818/1961 (*Chem. Abs.*, 1961, **55**, 23, 347d) give m.p. 44—45°) though no details are given.

⁴ E. V. Whitehead, R. A. Dean, and F. A. Fidler, *J. Amer. Chem. Soc.*, 1951, **73**, 3632.

⁵ R. M. Dodson and V. C. Nelson, *J. Org. Chem.*, 1968, **33**, 3966.

⁶ D. Gagnaire and P. Monzeglio, *Bull. Soc. chim. France*, 1965, 474.

⁷ G. M. Timmis and R. F. Hudson, *Ann. New York Acad. Sci.*, 1958, **68**, 727.

⁸ Ref. 2b; W. C. J. Ross, "Biological Alkylating Agents", Butterworths, London, 1962, p. 187, G. P. Warwick, *Cancer Res.*, 1963, **23**, 1315.

⁹ H. Jackson, B. W. Fox, and A. W. Craig, *J. Reprod. Fertil.*, 1961, **2**, 447.