¹H Nuclear Magnetic Resonance and Relaxation in [Co en(NH_3)₄]³⁺

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Summary The dynamics of ring inversion for [Co en- $(NH_3)_4$]³⁺ in solution are obscured by a relaxation process which dominates the CH magnetic resonance line width at low temperatures.

¹H NUCLEAR MAGNETIC RESONANCE spectra of cobalt(III) complexes containing ethylenediamine-type chelates usually exhibit broad NH_n and CH_n resonances, thus complicating analysis and extraction of information on the dynamics of processes such as conformational inversion. Some of the possible causes of this have been identified; for example, a line-width contribution from ${}^{3}J$ H–N–C–H coupling was shown by replacing the protons on nitrogen by deuterons,¹ and by proton spin decoupling.² Partially collapsed³ coupling to cobalt (${}^{3}J$ Co–N–C–H), postulated¹ in 1959, was recently confirmed by heteronuclear decoupling.⁴ Similar proton resonance broadening might be expected from ${}^{14}N$ –H and ${}^{14}N$ –C–H couplings. Intermediate rate ring-inversion has also been advanced as a cause of broad CH₂ resonances in ethylenediamine complexes.⁵

Fully protonated [Co en(NH₃)₄]Br₃ in D₂O/DCl showed broad NH₂. NH₃ and CH₂ resonances at room temperature (Figure, A). Irradiation of ¹⁴N (7 w dissipation at transmitter coil) caused a marked sharpening of the two NH3 resonances, a slight sharpening of the NH₂ resonance, but had no effect on the CH₂ resonance (Figure, B). Irradiation of 59Co was effective in sharpening the CH₂ resonances to resolve the "quintet" caused by coupling to the four NH protons, and slightly sharpened the NH₃, NH₂ resonances (Figure, C). In D₂O-NaOD, the CH₂ resonance had a halfheight width about half that in (A) and cobalt decoupling then gave a very sharp (<1.5 Hz) singlet (Figure, D). Given the evidence for non-coplanarity of the CoN₂C₂ ringatoms,⁶ this result shows that there is rapid interconversion between two equivalent puckered-ring conformations, excluding the unlikely possibility of identical chemical shifts of nonequivalent CH₂ protons in a fixed puckered conformation, particularly since a sharp singlet was obtained in a variety of solvents.

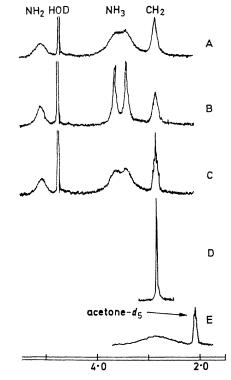


FIGURE. ¹H n.m.r. spectra (100 MHz, chemical shifts in p.p.m. from tetramethylsilane) of $[Co en(NH_3)_4]^{3+}$. [A] Normal spectrum in D₂O/DC1. [B] Spectrum with irradiation at ¹⁴N resonance frequency. [C] Spectrum with irradiation at ⁵⁹Co resonance frequency. [D] Spectrum in D₂O/NaOD with irradiation at ⁵⁹Co resonance frequency. [E] Spectrum in (CD₃)₂CO at -72°.

N-Deuteriated [Co $en(NH_3)_4$](CF₃CO₂)₃ in perdeuterioacetone containing a little D₂O had room-temperature spectra similar to those above for the bromide (Figure, C, D). At -20° the CH₂ resonance was broader than at room temperature and was unaffected by irradiation of ⁵⁹Co, ¹⁴N, or ²D. We conclude that decoupling of ⁵⁹Co by its own spin-lattice relaxation rate had occurred in the more viscous solution. The resonance was almost too broad to detect at -80° and no signals had reappeared by -95° as would be expected if the broadening was caused by a ringinversion rate process. Furthermore, spectra at 60 MHz were consistently broader than those at 100 MHz at the same temperatures in the range -10° to -70° ; the reverse would be expected for chemical shift averaging processes.

That the broadening is a spin relaxation phenomenon is readily corroborated. In the region where molecular rotational correlational times, τ_c , are comparable with the reciprocal Larmor frequency, ω^{-1} , all other things being equal, the 60 MHz spectrum should be broader than the 100 MHz spectrum by a factor of 1.05 for $\tau_{\rm c}=0.4$ ns, 1.18 for $\tau_c = 1.0$ ns, a maximum of 1.32 for $\tau_c = 2.5$ ns, and decreasing again for larger τ_c values, if we take the intramolecular uncorrelated dipolar relaxation⁷ to be dominant. This behaviour was observed for the N-deuteriated complex where the ratio varied between 1.07 and 1.18 for the temperature range -28° to -66° , thus suggesting that $\tau_{\rm c}$ is approximately 0.5 - 1.0 ns, respectively. From the Debye formula, $\tau_{\rm c} = \tau/3 = 4\pi\eta a^3/3kT$, using the measured vis-

cosities of 5 and 15 centipoise at -28° and -66° , respectively, and assuming a rigid complex having a radius of about 4 Å, we obtain $\tau_c = 0.3$ ns and 1.0 ns, respectively.

 $\tau_{\rm c}$ may also be estimated from n.m.r. line widths dominated by quadrupolar relaxation.7 The 59Co spectrum of $[Co en(NH_3)_4]X_3$ in the above solvents had a line width between points of maximum slope of 1 kHz, but we detected no pure quadrupole resonance for the tribromide in the region 2-16 MHz. trans-[Co(en)₂Cl₂]Cl,HCl,4H₂O on the other hand has three ⁵⁹Co quadrupole resonances in this region,⁸ and we found a ⁵⁹Co line width of 10.2 kHz in aqueous acid solution. Assuming that this line width is dominated by quadrupolar relaxation and that $T_1 = T_2$ in the extreme narrowing region, we obtain $\tau_c = 0.02$ ns, for a measured viscosity of 1.14 centipoise.

This behaviour will thus obscure conformational ratebroadening processes, particularly for the low barriers expected to obtain in ethylenediamine-type ligands. We are currently studying the dynamics of conformational processes in complexes chosen to minimise the relaxation effect.

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