

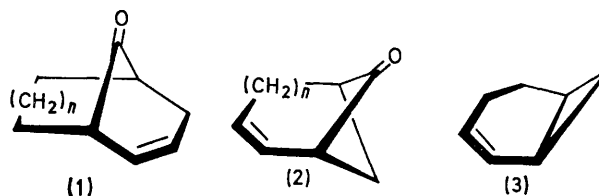
## The Unexpected Photodecarbonylation of Bicyclo[3,2,1]oct-2-en-8-one in Solution

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**Summary** Irradiation of bicyclo[3,2,1]oct-2-en-8-one in hydrocarbon solvents leads to photodecarbonylation as the preferred reaction pathway accompanied to a lesser extent by the previously reported photoisomerization to bicyclo[4,1,1]oct-2-en-8-one.

ERMAN AND KRETSCHMAR reported<sup>1</sup> that the photoisomerization of  $\beta\gamma$ -unsaturated bicyclic ketones of type (1) provided a useful synthetic route to the otherwise not readily accessible cyclobutanones of type (2) ( $n > 1$ ).



We report that in the case of at least one of the representative examples of this study,<sup>†</sup> extensive decarbonylation is

<sup>†</sup> The photochemical behaviour of the other examples is also being investigated.

the preferred reaction pathway rather than photoisomerization

Since bicyclo[4,1,1]oct-2-en-8-one (**2**,  $n = 2$ ) was required as an intermediate in our studies on highly strained molecules its synthesis *via* the reported<sup>1</sup> photoisomerization of bicyclo[3,2,1]oct-2-en-8-one (**1**,  $n = 2$ ) was attempted. Major discrepancies in our quantitative results with those reported prompted us to monitor the progress of the reaction carefully.

A pentane solution<sup>‡</sup> of bicyclo[3,2,1]oct-2-en-8-one (**1**,  $n = 2$ ) containing 4% of its dioxolane-derivative as internal standard was irradiated for 130 min with a 200W GE medium-pressure mercury lamp fitted with a Vycor filter ( $\lambda > 210$  nm). Periodic glpc analysis<sup>§</sup> indicated that in addition to the formation of the desired isomeric ketone a major new component was formed which was eluted at a considerably shorter retention time than the two isomeric ketones. Purification of this product by preparative glpc afforded a volatile liquid fraction which was identified as 2-norcarene (**3**) from its spectral data<sup>¶</sup>. The concentration relationship of the various components as a function of the exposure time is shown in the Figure. After 90 min of irradiation, the isomeric ketone constituted only 10% of the reaction mixture, together with almost 35% of 2-norcarene (*ca* 45% yield).

The optimum exposure time for the maximum yield of the isomeric ketone (*ca* 22%) appears to be only 30–35 min. It is also noteworthy that the concentration of 2-norcarene could be enhanced by irradiation at  $-60^\circ$  but with almost no change in the maximum yield of the desired ketone. Although photodecarbonylation could be suppressed by irradiating through a Pyrex filter at room temperature, the maximum yield of bicyclo[4,1,1]oct-2-en-8-one (**2**,  $n = 2$ ) remained essentially the same.

The mechanism of photodecarbonylation of ketones<sup>3</sup> and aldehydes<sup>4</sup> in solution is a topic of much present interest.

The report by Schaffner *et al*<sup>4</sup> represents an excellent mechanistic study on the solution photodecarbonylation of  $\beta\gamma$ -unsaturated aldehydes, but literature on the detailed mechanistic aspects of the analogous reaction of cyclic and polycyclic  $\beta\gamma$ -unsaturated ketones are almost non-existent<sup>5</sup>. Both the ease of this liquid phase decarbonylation and its selectivity in giving (**3**) in preference to the apparently equally likely isomer, bicyclo[3,2,0]hept-2-ene, are remarkable.

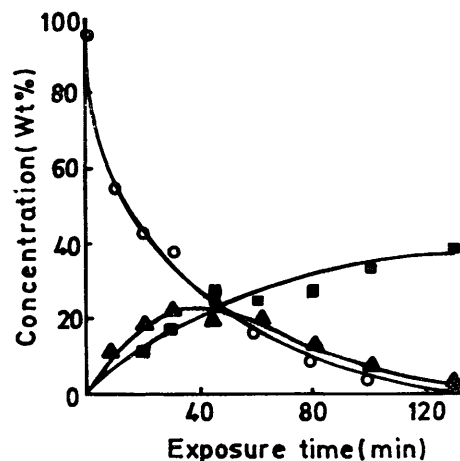


FIGURE Concentration of major components as a function of time of irradiation,  $\circ$ —bicyclo[3,2,1]oct-2-en-8-one (**1**,  $n = 2$ ),  $\blacktriangle$ —bicyclo[4,1,1]oct-2-en-8-one (**2**,  $n = 2$ ),  $\bullet$ —2-norcarene (**3**).

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<sup>‡</sup> While the results were similar when the original solvent, cyclohexane, was used, the separation of the hydrocarbon product is much more easily accomplished in the lower boiling solvent.

<sup>§</sup> A column of 6 ft  $\times$   $\frac{1}{8}$  in 30% Carbowax 20 M on 60/80 Firebrick was used, column temperature was programmed from 40–100°.

<sup>¶</sup> We thank Prof. Dr. Georg Wittig, Institut für Organische Chemie, Universität Heidelberg for sending us the spectral data of 2-norcarene prepared according to an independent procedure.<sup>2</sup>

<sup>1</sup> W. F. Erman and H. C. Kretschmar, *J. Amer. Chem. Soc.*, 1967, **89**, 3842.

<sup>2</sup> G. Wittig and F. Wingler, *Chem. Ber.*, 1964, **97**, 2139.

<sup>3</sup> P. Dowd, A. Gold, and K. Sachdev, *J. Amer. Chem. Soc.*, 1970, **92**, 5724 and 5725, P. Dowd, G. Sen Gupta, and K. Sachdev, *ibid.*, p. 5726; N. J. Turro and D. McDaniel, *ibid.*, p. 5727; P. S. Engel, *ibid.*, p. 6074; W. K. Robbins and R. H. Eastman, *ibid.*, p. 6076 and 6077.

<sup>4</sup> E. Baggiolini, H. P. Hamlow, and K. Schaffner, *J. Amer. Chem. Soc.*, 1970, **92**, 4906.

<sup>5</sup> Preliminary mechanistic details were given in a few reports on the relationship between structural features and photodecarbonylation of unsaturated ketones: D. I. Schuster, F.-T. H. Lee, A. Padwa, and P. G. Gassman, *J. Org. Chem.*, 1965, **30**, 2262; J. E. Starr and R. H. Eastman, *ibid.*, 1966, **31**, 1393 and references therein.