

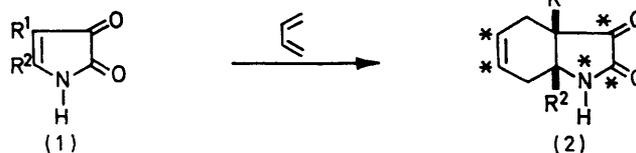
Synthesis and Cycloaddition of 3-Phenyl- Δ^2 -pyrroline-4,5-dione, a New Dienophile

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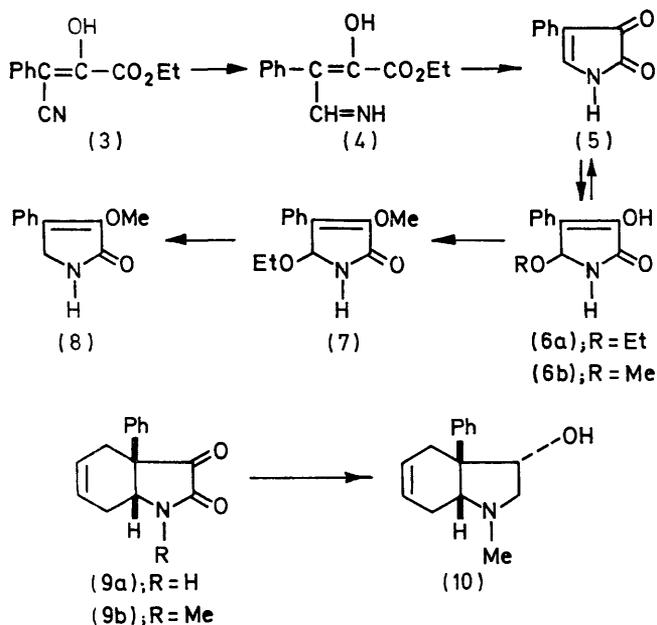
Summary 3-Phenyl- Δ^2 -pyrroline-4,5-dione has been synthesized; it is a useful dienophile and yields an angularly substituted *cis*-fused tetrahydro-indole.

CYCLOADDITIONS of butadienes to pyrroles yielding hydroindoles, although potentially useful reactions, in most cases give disappointing results. We considered that Δ^2 -pyrroline-4,5-diones (1) would be suitable dienophiles for this purpose, though little is known about them.¹ With appropriate substituents R¹ and R², the product (2) is a potential intermediate in the synthesis of various natural alkaloids possessing the hydroindole partial structure, particularly those of Amaryllidaceae (*e.g.* crinine), Aizoaceae (mesembrine), Menispermaceae (hasubanonine), and Erythrina (erythraline) species, since substitution at the positions marked with an asterisk is relatively easy. Here we report the synthesis and cycloaddition of 3-phenyl- Δ^2 -pyrroline-4,5-dione (5).



The pyruvate (3)² was converted (Raney Ni; H₂, 1 mol. equiv.) into the gummy imine (4) which cyclized to the white crystalline ethoxy-derivative (6a)†, m.p. 144–149° (30–40%) when heated in ethanol. (6a) was also obtained from (4) in lower yield by the action of EtO⁻. Spectroscopic evidence (i.r., u.v., and n.m.r.), colouration (green) with ferric chloride, and the following reactions confirmed its structure. Crystallization from methanol converted (6a) into (6b), m.p. 187–194°. Methylation of (6a)

† Satisfactory C, H, and N analyses were obtained for all crystalline compounds.



(MeI, EtONa-EtOH) yielded the methyl ether (7), m.p. 108–110°, which on reduction with LiAlH_4 afforded (8), m.p. 199–201°, (i.r. 1692 cm^{-1}).

Heating of (6a) in Me_2SO caused its conversion into the pyrrolidone (5); the solution turned red (λ_{max} 450 nm; ϵ 560) [cf. original absorptions at 295 (ϵ 14,500) (EtOH) and 330 nm (ϵ 14,500) (EtOH-EtONa)]. The change is reversible and the colour slowly diminished on cooling. Similar, but more rapid reversible changes were observed when (6a) was heated in toluene with acid (*p*- $\text{Me}_6\text{H}_4\cdot\text{SO}_3\text{H}$) or with base (Et_3N).

Thus heating of (6a) with butadiene in Me_2SO (or in HCONMe_2) at 150° furnished the expected oxindole (9a), m.p. 158–159° (60–70%), which was also obtained directly from (4) by similar treatment (15–20%) suggesting that the cyclization (4) \rightarrow (5) occurs thermally. The structure of (9a) was confirmed by its i.r. (1725 and 1770 cm^{-1}) and n.m.r. spectra [δ 7.35 (5H, s, Ar), 6.01 (2H, m, $\text{CH}=\text{CH}$), 4.58 (1H, t, J 4 Hz, $\text{CO}\cdot\text{NH}\cdot\text{CH}\cdot\text{CH}_2$), and 9.49 (1H, br, NH)], and also by its methylation with Me_2SO_4 and KOH to the *N*-methyl derivative (9b) (oil; 100%). Reduction of (9b) with LiAlH_4 quantitatively yielded the indole (10)† (picrate, m.p. 215–216°).

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† The stereochemistry of the hydroxy-group in (10) is discussed in the following communication.

¹ Cf. O. Mumm and G. Munchmeyer, *Ber.*, 1910, **43**, 3345; O. Mumm and H. Hornhardt, *ibid.*, 1937, **70**, 1930; L. Horwitz, *J. Amer. Chem. Soc.*, 1953, **75**, 4060.

² Erlenmeyer, jun., *Annalen*, 1892, **271**, 173.