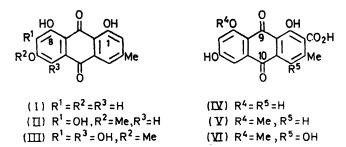
Biosynthesis of Anthraquinone Pigments in Dermocybe

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Summary The pathways of anthraquinone biosynthesis in Dermocybe sanguinea and D. semisanguinea have been investigated; labelled endocrocin is not incorporated into emodin and related anthraquinones.

THE structure of emodin (I) suggests that it may be biosynthesized from endocrocin (IV) by decarboxylation. To test this possibility experimentally, we fed the ammonium



salt of (IV), ¹⁴C-labelled at C-9 and the carboxy-group,¹ to young sporophores of *Dermocybe sanguinea* (Bull. ex Fr.) Wünsche. This mushroom has recently been shown to contain (I), (IV), and several other anthraquinone pigments.² After 3 days, the mushrooms (135 g) were collected and worked-up as described previously, without the use of ion-exchange resin, however. The anthraquinone

carboxylic acids (IV)—(VI) were separated by chromatography on acetylated polyamide and purified by preparative t.l.c. and paper chromatography (Table 1).

TABLE 1

Incorporation of [C-9 and CO₂H-¹⁴C] endocrocin into anthraquinones of D. sanguinea (Amount fed: 16 μ M, 0.32 μ Ci).

Pigment		Amount isolated μΜ μCi ^a		Incorporation (%)
Emodin	(I)	785	0ъ	0
Dermoglaucin	ÌÍ)	92	0	Ó
Dermocybin	(III)	182	0	0
Endocrocin	ĺΙV	57	0.085	26.5 recovered
Dermolutein	(V)	41	0.006	1.9
Dermorubin	(VI)	177	0.03	9.4

^a Determined by combustion analysis. ^b 210 µM sample.

The results indicate that (IV) is the precursor of the anthraquinone carboxylic acids (V) and (VI). It is not incorporated, however, into the neutral pigments (I)—(III). This casts strong doubt on the possible role of (IV) as a precursor of (I) and related pigments in *Dermocybe*.

That (I) is the precursor of (II) and (III) was shown by a separate experiment. $[2,4-{}^{3}\mathrm{H}_{2}]\mathrm{Emodin-6}$ -mono- β -D-glucoside was administered to the caps of young *D. semisan*guinea (Fr.), a species closely related to *D. sanguinea*. The labelled compound was prepared by treating (I) with

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tritiated 0.7N-KOH (24 h; 110°) followed by heating with 0.7N-KOH (3h; 110°)[†] and conversion into the glucoside.³ After 2 days, the mushrooms (40 g) were harvested and worked-up as before (Table 2).

TABLE 2

Incorporation of $[2,4-{}^{3}H_{2}]emodin-6-mono-\beta-D-glucoside$ into anthraquinones of D. semisanguinea (Amount fed: 7.4 µM, 7.85 µCi)

Amount isolated									
Pigmen	it		μΜ	μCia	Incorporation (%)				
(I)			$22 \cdot 5$	1.86	23.7 recovered				
(11)			57	2.06	26.3				
(III)			38	0.06	0.8				
(IV), (V),	(VI)		40	0	0				

^a Determined by combustion analysis,

[‡] Heating of (I) with KOT for 24 h affords $[2,4,5,7-^{3}H_{4}]$ -(I), which 're-exchanges' the 5- and 7-T with H on subsequent treatment with KOH. We have shown that the aromatic protons of (I) exchanged with D (0.7N-KOD; 100°) in the order: 7-H (complete within 15 min), 5-H (ca. 60 min), 2-H (ca. 12 h), 4-H (ca. 24 h) (Diplomarbeit W. Lösel, TH München 1966).

¹ W. Steglich and W. Reininger, Chem. Comm., 1970, 178.
² W. Steglich, W. Lösel, and V. Austel, Chem. Ber., 1969, 102, 4104.
³ L. Hörhammer, L. Farkas, H. Wagner, and E. Müller, Chem. Ber., 1964, 97, 1662.

The results are in accord with the sequence in the Scheme for anthraquinone biosynthesis in Dermocybe. It is likely

Polyacetate
$$\cdots > X <$$
 (I) \rightarrow (II) \rightarrow (III)
(IV) \rightarrow (V) \rightarrow (VI)

that both series have a common precursor X, which may be endocrocin-9-anthrone¹ or even a compound in which the ring carrying the carboxy-group is not yet aromatic.

We thank the Deutsche Forschungsgemeinschaft for financial support and Miss H. Pflaumer for the radioactive analyses. The help of Professor Dr. M. H. Zenk, Bochum, and Dr. H. Guenther, München, is gratefully acknowledged.

(Received, March 31st, 1971; Com. 429.)