New Silicon–Nickel Complexes: Bis(substituted silyl)bipyridylnickel(II)

By Y. KISO, K. TAMAO, and M. KUMADA*

(Department of Synthetic Chemistry, Kyoto University, Kyoto, Japan)

Summary Two bis(substituted silyl)bipyridylnickel(II) complexes have been prepared by the reaction of diethyl-(bipyridyl)nickel(II) with HSiCl₃ or HSiMeCl₂, which react with HCl and DCl to form, respectively, the corresponding hydrosilanes and deuteriosilanes.

ALTHOUGH various types of Group IVB derivatives of the transition elements have been studied,¹ only three compounds containing nickel-silicon bonds have been reported.2,3 During our studies on hydrosilylation of olefins catalysed by nickel complexes,⁴ we have isolated two complexes (Ia) and (Ib) containing nickel-silicon bonds and studied their reactions.

Diethyl(bipyridyl)nickel(II)⁵ was treated with Cl₃SiH in ether at -50 to -40° for 1.5 h under nitrogen. The colour of the reaction mixture changed from deep green to brown with evolution of 1 equiv of hydrogen, the only gaseous product detectable by mass spectrometry. Filtration of the reaction mixture, at $ca - 70^{\circ}$ gave bis(trichlorosilyl)bipyridylnickel(II) (Ia),† an insoluble yellow-brown powder, in 85% yield. G.l.c. analysis of the filtrate showed the formation of $EtSiCl_3$ (64%). Although detection of ethane has so far been unsuccessful, the principal process of the present reaction is stoicheiometric with equation (1).

$$\begin{array}{l} [\mathrm{NiEt_2(bipy)}] + 3\mathrm{HSiX_3} \rightarrow \\ [\mathrm{Ni}(\mathrm{SiX_3})_2(\mathrm{bipy})] + \mathrm{EtSiX_3} + \mathrm{H_2} + \mathrm{C_2H_6} \\ (\mathrm{I}) \ \mathbf{a}, \ \mathrm{X_3} = \mathrm{Cl_3} \\ \mathbf{b}, \ \mathrm{X_3} = \mathrm{MeCl_2} \end{array}$$
(1)

Similarly, but at higher temperature $(-40 \text{ to } -30^\circ)$, (Ib), the methyldichlorosilyl analogue of (Ia), was obtained.

It was of similar appearance and was obtained in 84% yield, along with EtMeSiCl₂ (96%).

Both these new complexes are stable in an inert atmosphere but they fume and decompose spontaneously in air leaving a light green substance, possibly dichloro(bipyridyl)nickel(II). Complex (Ia) is insoluble in most organic solvents and decomposes immediately in protic solvents, while (Ib) is soluble in benzene giving a reddish solution.

Compounds (Ia) and (Ib) reacted with HCl in benzene at room temperature to give trichlorosilane [v(Si-H) 2260 cm^{-1}] and methyldichlorosilane [v(Si-H) 2210 cm⁻¹], respectively, along with dichloro(bipyridyl)nickel(II). With DCl, trichlorodeuteriosilane [v(Si-D) 1645 cm^1]^6 was obtained from (Ia) and methyldichlorodeuteriosilane $[v(Si-D) 1600 \text{ cm}^{-1}]$ from (Ib). These reactions show the

$$\begin{array}{c} X_{3}Si & >C = C \\ \downarrow & \downarrow & \downarrow \\ M^{n} - C - C - H & >c = C \\ \downarrow & M^{n} - H + H - C - C - SiX_{3} \\ \downarrow & \downarrow & \downarrow \\ \end{pmatrix}$$

presence of Ni-Si bonds in (Ia) and (Ib).

The formation of EtSiX₃ in excellent yields in the reaction (1) may be of interest in the light of a suggested mechanism,⁷ equation (2), for transition metal complex catalysed hydrosilvlation of an olefin, which involves a possible intermediate with a silyl and an alkyl group attached to a transition metal atom.

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† The new compounds reported here have satisfactory analytical data.

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