

## A Novel Heterocyclization Method. Synthesis of some Benzimidazoles and Benzoxazoles

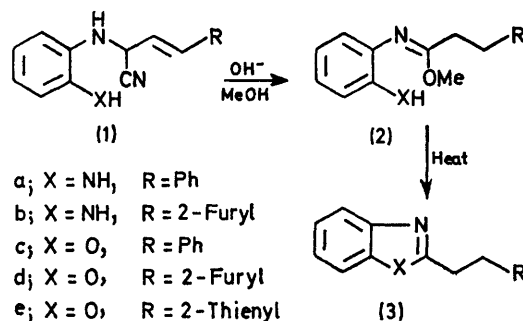
By JASJIT SINGH WALIA,\*PARVEEN S. WALIA, LINDA A. HEINDL, and PHILIP ZBYLOT  
 [Department of Chemistry, Loyola University (New Orleans), New Orleans, Louisiana 70118]

**Summary** Intramolecular cyclization of 1-methoxy-3-substituted-propylidene-*o*-aminoanilines and -*o*-aminophenols provides a convenient new synthesis of some 2-( $\beta$ -substituted-ethyl)-benzimidazoles and -benzoxazoles, respectively.

DISPLACEMENT of the alkoxy-group of imino-esters by nucleophiles is fairly well known.<sup>1</sup> It would be expected that a nucleophile appropriately situated within an imino-ester could, by an addition-elimination mechanism, result in cyclization. We report such a ready heterocyclization for the synthesis of benzimidazoles (**3a** and **b**) and benzoxazoles (**3c-e**).

The *ortho*-substituted anilinonitriles (**1**) were readily prepared by reaction<sup>2</sup> of  $\alpha\beta$ -unsaturated aldehydes with KCN, AcOH and *o*-phenylenediamine or *o*-aminophenol. Treatment of  $\alpha$ -aminonitriles (**1**) with methanolic KOH at room temperature<sup>2</sup> apparently gave good yields of imino-

esters (**2**). Imino-esters (**2a** and **b**), easily identified by a characteristic i.r. absorption band at *ca.* 1645 cm<sup>-1</sup>, were



thermally labile and on heating were converted into benzimidazoles (**3a** and **b**), respectively by loss of MeOH.

However, imino-esters (**2c-e**) appeared to be considerably less stable than (**2a**) or (**2b**), since here the only products isolated were benzoxazoles (**3c-e**). These apparently were the result of cyclization of the intermediate imino-esters (**2c-e**) during work-up. Physical and synthetic data are presented in the Table.

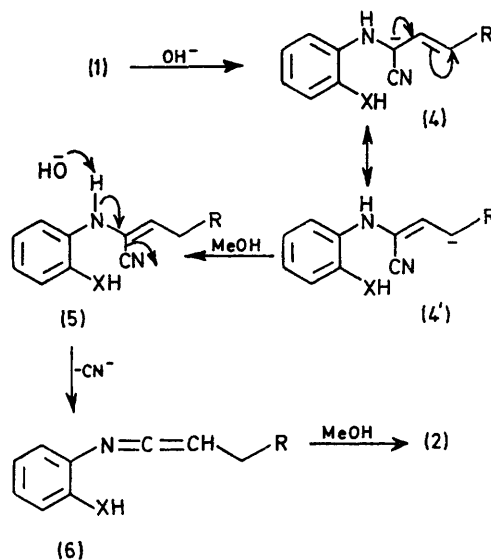
TABLE  
Physical and synthetic data for  $\alpha$ -aminonitriles, benzimidazoles, and benzoxazoles

Compound <sup>a</sup>	$\alpha$ -Aminonitrile ( <b>1</b> )		Benzimidazole or benzoxazole ( <b>3</b> )	
	Yield (%)	M.p. (°C)	Yield (%)	M.p. (°C) or b.p.
<b>a</b>	73	106—108	95	184—185 <sup>b</sup>
<b>b</b>	80	92.5—94	92	167—168
<b>c</b>	97	115—116	62	52—53.5 <sup>c</sup>
<b>d</b>	92	85—87	82	104—109 at 0.5 mmHg
<b>e</b>	75	116—118	85	47—48

<sup>a</sup> The new compounds in the Table gave satisfactory analytical data; i.r., u.v., and n.m.r. data are in agreement with the assigned structures. <sup>b</sup> Lit.,<sup>3</sup> m.p. 189—190°. <sup>c</sup> Lit.,<sup>4</sup> m.p. 54.5°.

The mechanism for the formation of imino-esters (**2**) from  $\alpha$ -anilinnitriles (**1**) is similar to that proposed<sup>2</sup> for the conversion of  $\alpha\beta$ -unsaturated aldehydes into saturated imino-esters via  $\alpha$ -aminonitriles. A possible route is presented in the Scheme.

Since imino-esters (**2c-e**) have not been isolated, it is possible that benzoxazoles (**3c-e**) arise directly from the intramolecular addition of OH to the ketenimine function in intermediate (**6**).



SCHEME

Financial support from Edward G. Schlieder Educational Foundation, Merck, Sharp and Dohme Research Laboratories and Hoffman-LaRoche Inc. is gratefully acknowledged. We thank Dr. Sudhir Bannore and Miss Catherine Voisin for assistance in some experiments. We also thank Mr. Gordon Boudreaux (Southern Regional Research Laboratory) for the n.m.r. spectra.

(Received, November 5th, 1971; Com. 1919.)

<sup>1</sup> R. Bonnett, K. S. Chan, and I. A. D. Gale, *Canad. J. Chem.*, 1964, **42**, 1073; J. Korosi and P. Berencsi, *Chem. Ber.*, 1968, **101**, 1979; H. Paul, G. Hilgetag, and G. Jahnchen, *ibid.*, p. 2033; W. Reid and R. Giesse, *Annalen*, 1968, **713**, 143, 149.

<sup>2</sup> J. S. Walia, P. S. Walia, H. Lader, and L. Heindl, *Chem. Comm.*, 1967, 1290.

<sup>3</sup> B. A. Porai-Koshits and G. M. Kharkharova, *Zhur. obshchei Khim.*, 1955, **25**, 2138 (*Chem. Abs.*, 1956, **50**, 8608h).

<sup>4</sup> J. Haginiwa, *J. Pharm. Soc. Japan*, 1953, **73**, 1312 (*Chem. Abs.*, 1955, **49**, 598h).