Athrotaxin, a C₁₇-Phenolic Constituent from Athrotaxis selaginoides Don.

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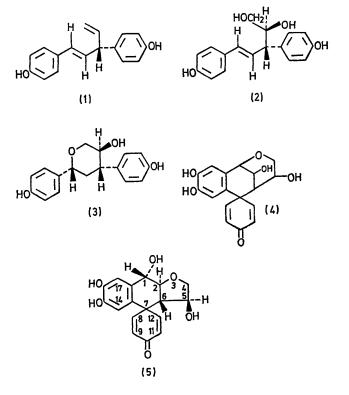
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Summary The heartwood of Athrotaxis selaginoides Don. (Cupressales, Taxodiaceae) contains hinokiresinol (1), agatharesinol (2), and a novel C_{17} -phenol, athrotaxin, the structure, (5), of which has been derived from chemical results, spectroscopic data, and X-ray crystallography.

IN 1960 Erdtman and Vorbrüggen¹ briefly described a phenolic product, $C_{17}H_{18}O_7$, from the heartwood of the Tasmanian conifer *Athrotaxis selaginoides* Don. (Cupressales, Taxodiaceae), obtained in small and varying amounts from different samples, but it was not pure. Re-isolations afforded the pure phenol, $C_{17}H_{16}O_6$, $[\alpha]_D - 225^{\circ}$ (DMF- $H_2O, 1:1; c 0.23$), now called (cf. ref. 2) athrotaxin. From dilute alcohol it forms plates containing 1 mol. equiv. of water, but has no characteristic m.p. It turns brown and sometimes partly melts at ca. 183—185°, then resolidifies and remains solid up to 300°.

Athrotaxin is accompanied by several other C_{17} -phenols including hinokiresinol (1) and agatharesinol (2). The latter has now been obtained in a crystalline state, m.p. 105—109° (cf. ref. 3). T.l.c. analyses of small-scale extracts gave several spots none of which corresponded to sugiresinol (3). This compound, however, was present in large-scale extracts. After boiling a solution of pure agatharesinol in methanol containing a trace of sulphuric acid it gave a strong spot corresponding to sugiresinol. Moreover, agatharesinol and sugiresinol gave identical phenylurethanes (m.p. 179—180°). Thus the sugiresinol from large-scale extracts is probably an artefact.



Athrotaxin gave a dimethyl ether $C_{17}H_{12}O_2(OH)_2(OMe)_2$, m.p. 120–122°, $[\alpha]_{D} - 257^{\circ}$ (MeOH; c 0.12), which gave metahemipinic acid on oxidation with permanganate, and the dimethyl ether diacetate C₁₇H₁₂O₂(OAc)₂(OMe)₂, m.p. 216—217°, $[\alpha]_{D}$ —158° (CHCl₃; c 0.36), on acetylation. Alkaline hydrolysis of the diacetate gave athrotaxin dimethyl ether.

The presence of a dienone group in athrotaxin was clearly demonstrated by its n.m.r. (see Table), u.v. $[\lambda_{max}]$

it was decided to submit the compound to a direct X-ray phase structure determination. The compound crystallized in space group $P2_12_12_1$; Z = 4; a = 15.263, b = 6.960, c =14.529 Å. X-Ray intensity data were collected on a Siemen's automatic diffractometer.

The phase determinations were carried out by application of the Σ_1 , Σ_2 and 'weighed sum' formulae.⁶ An E-map based on the correct set of phases (for reflections with |E|> 1.5) revealed the positions of 15 atoms. The remainder

N.m.r. data ^a for athrotaxin and derivatives			
	Athrotaxin in (CD ₃) ₂ SO	Athrotaxin dimethyl ether in (CD ₃) ₂ CO/(CD ₃) ₂ SO (10:1)	Athrotaxin dimethyl ether diacetate in CDCl ₃
1-H 2-H 6-H 5-H 4-H 12-H 12-H 11-H 14-H 14-H	$\begin{array}{l} 4.64(d); \ J_{1,2} \ 8.5 \\ 3.8b \\ 2.25(q); \ J_{6,2} \ 11; \ J_{6,5} \ 6 \\ 3.8b \\ 3.8b \end{array}$	4·76(d); J _{1,2} 8·5 4·0 ^b 2·25(q); J _{6,2} 12·5; J _{6,5} 8 4·0 ^b 4·0 ^b	$\begin{array}{c} 6\cdot21(\mathrm{d}); \ J_{1,2} \ 8\cdot5 \\ 4\cdot00(\mathrm{q}); \ J_{2,1} \ 8\cdot5; \ J_{2,6} \ 12\cdot5 \\ 2\cdot60(\mathrm{q}); \ J_{6,2} \ 12\cdot5; \ J_{6,5} \ 8 \\ 4\cdot97(\mathrm{s}); \ J_{5,6} \ 8; \ J_{5,4} \ 7 \ \mathrm{and} \ 4 \\ 3\cdot9(\mathrm{m}) \end{array}$
	$6.32(m); ^{b} J_{8,9(12,11)} ca. 10$	6·27 (m) ^b	6·25(m)b
	7.00(bd); $J_{9,8(11,12)}$ ca. 10	6.88(bd); $J_{9,3(11,12)}$ ca. 10	6·80(m)b
	6•36(s); 7•13(s)	6·29(s); 7·22(s)	6·35(s); 6·64(s)

TABLE

^a p.p.m. from Me₄Si, internal standard; J, observed coupling constants, Hz. ^b Unresolved.

(EtOH) 217.5, 241, and 283 nm; log ϵ 4.49, 4.25 and 3.40, respectively] and i.r. ($\nu_{\rm KBF}$ 1659 and 1615 cm⁻¹) spectra.

A difference curve⁴ of athrotaxin dimethyl ether diacetate and its hexahydro-derivative (compound A, see below) showed λ_{max} (EtOH) 245 nm (log $\Delta \epsilon$ 4.20), characteristic of para-blocked cyclohexadienones.

Since athrotaxin occurs together with the C₁₇-phenols hinokiresinol and agatharesinol it was natural to assume a structural relation between these compounds. The dienone athrotaxin could have been formed by intramolecular phenol coupling.

On catalytic hydrogenation (PtO2-HOAc) athrotaxin dimethyl ether diacetate gave a mixture of products from which a compound A (m.p. 179–180°; $[\alpha]_D - 46^\circ$, MeOH; c 0.52) was isolated. Analytical results indicated the composition C₁₇H₁₇O(OH)(OMe)₂(OAc)₂. It contained an oxogroup (i.r.) and one free hydroxy-group (i.r., active H). Contrary to athrotaxin dimethyl ether, compound A on hydrolysis yielded a product which on oxidation with periodate gave formaldehyde (80% yield calc. for one 1,2-glycol group).

Structure (4) has been advanced for athrotaxin (ref. 5, cf. ref. 2). The formation of the HO-CH₂-CH(OAc) group in compound A was assumed to proceed via a hydrogenolysis of the benzyl ether linkage. However, athrotaxin exhibits an n.m.r. signal at δ 4.64 p.p.m. (corresponding signal of its dimethyl ether diacetate δ 6.21 p.p.m.) at low field indicating that it arises from an Ar-CH(OH) group as in structure (5). This structure is compatible with all the observed coupling constants. The formation of the HO-CH2-CH(OAc) group in compound A might involve solvolytic elimination of the C(1)-(OAc) group with formation of a cyclic ether which undergoes hydrolytic cleavage.

Since only small amounts of athrotaxin were available

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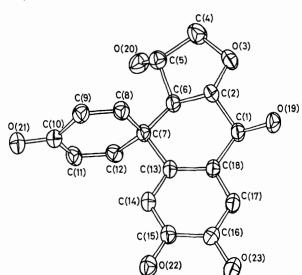


FIGURE. Molecular structure of athrotaxin

of the structure was obtained by a combination of difference syntheses and least-squares refinements. Anisotropic refinement of the non-hydrogen atoms resulted in an *R*-index of 0.07, which was reduced to 0.044 after introduction of the hydrogen atoms into the least-squares analysis with fixed isotropic temperature factors. The molecular structure of athrotaxin [identical with structure (5)] is shown in the Figure. The absolute configuration has not been determined but considering the probable biosynthetic relation between athrotaxin and the compounds (1) and (2) that indicated in formula (5) appears to be the most likely one.

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