

Synthesis of *trans*- and *cis*- β -Lactams Related to Penicillin. Nucleophilic Substitution of a 3-Chloro- β -lactam

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Summary β -Lactams having a common nitrogen atom with valine methyl ester and carrying a benzylthio-substituent at the 4-position, and a phthalimido (**8**), chloro (**9**), azido (**10**), or amino (**11**) substituent at the 3-position of the ring have been synthesized.

PENICILLINS (**1**) and cephalosporins (**4**) have the common structural feature (**5**) and are believed to employ the same mode of action when inhibiting the synthesis of bacterial cell walls.¹ The synthesis of new β -lactams having the partial structure (**5**) is therefore of considerable interest.

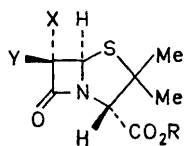
We have recently investigated the preparation of monocyclic β -lactams (**6**) having a common nitrogen atom with α -amino-acid esters and carrying a thio-substituent in the 4-position, and a potential amino-group at the 3-position of the ring. If the side chain, R¹, contains a functional group allowing ring closure on the sulphur, compounds (**6**) may serve as intermediates in a general method for the synthesis of bicyclic β -lactams structurally related to penicillin

and cephalosporin. The present paper deals with the preparation and the stereochemistry of some model compounds in which the amino-acid moiety is valine.

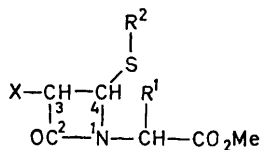
N-Thioformyl-L-valine methyl ester [α]_D²⁰ - 8° (*c* 2; CHCl₃), prepared by thioformylation² of L-valine methyl ester with ethyl thioformate, was treated with sodium hydride in tetrahydrofuran and then alkylated with benzyl bromide to give (**7**) (85%) b.p. 95–110° at 0.02 mmHg, ν_{\max} (CHCl₃) 5.75 (CO₂Me), 6.25 (C=N) μm . Phthaloylglycyl chloride was added during 5 h to a solution of the imidate (**7**) and triethylamine in methylene dichloride. Chromatography of the crude product over silica gel afforded the β -lactam (**8**) (39%), m.p. 88–89°.^{3†} The n.m.r. spectrum indicated the presence of two partially overlapping AB patterns characteristic⁴ of two *trans*- β -lactams: δ (C₆D₆) 5.17 (d, *J* 2.6 Hz) and 5.23 (d, *J* 2.6 Hz) (total 1H); and 5.39 (d, *J* 2.6 Hz) and 5.41 (d, *J* 2.6 Hz) (total 1H); no *cis*-isomer could be isolated. Since only β -lactams of *cis*-stereochemistry can be expected to have

† All the new β -lactams were fully characterized by elemental analysis, and i.r., ¹H n.m.r., and mass spectra.

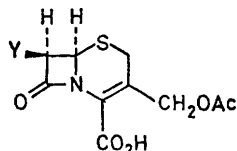
antibacterial activity, another route for their preparation was developed.



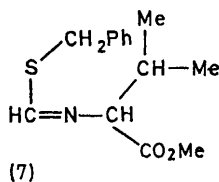
- (1) X=H, Y=acylamino, R=H
 (2) X=Cl, Y=H, R=Me
 (3) X=OSO₂Me, Y=H, R=CH₂Ph



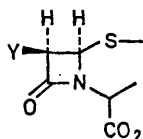
- (6) R¹=side chain
 R²=protecting group
 X=phthalimido, N₃,
 or halogen



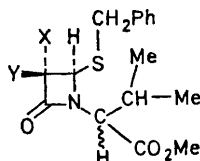
- (4) Y=acylamino



(7)



- (5) Y=acylamino



- (8) X=phthalimido, Y=H
 (9) X=Cl, Y=H
 (10) X=H, Y=N₃
 (11) X=H, Y=NH₂

A solution of (7) and triethylamine in toluene was treated

with chloroacetyl chloride to give, after column chromatography on alumina, the chloro- β -lactam (9) (45%).³ N.m.r. spectroscopy revealed the presence of a mixture of two *trans*-isomers, subsequently separated into its components by column chromatography on Florisil. The n.m.r. spectrum of the first compound to be eluted, m.p. 52–53°, [α]_D²⁵ – 65° (c 1; CHCl₃), showed the characteristic⁴ AB pattern: δ (C₆D₆) 4.38 (d, *J* 1.9 Hz, 1H); 4.55 (d, *J* 1.9 Hz, 1H) as did the second compound (an oil): δ (C₆D₆) 4.33 (d, *J* 1.7 Hz, 1H) and 4.68 (d, *J* 1.7 Hz, 1H). A solution of the lactam (9) (as two *trans*-isomers), and sodium azide in dimethyl sulphoxide was kept in the dark at 85–95° for 24 h to afford after silica gel thick-layer chromatography the *cis*-azido- β -lactam (10) (25%) as a mixture of two diastereoisomers which were separated by chromatography on Florisil. The n.m.r. spectra of these compounds display AB patterns characteristic of *cis*- β -lactams;⁴ the first eluted, m.p. 53–54°: δ (C₆D₆) 3.96 (d, *J* 4.7 Hz, 1H) and 4.81 (d, *J* 4.7 Hz, 1H), and the second, m.p. 76–77°: δ (C₆D₆) 3.98 (d, *J* 5 Hz, 1H) and 4.40 (d, *J* 5 Hz, 1H).

Reduction of (10), m.p. 53–54°, with zinc in 90% acetic acid gave the *cis*-amino- β -lactam (11) (70%), δ (CDCl₃) 4.31 (d, *J* 4.6 Hz, 1H); 5.03 (d, *J* 4.6 Hz, 1H), toluene-*p*-sulphonate, m.p. 137–138°.

The nucleophilic substitution of a chlorine atom by an azido-group in position 3 of a β -lactam ring opens a new route for the preparation of β -lactams bearing sulphur and nitrogen substituents at the same positions as in the natural penicillins and cephalosporins and having the same stereochemistry. It has been reported^{5,6} that attempts to perform similar substitutions in (2)⁵ and in (3)⁶ resulted in the cleavage of the β -lactam ring, while 3-chloro-4,4-diethoxycarbonyl-1-phenyl- β -lactam could not be converted into the respective 3-amino analogue.⁷

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³ For examples of β -lactam synthesis by the acid chloride-triethylamine method see: (a) J. C. Sheehan and E. J. Corey, *Org. Reactions*, 1958, **9**, 388; (b) L. Paul, A. Draeger, and G. Hilgetag, *Chem. Ber.*, 1966, **99**, 1957; (c) A. K. Bose, G. Spiegelman, and M. S. Manhas, *J. Amer. Chem. Soc.*, 1968, **90**, 4506; (d) F. Duran and L. Ghosez, *Tetrahedron Letters*, 1970, 245.

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